

Use of constructed wetlands to prevent overloading of wastewater treatment plants

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Abstract:	<p>The fluctuation in the number of people in tourist areas affects the wastewater quality and quantity. Constructed wetlands (CWs) aim to simulate physical, chemical, and biological processes occurring in natural environments for wastewater treatment and are considered a sustainable system. The current study aimed at evaluating the effectiveness of containerized CWs for supporting the wastewaters treatment plants in periods of overloading. Such approach can be quickly implementable, economic, and the CWs can be fast regenerated in the framework of sustainable good practices. Three CWs were prepared in as many containers layering 10 cm of gravel, 60 cm of sand and 10 cm of gravel, and placing pieces of giant reed rhizomes in the upper layers. The bottom of each CW had a tap, and CWs were irrigated with a real municipal sewage three times a week. Before each new irrigation, the tap was opened, and the effluent collected for determining gross parameters, elemental composition and contaminants of emerging concern (CECs).</p> <p>CWs significantly reduced almost all gross parameters considered and half the CECs, except for a couple of metabolites of corresponding parental compounds. With regards to the potentially toxic elements, all reduced their concentration from the influents to the effluents.</p> <p>The results of this study were promising and highlighted good efficiency of constructed wetlands as pre-treatment of real municipal sewage to reduce the overloading of the wastewater treatment plant.</p>
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1 **Use of constructed wetlands to prevent overloading of wastewater treatment plants**

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18

19 **Abstract**

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39

40 **1. Introduction**

41 Tourist areas are commonly characterized by a considerable increase of people during the holiday
42 season, which leads to an intense seasonal water demand (Salgot and Tapias, 2004). The fluctuation
43 in the number of people affects not only water demand but also wastewater quality and quantity. In
44 fact, in comparison with a typical municipal sewage, the sewages generated in tourist areas are usually
45 more concentrated in terms of typical pollutant parameters, more variable in terms of both flow and
46 contaminant variation, and less disintegrated in terms of particulate matter because of the shorter
47 sewerage system (Odegaard, 1989). Wastewater treatment plants operated in tourist areas often suffer
48 from a series of problems, including shock loads and and flow fluctuations, which may affect effluent
49 quality (Christoulas and Andreadakis, 1989). Therefore, a more reliable and robust wastewater
50 treatment train is needed in tourist areas for coping with the organic load variations typical of these
51 areas.

52 Constructed wetlands (CWs) aim to simulate physical, chemical and biological processes
53 occurring in natural environments for wastewater treating (Bakhshoodeh et al., 2017), and, due to
54 their significant reduction in energy demand, economic costs, less operational and maintenance
55 requirements and environmental pollution, they can be considered a sustainable system capable to
56 enhance the treatment of municipal sewage in tourist areas (Nuamah et al., 2020). With respect to
57 conventional treatment systems, CWs require renewable sources, emit very low amount of
58 greenhouse gas, do not produce by-products, and have small problems of odours and insects
59 (Stefanakis et al., 2014).

60 The physical and chemical processes that occur in CWs are adsorption, oxidation, chemical
61 degradation, photodegradation, volatilization and hydrolysis of contaminants, while biodegradation
62 and plant absorption are the main biological processes (Hijosa-Valsero et al., 2016). These processes
63 can occur simultaneously and lead to several advantages, such as plant nutrients cycling, adsorption
64 of potentially toxic elements (PTE) and organic pollutants.

65 The components of CWs that contribute to the partial or total removal of pollutants are the
66 substrate, the plant, the quality of the influent and the microbial population associated to the CWs
67 (Kosolapov et al., 2004). The substrate plays the most important role in removing contaminants
68 through filtration and adsorption of contaminants, electron donor process, support for
69 microorganisms and wetland plants. Gravel is the most common substrate used in CWs (Li et al.,
70 2014) because provides sufficient hydraulic conductivity, supports the vegetation, retains suspended
71 solids and affects the percolation and infiltration of contaminants (Nuamah et al., 2020; Vymazal et
72 al., 2021) and is largely used also in combination with sand for their affordability and large
73 availability. In addition, sand provides an ideal texture for the establishment of plants, even if it
74 requires frequent addition of wastewater because dries out quickly.

75 Plants play an important role both alive and dead; when alive they deflect flows and provide
76 attachment sites for microorganisms, while when dead they release organic carbon usable by
77 microorganisms for their metabolism (Davis, 1995). In this regard, giant reed (*Arundo donax*) is a
78 perennial herbaceous plant widely spread all over the world that requires high amount of water, and
79 is highly tolerant to salinity, with a fast growth rate and a large amount of biomass produced that
80 make it suitable for use in CWs (Allinson, 2017). Giant reed is also capable to absorb PTE and to
81 improve saline soils (Herrera-Alamillo and Robert, 2012; Quinn et al., 2015) and, due to its great
82 biomass production, can be suitable for the anaerobic digestion for production of energy and digestate
83 (Shilpi et al., 2019; Zhang et al., 2021).

84 The microbial diversity depends on the configuration and design of the CWs (Rajan et al., 2019),
85 and is a function of the concentration of different forms of nitrogen and carbonaceous materials,
86 amount of plant litter, amount of oxygen and available nutrients, and the type of plant used (Liang
87 et al., 2003; Ligi et al., 2014).

88 The level of contamination of influents affects the efficiency of CWs in the removal of
89 contaminants influencing the composition of the microbial community. In fact, several studies have
90 reported the presence of certain microorganisms in presence of specific compounds (Guo et al.,

91 2020; Li et al., 2020; Liu et al., 2020). In addition, CWs can also allow a partial removal of
92 compounds of emerging concern (CECs), such as pharmaceuticals and personal care products,
93 among others. Indeed, this is a side benefit that can be also obtained in the soil due to some
94 components present in it having catalytic behaviour (Pizzigallo et al., 1998; Herrera et al., 2001).
95 CECs, in fact, are poorly removed by the activated sludge process in conventional wastewater
96 treatment plants (WWTPs) and the long-term exposure to them could cause reproductive and
97 hormonal disorders as potential health problems and the increase of bacterial antibiotic resistance
98 (Gonsioroski et al., 2020). Therefore, CECs removal usually requires an additional process (tertiary
99 treatment) after the biological step that can be effectively carried out employing one of the advanced
100 oxidation processes (Murgolo et al., 2021; Maniakova et al., 2020).

101 The current study aims at evaluating the effectiveness of containerized CWs for supporting the
102 wastewaters treatment plants in periods of overloading. Such approach can be quickly implementable,
103 economic, and the CWs can be fast regenerated in the framework of sustainable good practices.

104

105 **2. Materials and Methods**

106 *2.1. Experimental site and constructed wetlands*

107 The study was carried out with three verticals containerized CWs that were built at the IRSA-
108 CNR located in Bari, Southern Italy (5-meter a.s.l., 41°07'31"N, 16°52'00"E). The CWs have been
109 set up according to Hoffmann et al. (2011). In details, from the bottom to the top of each CW, there
110 were 10 cm of gravel, 60 cm of sand and 10 cm of gravel, and the volume of each CW was 0.8-1 m³.
111 The bottom of each CW had a tap and pieces of giant reed rhizomes have been placed in the upper
112 layer. Each CW was irrigated with a real municipal sewage, collected from a local municipal
113 wastewater treatment plant in Bari (a city of Southern Italy), three times a week keeping the tap closed
114 until a 15-20 cm layer of wastewater was obtained on the surface (about 400-500 L for each irrigation
115 per CW). Before each new irrigation, the tap was opened, and the effluent collected for further
116 analyses.

117 According to the Köppen–Geiger climate classification (Kottek et al., 2006), the location is
118 characterized by a warm temperate Mediterranean climate with dry and hot summers. With reference
119 to time series 1921–2019, annual average rainfall is approximately 543 mm, mainly distributed
120 between September and March. July is the driest month with 16.9 mm of rain, while November is the
121 month with the greatest rainfall, having an average of 69.8 mm. With reference to time series 1926–
122 2019, the annual average temperature was 17.0 °C, with average maximum temperature of 25.4 °C
123 in July and average minimum temperature of 9.4 °C in January.

124

125 *2.2. Gross parameters*

126 Composite samples of influent and the effluent of the biological treatment of wastewater were
127 collected three times a week and subjected to gross parameter analysis: electrical conductivity (EC),
128 chemical oxygen demand (COD), soluble COD (sCOD), total suspended solids (TSS), volatile
129 suspended solids (VSS), total nitrogen (TN), ammonia (NH₄⁺), total phosphorus (TP), phosphates,
130 chlorides, and sulphates. COD and TN were measured using DR Lange test kits. sCOD concentration
131 was determined after sample filtration through a 0.45 µm nylon filter. TSS and VSS were analysed
132 according to the procedure proposed by APHA (2005). For NH₄⁺ determination, an online
133 water/liquid L800 ammonia analyser was utilized (Hemera Analyzers, Meylan, France), while anion
134 species such as PO₄³⁻, Cl⁻ and SO₄²⁻ were measured by ion chromatography (Thermo Scientific
135 Dionex Aquion; column IonPac™ AS23; carbonate/bicarbonate as eluent).

136

137 *2.3. Elemental analysis*

138 For the determination of the total content of PTE, plant micronutrients and the exchangeable bases,
139 3 mL of HNO₃ were added to 5 mL of influents or effluents and microwave digested (Multiwave
140 Perkin Elmer 3000). Finally, all samples have been filtered through Whatman 42-filter paper and
141 analysed by using an inductively coupled plasma optical emission spectrometer (ICAP 6300 Thermo
142 Electron).

143

144 2.4. *Contaminants of emerging concern (CECs)*

145 The occurrence of CECs in both influent and effluent wastewater was performed according to an
146 experimental procedure described elsewhere (Montagna et al., 2020). Briefly, influent and effluent
147 samples were diluted 1:4 and 1:2, respectively. They were filtered (0.2 μm nylon filter) and aliquots
148 of 2 mL were analyzed by on-line solid phase extraction (SPE) by a Hypersil GOLDaQ column (20
149 x 2.1 mm, 5 μm) that was used operating at a flow rate of 0.250 mL min⁻¹ of 100 % water. The
150 analytes retained on the SPE column were then separated and detected by ultra-high pressure liquid
151 chromatograph (Ultimate 3000 System, Thermo Fisher Scientific) interfaced with a high-resolution
152 tandem mass spectrometer (TripleTOF® 5600+ System, AB Sciex) by means of a duo-spray ion
153 source operated in electrospray (ESI) positive mode was employed. All MS analyses were acquired
154 by an acquisition method based on double experiments TOFMS/IDA (Information dependent
155 Acquisition) in the mass scan range 50-1000 Da. In order to make the analytical method more robust,
156 each analysed sample was added with the internal standard carbamazepine D10, prior to the analytical
157 determination. In this way, it was possible to take into consideration the relative reduction of the
158 signal intensity due to the matrix effect of each sample.

159 All the raw data obtained by the high-resolution mass spectrometry analysis were processed using
160 a database of emerging pollutants from the NORMAN network and Sciex OS software (AB Sciex).
161 Briefly, the data processing was performed as follows: each sample was screened, principally on the
162 basis of the mass exact and isotopic pattern, using CECs list of the SusDat database (20 suspect lists
163 comprising ~58000 compounds) compiled by the Norman network (Alygizakis et al., 2019). A
164 confident identification of detected compounds in investigated samples was performed by comparing
165 retention time and MS/MS fragmentation pattern of the suspect identified compounds with those of
166 relative standard compounds, where available, analyzed employing the same analytical method. The
167 list of identified compounds was then re-processed in Sciex OS software for confirmation of detected
168 CECs. The screening allowed the confident detection of 20 compounds listed in Table 1.

169

170 2.5 *Statistical analysis*

171 Experimental data were tested against the normal distribution of variables (Shapiro—Wilk test)
172 and the homogeneity of variance (Bartlett test) using R studio software. The variables normally
173 distributed with homogeneity of variances verified were subjected to an ANOVA and HSD test. Data
174 not normally distributed were subjected to the Levene test and a nonparametric ANOVA analysis
175 (Kruskal-Wallis test) and Dunn test.

176

177 **3. Results and discussion**

178 **3.1. Gross parameters**

179 The main gross parameters of influents and effluents of the three vertical CWs are reported in Fig.
180 1. Inspecting Fig. 1A, it is possible to observe that a significant increase of EC was obtained after the
181 percolation of wastewater through the CWs. This should be ascribed to the evapotranspiration process
182 that increased the solute concentration in the effluents. This was also confirmed by chloride (Fig. 1B)
183 which shows a similar trend, also because it is a soluble anion, a plant micronutrient and apparently
184 the excess was not uptaken by *Arundo donax*.

185 The COD content was significantly and drastically reduced by CWs thanks to the action of sand
186 used as adsorbent material. In fact, removal efficiencies in the range of 78.7%-85.7 % were observed
187 with residual effluent concentrations always lower than 200 mg L⁻¹ and sometimes even below 100
188 mg L⁻¹. Similar results have been obtained by Healy et al. (2007) who adopted a stratified sand filter
189 for removal of COD from wastewaters. Other recent studies on CWs have shown a significant
190 reduction in COD, even using different absorbent materials and different technologies (Tuttolomondo
191 et al., 2020; Ruan et al., 2021). The soluble COD shows a trend similar to the COD but less
192 pronounced.

193 TSS and VSS have practically zeroed out after crossing the wetlands due to the substrate's
194 filtration and sedimentation (Batoool and Saleh, 2020) with removal efficiencies comprise in the range

195 89-94.9% and 87.7-93.3%, respectively. Regarding the role of the plant, Toscano et al. (2015) found
196 that *Arundo donax* showed the highest removal rate of TSS with respect to *Vetiveria*, *Miscanthus* and
197 *Phragmites*.

198 As far as nitrogen is concerned, a significant removal of influent TN and ammonia content was
199 also recorded. In fact, looking at Fig. 1B, it is possible to observe that TN and ammonia removal
200 efficiencies from 86.1 to 93.2% and from 77.4 to 98.1%, respectively, were recorded. Considering
201 the absence of an external aeration system, the atmospheric oxygen diffusion through the upper layer
202 into the bed is normally low and therefore nitrification process will not occur. Hence, ammonia
203 removed, included ammonia from ammonification process of organic nitrogen, should be ascribed
204 mainly to the giant reed uptake and, secondarily, to ion exchange with cations in substrates (Ji et al.,
205 2022; Józwiakowski et al., 2017).

206 Finally, phosphorous removal was almost completed with removal efficiencies higher than 95%
207 and residual effluent concentration practically nil (Fig. 1B). A part of P was subjected to biotic uptake
208 by plants and microorganisms, and another one was possibly retained by the substrate through abiotic
209 processes such as co-precipitation, sedimentation, adsorption, and burial (Reddy et al., 1999). The
210 abiotic P removal is mainly governed by the nature and the properties of substrates, by their content
211 of Ca, Fe, and Al (Vohla et al., 2011; Park et al., 2017). Wang et al. (2008) tested the potential of
212 seven plants in removal COD, total nitrogen and total phosphorus from sewage and found that giant
213 reed was the most efficient.

214

215 **3.2. Potentially toxic elements, plant micronutrients and exchangeable bases**

216 This study highlighted different results related to the destiny of PTE, plants micronutrients and
217 exchangeable bases in effluents (Figs. 2-3). Generally speaking, all PTE considered showed a large
218 variability of concentration in the influents, while their content in the effluents was lower and less
219 variable (Fig. 2). In particular, Cu and Zn content was reduced in effluents reaching almost zero (Fig.
220 2A), according to various Authors that considered *Arundo donax* promising for the removal of

221 different metals from the environment (Miao et al., 2012; Delplace et al., 2020; Danelli et al., 2021).
222 Birch et al. (2004) observed a significant reduction of Cr, Cu, Pb and Zn, and a variable removal of
223 Ni from stormwater by CWs in accord with our study in which the removal of Ni did not present any
224 trend (Fig. 2B). Another study supported the reduction of Cu, Pb and Zn content thanks to the CW
225 (Walker and Hurl, 2002), and the mechanism proposed by Authors was the sedimentation. Kadlec et
226 al. (2000) reported different processes involved in removal of PTE in CWs depending on the type of
227 plants adopted, the substrate used and the type of water to restore. According to Ji et al. (2022), the
228 substrate plays the most important role in the removal of PTE and the greater the contact surface, the
229 greater the absorption (Yadav et al., 2010).

230 Apparently, B passed through the CWs without any interaction with the substrate (Fig. 3A), while
231 Türker et al. (2014) found the removal of B from wastewater by adsorption onto the substrate,
232 especially in the presence of organic matter. Therefore, the initial absence of organic matter in the
233 inorganic substrate could explain the present result. The behaviour of Mn was different because it
234 significantly increased in effluents with respect to influents (Fig. 3A), and a similar result has been
235 found by Birch et al. (2004). These Authors hypothesized a release of Mn from the coarse-grained
236 Mn-oxides coated grains during the periods of addition of the influent and more abundant flow. In
237 addition, Mays and Edwards (2001) have shown a poor potential of plants in the removal of Fe and
238 Mn because found in plants only 1% and 2% of the annual Fe and Mn loading in the wetlands. In
239 fact, those elements are micronutrients, required in concentration of mg kg^{-1} in the plant tissues.

240 Figure 3B shows the content of exchangeable bases: Mg and Na content increased significantly in the
241 effluents possibly because the plant uptake was lower than the sum of input and enrichment deriving
242 from the cation exchange between the bases from the CW substrate and the PTE in wastewaters, as
243 reported by Yu et al. (2021). In contrast, K and Ca concentration did not change significantly between
244 influents and effluents due to their higher plant uptake with respect to Mg and Na. Anyway, the
245 exchangeable bases are not of concern.

246

247 3.3. *CECs removal*

248 The screening for CECs in the influent wastewater revealed the presence of 20 main compounds
249 that were detected in all samples, being both influent and effluent. They include pharmaceuticals,
250 cosmetics and personal care products, metabolites of both pharmaceuticals and drugs of abuse, food
251 additives and chemicals employed in various preparations. In Fig. 4 are depicted the box and whiskers
252 plots of abundance of detected CECs in influent and effluents during the whole experimental
253 campaign. Half the CECs showed a statistically significant reduction of abundance when flowing
254 through the CWs.

255 The processes responsible for reducing the presence of organic pollutants in effluents are
256 photodegradation, volatilization, adsorption/sorption, plant uptake and accumulation, and
257 biodegradation (aerobic and anaerobic) (Zhang et al. 2014; Gorito et al. 2017), even if the
258 volatilization and the photodegradation can be insignificant in the case of pollutants with relatively
259 greater molecular weight and in cloudy wastewater, respectively (Wu et al., 2020). A crucial role in
260 the removal of organic pollutants in vertical CWs is generally played by aerobic biodegradation (Ilyas
261 and van Hullebusch, 2020) and plants influence the potential of CWs by creating suitable conditions
262 for the pollutant's removal processes (Ji et al., 2022). Processes of adsorption and desorption of
263 organic pollutants can also occur in the surface of substrates (Karimi et al., 2019).

264 Two CECs, namely Atenolol acid (Fig. 4A) and Deet-carboxylic acid (Fig. 4B), showed a
265 significantly and numerically increase of abundance in the effluents, respectively. This is consistent
266 with the concomitant statistically significant reduction of Atenolol and Deet that are the respective
267 parent compounds of the aforementioned CECs. This finding supports the evidence that
268 biodegradation processes are occurring in the CWs resulting in the transformation of some CECs. In
269 any case, such biodegradation processes are not expecting to allow the complete mineralization of
270 CECs due to the residence time in the CWs not being sufficiently long. The CECs biodegradation is
271 also evident on some CECs metabolite, namely Benzoylcegonine that is a metabolite of cocaine (Fig.
272 4A), resulting in its numerically removal.

273 Overall, the CWs showed a reduction of detected CECs between 10 and 90 %, being the lowest
274 removal percentage (10 %) for the 1,5-Dihydro-3-ethyl-4-methyl-2H-pyrrol-2-one and the highest
275 removals for 2-(phenylmethylidene)octanal (83 %), Equol (89 %) and Deet (87 %) (Fig. 5). The two
276 negative removal values are for the transformation of Deet and Atenolol (Fig. 5) into Deet-carboxylic
277 acid and Atenolol acid, respectively, as reported before. It is worth noting that abundance of
278 transformation products could not be directly compared to the respective parent compounds due to
279 the different response in the mass spectrometer.

280

281 **Conclusions**

282 Containerized constructed wetlands can be useful as pre-treatment in case of overloading of the
283 wastewater treatment plants. In fact, CWs significantly reduce the main gross parameters and PTE
284 content of wastewater to be sent to the main wastewater treatment facility. In addition, the
285 evapotranspiration of the CWs reduces slightly the volume of wastewater. With regards to the CECs,
286 CWs contribute to lowering some organic contaminants significantly. The number of CWs can be
287 adjusted to seasonal needs, they do not release anything into the environment, and can be set aside
288 during periods of low load and regenerated for subsequent high load seasons.

289 Further studies are required to test their efficiency in longer periods of time, and to improve their
290 features for obtaining safe effluents for irrigation.

291

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474

475 **Figure captions**

476

477 **Figure 1.** Gross parameters values in influents and effluents during the whole trial. Error bars
478 correspond to the standard deviations and columns with the same letter are not statistically
479 different.

480 **Figure 2.** Concentrations of potentially toxic elements in influents and effluents during the whole
481 trial. Error bars correspond to the standard deviations and columns with the same letter are not
482 statistically different.

483 **Figure 3.** Concentrations of micronutrients (A) and exchangeable bases (B) in influents and effluents
484 during the whole trial. Error bars correspond to the standard deviations and columns with the same
485 letter are not statistically different.

486 **Figure 4.** Box and whiskers plots of abundance of detected CECs in influents and effluents during
487 the whole trial. Error bars correspond to the standard deviations and columns with the same letter
488 are not statistically different.

489 **Figure 5.** Average percent removal of detected CECs as a result of percolation through the CWs.

490

Highlights

- Containerized constructed wetlands were used to pre-treat real municipal sewage
- Almost all gross parameters significantly reduced in effluents
- Almost all potentially toxic elements significantly reduced in effluents
- Ten contaminants of emerging concern out of 20 significantly reduced in effluents
- Two metabolites increased their concentration in effluents



UNIVERSITÀ
DEGLI STUDI DI BARI
ALDO MORO

Dipartimento di Scienze del Suolo,
della Pianta e degli Alimenti -
Di.S.S.P.A.

Bari, 5th of August, 2022

Dear Prof.

Yeomin Yoon, PhD

Editor

Chemosphere

Dear Editor,

On behalf of all Authors, I like to submit the manuscript entitled “Use of constructed wetlands to prevent overloading of wastewater treatment plants” by Claudio Coccozza, Claudio Di Iaconi, Sapia Murgolo, Andreina Traversa, Francesco De Mastro, Marco De Sanctis, Valerio Guido Altieri, Claudio Cacace, Gennaro Brunetti, Giuseppe Mascolo.

All the authors have read and approved the paper and it has not been published previously nor is it being considered by any other peer-reviewed journal.

I look forward to hearing from you about this matter at your nearest convenience.

With my best regards

Sincerely yours

Andreina Traversa, PhD

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Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

Figure 1

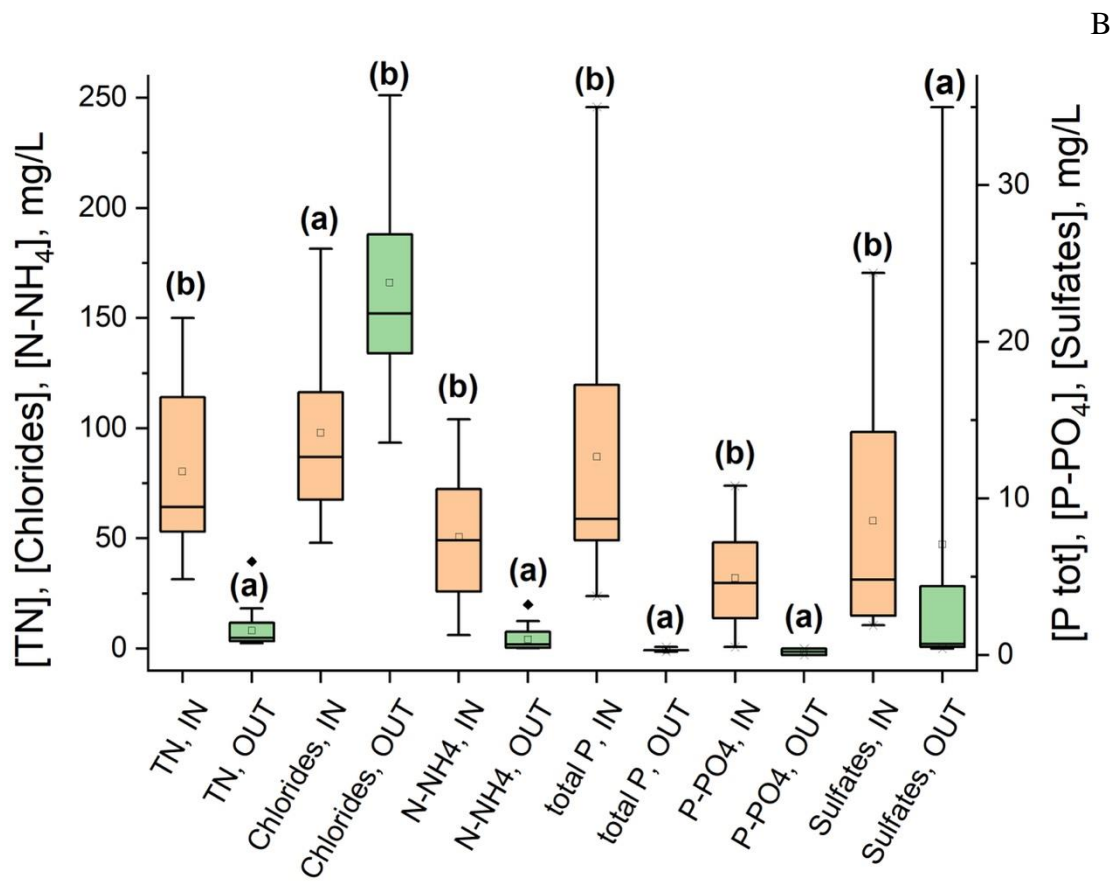
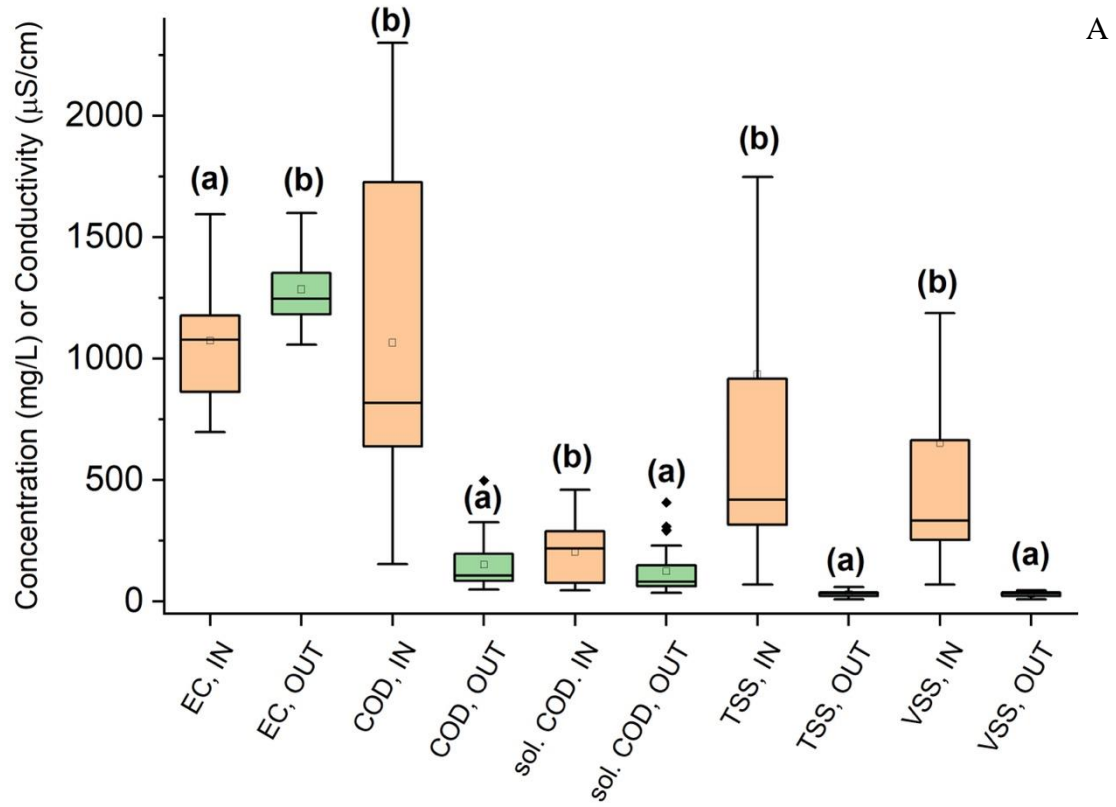


Figure 1

Figure 2

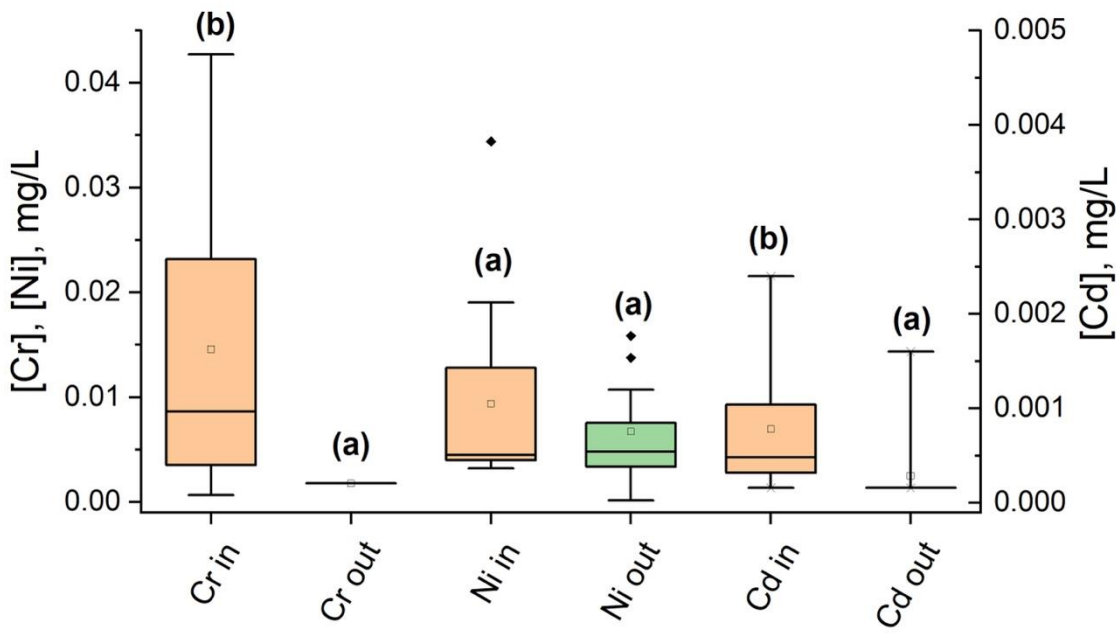
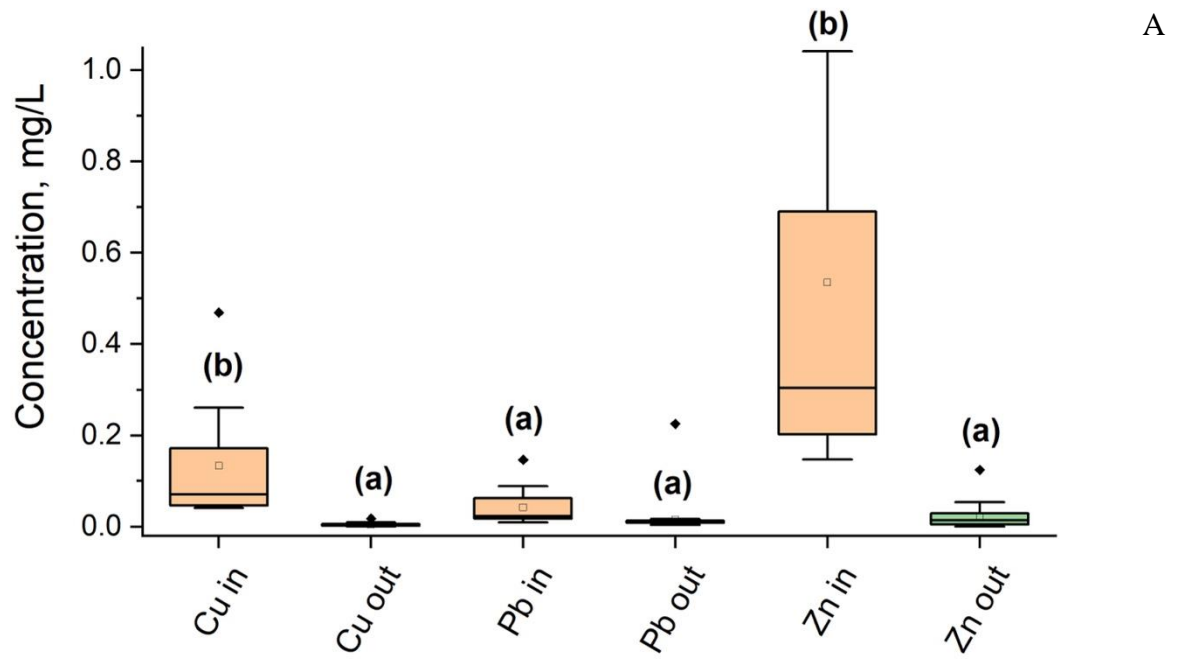


Figure 2

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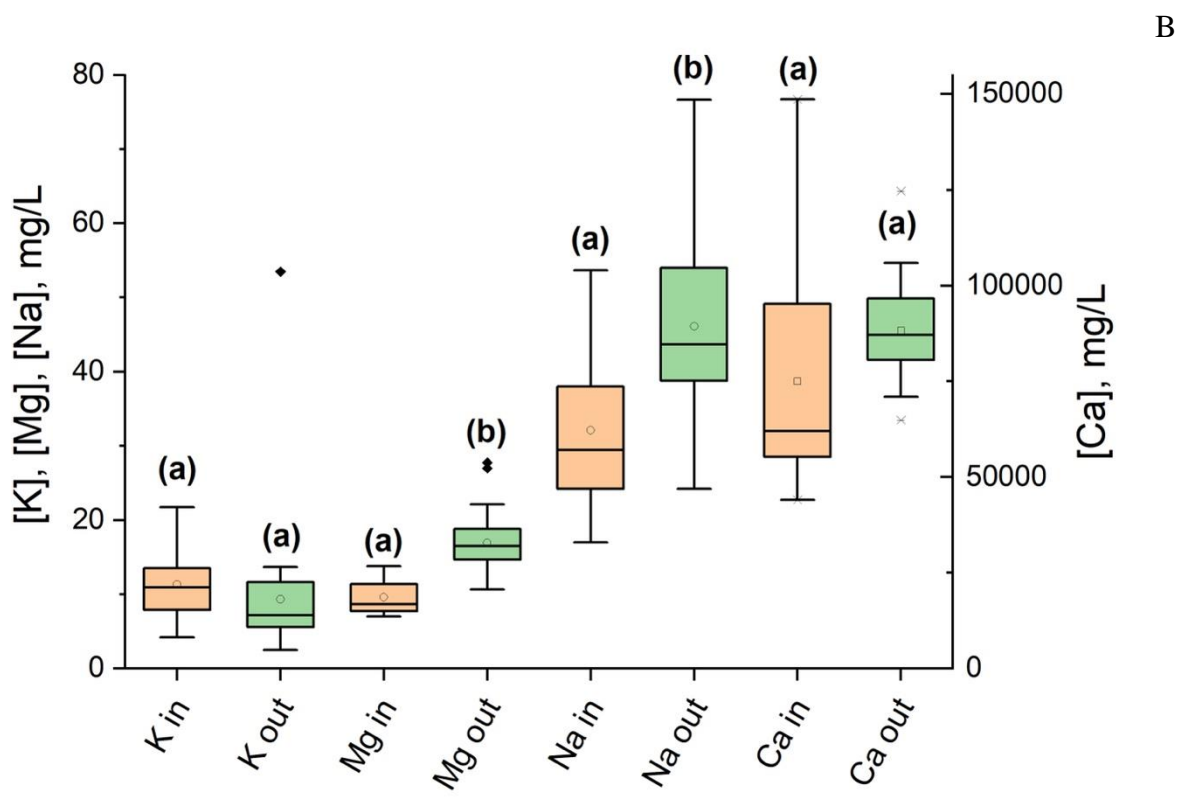
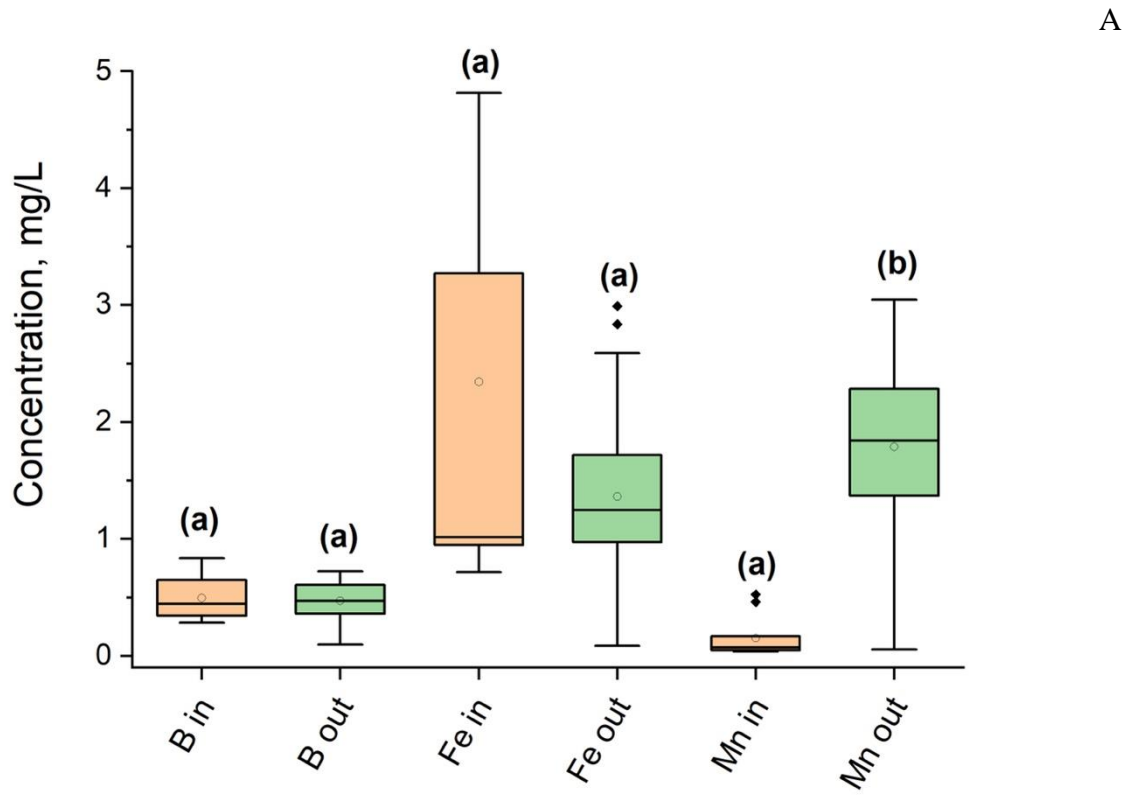


Figure 3

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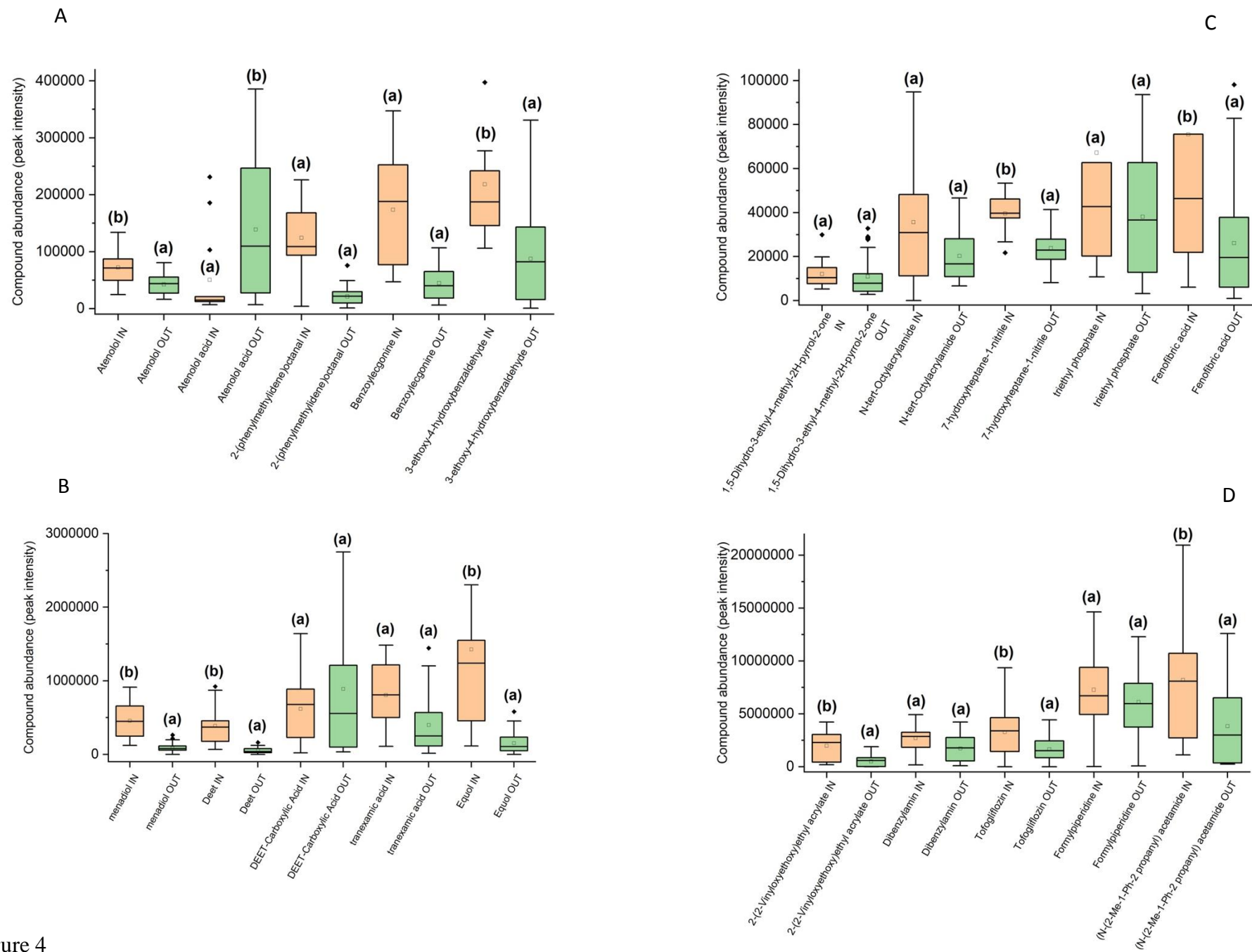


Figure 4

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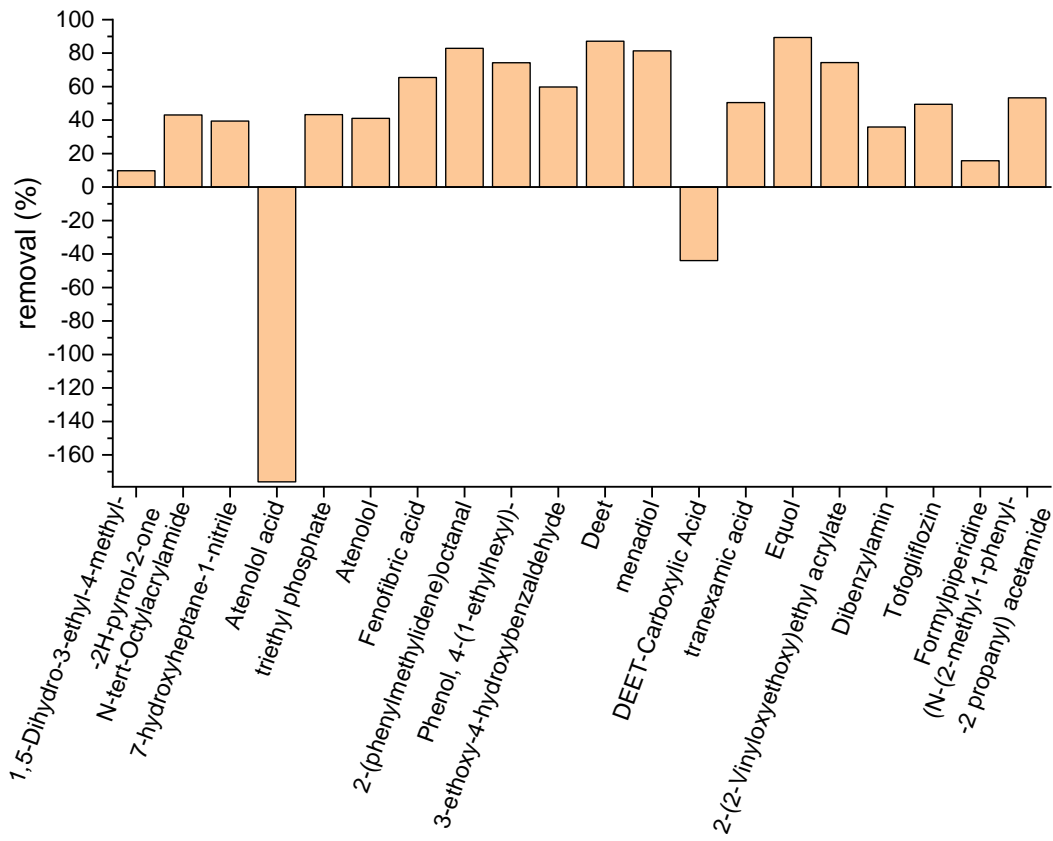
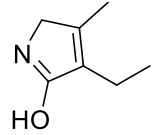
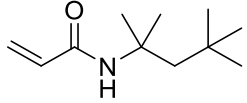
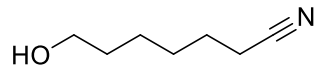
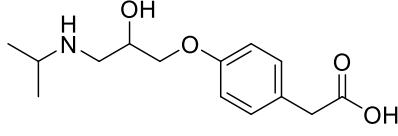
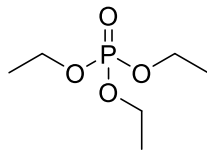
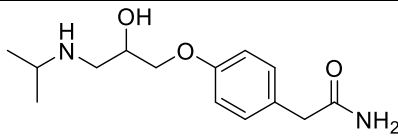
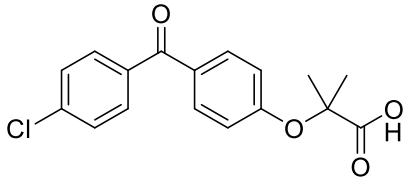
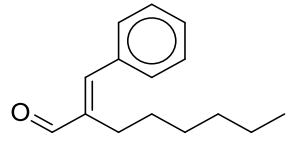
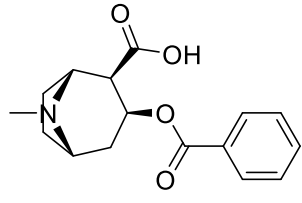
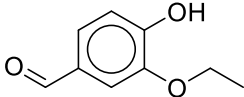
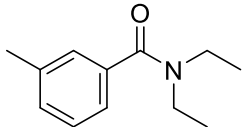
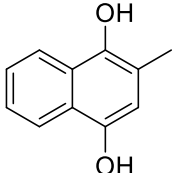
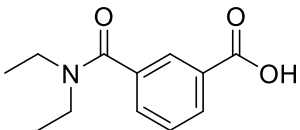
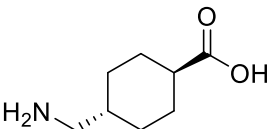
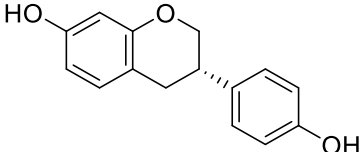
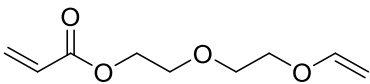
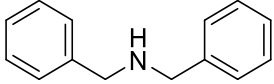
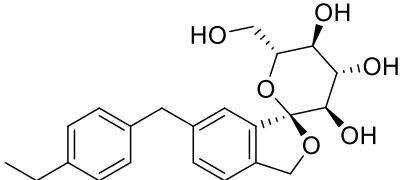
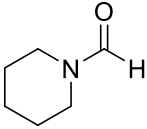
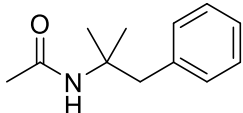


Figure 5

Table 1. CECs detected in the influent wastewater employing the database of Norman network and then refined by Sciex OS software.

Organic pollutant	Molecular formula	Molecular configuration	Use
1,5 dihydro - 3 - ethyl - 4- methyl - 2H - pyrrol- 2 one	C ₇ H ₁₁ NO		Chemical used for the preparation of Glimepiride, and its intermediates, used as antidiabetic agents.
N-(1,1,3,3 tetramethyl butyl) acrylamide	C ₁₁ H ₂₁ NO		Cosmetic (protects against UV rays)
7 hydroxyheptane 1 nitrile	C ₇ H ₁₃ NO		Industrial and manufacturing fields
Atenolol acid	C ₁₄ H ₂₁ NO ₄		“Cardioselective” <u>beta-adrenoceptor</u> antagonist
Triethyl phosphate	C ₆ H ₁₅ O ₄ P		Plasticizer
Atenolol	C ₁₄ H ₂₂ N ₂ O ₃		Beta blocker pharmaceutical
Fenofibric acid	C ₁₇ H ₁₅ ClO ₄		Food diets
2-(phenylmethylidene)oc tanal	C ₁₅ H ₂₀ O		Natural product found in <i>Plectranthus glabratus</i>
Benzoylecgonine	C ₁₆ H ₁₉ NO ₄		Metabolite of cocaine

3-ethoxy-4-hydroxybenzaldehyde	C ₉ H ₁₀ O ₃		Food additive
Deet	C ₁₂ H ₁₇ NO		Mosquito repellent
Menadiol	C ₁₁ H ₁₀ O ₂		Food supplements
DEET-Carboxylic Acid	C ₁₂ H ₁₇ NO		Mosquito repellent
Tranexamic acid	C ₈ H ₁₅ NO ₂		Used in medicine against fibrinolysis
Equol	C ₁₅ H ₁₄ O ₃		Food supplements
2-(2-vinyloxyethoxy)ethyl acrylate	C ₉ H ₁₄ O ₄		Chemical used to make special coatings, dehydrating agents, adhesives
Dibenzylamine	C ₁₄ H ₁₅ N		Chemical harmful to aquatic life
Tofogliflozin	C ₂₂ H ₂₆ O ₆		Drug for the treatment of diabetes mellitus
Formylpiperidine	C ₆ H ₁₁ NO		Solvent
N-(2-methyl-1-phenyl-2-propanyl)acetamide	C ₁₂ H ₁₇ NO		Derives from phentermine that is similar to amphetamines and acts at

			the level of the nervous system
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