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# New diphenylmethane derivatives as Peroxisome Proliferator-Activated Receptor alpha/gamma dual agonists endowed with anti-proliferative effects and mitochondrial activity 

Luca Piemontese ${ }^{\text {a,b }}$, Carmen Cerchia ${ }^{\mathrm{c}}$, Antonio Laghezza ${ }^{\mathrm{a}}$, Pamela Ziccardi ${ }^{\mathrm{d}}$, Sabina Sblano ${ }^{\mathrm{a}}$, Paolo Tortorella ${ }^{\text {a }}$, Vito Iacobazzi ${ }^{\text {e,f }}$, Vittoria Infantino ${ }^{\mathrm{e}, \mathrm{g}}$, Paolo Convertini ${ }^{\mathrm{e}, \mathrm{g}}$, Fabrizio Dal Piaz, ${ }^{\text {h }}$ Angelo Lupo ${ }^{\text {d }}$, Vittorio Colantuoni ${ }^{\text {d }}$, Antonio Lavecchia ${ }^{\text {c }}$, ${ }^{\text {Fulvio Loiodice }}{ }^{\text {a,* }}$

${ }^{\text {a }}$ Dipartimento Farmacia-Scienze del Farmaco, Università degli Studi di Bari "Aldo Moro", via Orabona 4, 70125 Bari (Italy)
${ }^{\mathrm{b}}$ Istituto di Scienze delle Produzioni Alimentari, Consiglio Nazionale delle Ricerche, via Amendola 122/O, 70126 Bari (Italy)
c Dipartimento di Farmacia, "Drug Discovery" Laboratory, Università degli Studi di Napoli "Federico II", via D. Montesano 49, 80131 Napoli (Italy)
${ }^{\mathrm{d}}$ Dipartimento di Scienze e Tecnologie, Università degli Studi del Sannio, via Port'Arsa 11, 82100 Benevento (Italy)
${ }^{\text {e }}$ Dipartimento di Bioscienze, Biotecnologie e Biofarmaceutica, Laboratorio di Biochimica e Biologia Molecolare, Università degli Studi di Bari "Aldo Moro", via Orabona 4, 70125 Bari (Italy)
${ }^{\mathrm{f}}$ Istituto di Biomembrane e Bioenergetica, Consiglio Nazionale delle Ricerche, via Orabona 4, 70125 Bari (Italy)
${ }^{\mathrm{g}}$ Dipartimento di Scienze, Università della Basilicata, viale dell'Ateneo Lucano 10, 85100 Potenza (Italy)
${ }^{\mathrm{h}}$ Dipartimento di Medicina e Chirurgia, Università di Salerno, via Giovanni Paolo II 132, 84084 Fisciano, SA, (Italy)

Corresponding Authors: Prof. Fulvio Loiodice, Dipartimento di Farmacia-Scienze del Farmaco, Università degli Studi di Bari "Aldo Moro", Via Orabona 4, 70126 Bari, Italy; phone: +39 0805442778, e-mail: fulvio.loiodice@uniba.it; Prof. Antonio Lavecchia, Dipartimento di Farmacia, "Drug Discovery" Laboratory, Università degli Studi di Napoli "Federico II", Via Domenico Montesano 49, 80131 Napoli, Italy; phone: +39 081-678613, e-mail: antonio.lavecchia@unina.it.

Abbreviations: PPAR, peroxisome proliferator-activated receptor; LBD, ligand binding domain; TZDs, thiazolidinediones; CAC, carnitine/acylcarnitine carrier; CPT1, carnitine-palmitoyltransferase 1; PGC-1 $\alpha$, peroxisome proliferator-activated receptor gamma coactivator 1-alpha; SPR, Surface Plasmon Resonance.


#### Abstract

We screened a short series of new chiral diphenylmethane derivatives and identified potent dual PPAR $\alpha / \gamma$ partial agonists. As both enantiomers of the most active compound $\mathbf{1}$ displayed an unexpected similar transactivation activity, we performed docking experiments to provide a molecular understanding of their similar partial agonism. We also evaluated the ability of both enantiomers of $\mathbf{1}$ and racemic $\mathbf{2}$ to inhibit colorectal cancer cells proliferation: ( $S$ )-1 displayed a more robust activity due, at least in part, to a partial inhibition of the Wnt/ $\beta$-catenin signalling pathway that is upregulated in the majority of colorectal cancers. Finally, we investigated the effects of $(R) \mathbf{- 1},(S)$ $\mathbf{1}$ and $(R, S)-\mathbf{2}$ on mitochondrial function and demonstrated that they activate the carnitine shuttle system through upregulation of carnitine/acylcarnitine carrier (CAC) and carnitine-palmitoyl-transferase 1 (CPT1) genes. Consistent with the notion that these are PPAR $\alpha$ target genes, we tested and found that PPAR $\alpha$ itself is regulated by a positive loop. Moreover, these compounds induced a significant mitochondrial biogenesis. In conclusion, we identified a new series of dual PPAR $\alpha / \gamma$ agonists endowed with novel anti-proliferative properties associated with a strong activation of mitochondrial functions and biogenesis, a potential therapeutic target of the treatment of insulin resistance.


Keywords: PPAR, docking experiments, mitochondrial biogenesis, anti-proliferative effects, gene expression analysis.

## 1. Introduction

Peroxisome proliferator-activated receptors (PPARs) are ligand-activated transcription factors belonging to the nuclear hormone receptor superfamily. The interaction with specific ligands of these receptors elicits a conformational change of the ligand binding domain (LBD) that allows the recruitment of coactivators on the promoter regions of target genes. The enhanced expression of these genes is directly or indirectly involved in energy homeostasis, lipid and carbohydrate metabolism.[16] Three PPAR subtypes (designated $\alpha, \gamma$ and $\delta$ ) have been identified so far with different tissue distribution and ability to modulate diverse physiological functions. The activation of PPAR $\alpha$ by hypolipidemic agents (e.g. fenofibrate) stimulates fatty acid $\beta$-oxidation, increases high density lipoprotein (HDL), and decreases very low and low density lipoproteins (VLDL and LDL) in blood. [7,8] The thiazolidinedione (TZD) class of antidiabetic agents (rosiglitazone, pioglitazone) are PPAR $\gamma$ agonists [9] acting as insulin sensitizers in target tissues and lowering glucose and fatty acid levels in type 2 diabetic patients. However, despite their proven benefits, PPAR $\gamma$ full agonists possess a number of deleterious side effects such as weight gain, peripheral edema, increased risk of congestive heart failure and higher rate of bone fracture.[10-13] Recently, one of the most prescribed TZD, rosiglitazone, has been withdrawn from the market by the EMEA and european patients have been forced to change their antidiabetic therapy.[14] For this reason, the current therapeutic frontier is to search for PPAR $\gamma$ partial agonists or modulators that retain the therapeutic activity while displaying reduced adverse effects.[15-17] PPAR $\delta$, albeit ubiquitously expressed, has functions not fully elucidated. Recent studies have identified a role in cholesterol metabolism, adipocyte differentiation, neuronal function and colon cancer.[18] To date, no PPAR $\delta$ agonist has been developed and the clinical potential of its therapeutic targeting remains to be clearly determined. In line with this reasoning, in the last few years many efforts have been directed toward the design of new molecules that combine the insulin-sensitizing effect of PPAR $\gamma$ and the additional
lipid-modifying activity of PPAR $\alpha$. In fact, the development of dual PPAR $\alpha / \gamma$ agonists has been considered a very attractive option in the treatment of dyslipidemic type 2 diabetes.[19-20] A number of drugs acting as full PPAR $\alpha$ agonists and partial PPAR $\gamma$ agonists have recently been developed. However, except for saroglitazar, which has recently been approved in India for the treatment of diabetic dyslipidemia and hypertriglyceridemia with type 2 diabetes not controlled by statin therapy,[21] none of them has progressed further to preclinical or phase II trials. In search for novel pharmacophores able to provide a suitable dual activation of these PPAR subtypes, we designed a new class of ligands characterized by the presence of the 2-oxy-propanoic acid moiety linked to the diphenylmethane skeleton that has already shown its usefulness as a multi-template for nuclear receptors.[22-24] The diphenylmethane skeleton was fused, through an oymethylene linker, with the 2-phenyl-1,3-oxazole nucleus that is typical of a number of PPAR $\alpha / \gamma$ agonists.[25] The combination of these two structural moieties provided compounds $\mathbf{1}$ and $\mathbf{2}$ (Figure 1). With the aim to obtain compounds endowed with activity also toward the farnesoid X receptor (FXR), we introduced into the structure of $\mathbf{1}$ and $\mathbf{2}$ some chemical modifications similar to those reported in some diphenylpentane-based phenoxyacetic acids acting as FXR agonists.[24] Modulation of FXR may be a valuable therapeutic approach for various pathophysiological conditions. In particular, FXR activation showed promising results in vitro and in vivo for the treatment of metabolic diseases such as type 2 diabetes.[26-28]

To this goal, we investigated the effects on the activity of $\mathbf{1}$ and $\mathbf{2}$ originating from the presence of two chlorine atoms on the ortho position of the distal aromatic ring, of a methyl on the oxazole nucleus, and of two ethyl groups on the benzylic carbon. In this way, we prepared the diphenylpentane derivatives 3-6 displayed in Figure 1. In addition, considering the well-known sensitivity of PPARs to stereochemistry, we explored the influence of absolute configuration on receptor activation. In the present series, only diphenylmethane derivatives showed potent dual PPAR $\alpha / \gamma$
agonist activity with a $20-40 \%$ efficacy with respect to reference compounds. Surprisingly, both enantiomers of the most active compound $\mathbf{1}$ displayed similar activity. Docking experiments were performed to provide a molecular explanation for this unexpected effect and also to rationalize the lack of activity of diphenylpentane derivatives. In addition, we tested whether both enantiomers of $\mathbf{1}$ and racemate $\mathbf{2}$ were endowed with an antiproliferative activity mediated by PPAR $\gamma$. This was in line with the large body of data showing such a ligand-directed function in cells in vitro and in animal models in vivo.[29-33] Furthermore, we investigated whether (S)-1 and $(R)$ - $\mathbf{1}$ could affect mitochondrial carnitine/acylcarnitine carrier (CAC) and carnitine-palmitoyl-transferase 1 (CPT1) gene expression. These are two components of the carnitine shuttle system essential for fatty acids mitochondrial transport and subsequent $\beta$-oxidation. Transcription of the CAC gene has recently been shown to be enhanced by statins and fibrates, providing a novel contribution to the understanding of their hypolipidemic action.[34-37] Finally, we evaluated the effect of $(R)-\mathbf{1},(S) \mathbf{- 1}$ and $(R, S) \mathbf{- 2}$ on mitochondrial biogenesis. In fact, the increase of mitochondrial mass and oxidative activity is currently viewed as a potential therapeutic approach for the treatment of insulin resistance.


| Compound | $\mathbf{R}_{\mathbf{1}}$ | $\mathbf{R 2}_{\mathbf{2}}$ | $\mathbf{R 3}$ |
| :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | H | H | H |
| $\mathbf{2}$ | H | $\mathrm{CH}_{3}$ | H |
| $\mathbf{3}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | H | H |
| $\mathbf{4}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | H |
| $\mathbf{5}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | H | Cl |


| $\mathbf{6}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | Cl |
| :--- | :--- | :--- | :--- |

Figure 1. Diphenylmethane derivatives reported in the present study.

## 2. Chemistry

The synthesis of compounds 1-6 started with the preparation of the intermediates 7-11 (Scheme 1). The condensation of benzamide or 2,6-dichlorobenzamide with 1,3-dichloroacetone, in a vial at 180 ${ }^{\circ} \mathrm{C}$ gave compounds $\mathbf{7}$ and $\mathbf{8}$, respectively. Compounds $\mathbf{9 - 1 0}$ were obtained by condensation of butan-2,3-dione with hydroxylamine hydrochloride in the presence of sodium acetate and subsequent reaction with benzaldehyde or 2,6-dichloro-benzaldehyde. The final treatment with $\mathrm{POCl}_{3}$ allowed to obtain the target intermediates. Finally, the reaction between phenol and 3-pentanone in the presence of methansulfonic acid gave the intermediate $\mathbf{1 1}$.



Scheme 1. i) 1,3-dichloroacetone, $180{ }^{\circ} \mathrm{C}$; ii) $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{CH}_{3} \mathrm{COONa}, \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$, rt; iii) benzaldheyde or 2,6dichlorobenzaldheyde, $4 \mathrm{M} \mathrm{HCl} /$ dioxane, rt; iv) $\mathrm{POCl}_{3}$, anhydrous $\mathrm{CHCl}_{3}$, reflux; v) 3-pentanone, methansulfonic acid, rt.

The reaction of the commercially available bis(4-hydroxyphenyl)methane or $\mathbf{1 1}$ with ethyl or methyl 2-bromopropionate in the presence of $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ led to compounds $\mathbf{1 2 - 1 3}$. These derivatives were condensed with the intermediates $\mathbf{7 - 1 0}$ in the presence of NaH to give esters $\mathbf{1 4 - 1 9}$ whose alkaline hydrolysis afforded the target acids 1-6 (Scheme 2).


Scheme 2. i) ethyl or methyl 2-bromopropionate, $\mathrm{Cs}_{2} \mathrm{CO}_{3}$, acetonitrile, reflux; ii) $95 \% \mathrm{NaH}$, anhydrous THF, reflux; iii) 1 N NaOH , THF, rt.

The synthesis of the enantiomers of $\mathbf{1}$ started with the condensation between bis(4hydroxyphenyl)methane and $(R)$ - or ( $S$ )-methyl lactate via a Mitsunobu reaction which is known to occur with inversion of configuration. The so obtained methyl esters $(R)$ - $\mathbf{1 2}$ and $(S)$ - $\mathbf{1 2}$ were condensed with 7 following the same procedure shown in scheme 2 which finally afforded the target acids ( $R$ )-1 and ( $S$ )-1 (Scheme 3).


Scheme 3. i) ( $R$ )- or (S)-methyl lactate, DIAD, triphenylphosphine, anhydrous THF, rt; ii) $95 \% \mathrm{NaH}$, anhydrous THF, reflux; iii) 1 N NaOH , THF, rt.

## 3. Results and discussion

Compounds 1-6 were evaluated for their agonist activity toward the human PPAR $\alpha(\mathrm{hPPAR} \alpha)$, $\operatorname{PPAR} \gamma(\mathrm{hPPAR} \gamma), \operatorname{PPAR} \delta(\mathrm{hPPAR} \delta)$, and FXR (hFXR) subtypes. For this purpose, GAL4-PPAR or GAL4-FXR chimeric receptors were transiently transfected in HepG2 cells according to a previously reported procedure.[38] The results were compared with those obtained with $\mathrm{Wy}-14,643$, rosiglitazone, L-165,041, and chenodeoxycholic acid used as reference compounds for $\operatorname{PPAR} \alpha, \operatorname{PPAR} \gamma, \operatorname{PPAR} \delta$, and FXR respectively (Table 1). Maximum fold induction obtained with the reference agonist was set at $100 \%$.

Table 1. Activity of the tested compounds in a cell-based transactivation assay.

| Cpd | PPAR $\alpha$ |  | PPAR $\gamma$ |  | PPAR \% |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | EC50 ( $\mu \mathrm{M}$ ) | $\mathbf{E}_{\max }^{a}$ | EC50 ( $\mu \mathrm{M}$ ) | $\mathbf{E}_{\max }^{a}$ | EC50 $(\mu \mathrm{M})$ | $\mathbf{E}_{\max }^{a}$ |
| 1 | $0.039 \pm 0.013$ | $47 \pm 13$ | $0.29 \pm 0.08$ | $26 \pm 3$ | nc | 15 |
| (S)-1 | $0.046 \pm 0.023$ | $39 \pm 3$ | $0.30 \pm 0.21$ | $17 \pm 5$ | nc | 11 |
| (R)-1 | $0.042 \pm 0.031$ | $44 \pm 9$ | $0.28 \pm 0.16$ | $26 \pm 4$ | nc | 14 |
| 2 | $0.100 \pm 0.010$ | $34 \pm 14$ | $0.29 \pm 0.06$ | $20 \pm 3$ | nc | 13 |
| 3 | nc | 40 | nc | 10 | ia | ia |
| 4 | nc | 17 | nc | 3 | ia | ia |
| 5 | nc | 18 | nc | 10 | ia | ia |
| 6 | nc | 24 | nc | 9 | ia | ia |
| Wy-14,643 | $1.6 \pm 0.3$ | $100 \pm 10$ | ia | ia | ia | ia |
| Rosiglitazone | ia | ia | $0.04 \pm 0.02$ | $100 \pm 9$ | ia | ia |
| L-165,041 | ia | ia | ia | ia | $0.02 \pm 0.002$ | $100 \pm 4$ |

${ }^{a}$ Efficacy values were calculated as the percentage of the maximum obtained fold induction relative to the reference compounds. ia = inactive at tested concentrations (up to $2 \mu \mathrm{M}$ ). nc = not computable; the activity, in fact, increases with increasing concentrations up to $2 \mu \mathrm{M}$, above which the activity begins to decrease.

Unexpectedly, all compounds exhibited no activity on FXR (data not shown), whereas most compounds displayed partial agonism on PPARs. Specifically, compound $\mathbf{1}$ turned out to be a potent partial agonist on PPAR $\alpha$ and PPAR $\gamma$ subtypes; at higher doses it showed moderate activity even on PPAR $\delta$; above $2 \mu \mathrm{M}$ the activity started to decline due to the toxicity toward the cell line employed in the assay. The presence of an additional methyl group on the oxazole ring (compound
2) did not significantly affect the activity, which resulted 2.5 times lower than $\mathbf{1}$ only on PPAR $\alpha$. The introduction of the ethyl groups on the methylene bridge between the two phenyl rings (compounds 3 and 4) as well as the presence of two chlorine atoms on the distal benzene linked to oxazole (compounds $\mathbf{5}$ and $\mathbf{6}$ ) dramatically reduced the activity. In fact, PPAR $\alpha$ and PPAR $\gamma$ activity only marginally increased upon exposure to increasing concentrations of these compounds up to 2 $\mu \mathrm{M}$, above which the activity started to decrease due to cytotoxicity. By contrast, no activity was detected toward the PPAR $\delta$ subtype. On the whole, $\mathbf{1}$ turned out to be the one with the best pharmacological profile. For this reason, we decided to prepare its stereoisomers to evaluate whether the absolute configuration of the stereocenter close to the carboxylic function could affect the activity. Surprisingly, $(R) \mathbf{- 1}$ and $(S) \mathbf{- 1}$ displayed the same effects toward all PPAR subtypes. To evaluate if the observed results were dependent upon the direct interaction of $(R)-\mathbf{1}$ and $(S)$ - $\mathbf{1}$ with the receptors, we performed a surface plasmon resonance (SPR) assay; different concentrations of the two stereoisomers were injected on each of the three PPAR subtypes, singularly immobilized on sensor chips.[39] Wy-14,643, rosiglitazone and L-165,041 were used as controls. The results clearly indicated a high affinity of $(R) \mathbf{- 1}$ and $(S) \mathbf{- 1}$ towards PPAR $\alpha$ (Figure 2 and Table 2); both compounds also effectively bound PPAR $\gamma$, whereas no interaction with PPAR $\delta$ was observed.

Table 2. Thermodynamic $\left(\mathrm{K}_{\mathrm{D}}\right)$ and kinetics $\left(\mathrm{K}_{\mathrm{d}}\right)$ dissociation constants measured by SPR for the interaction of the stereoisomers of $\mathbf{1}$ with PPAR subtypes.

|  | PPAR $\alpha$ |  | PPAR $\gamma$ |  | PPAR $\delta$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Cpd | $\mathbf{K}_{\mathrm{D}}(\mathbf{n M})$ | $\mathbf{K}_{\text {d }}\left(\mathbf{s}^{-1}\right)$ | $\mathrm{K}_{\mathrm{D}}(\mathbf{n M})$ | $\mathbf{K}_{\text {d }}\left(\mathbf{s}^{-1}\right)$ | $\mathrm{K}_{\mathrm{D}}(\mathbf{n M})$ | $\mathbf{K}_{\text {d }}\left(\mathbf{s}^{-1}\right)$ |
| (S)-1 | $0.75 \pm 0.18$ | $0.016 \pm 0.009$ | $4.9 \pm 1.1$ | $0.099 \pm 0.023$ | n.b. | n.b. |
| (R)-1 | $0.86 \pm 0.11$ | $0.025 \pm 0.007$ | $3.7 \pm 0.5$ | $0.095 \pm 0.016$ | n.b. | n.b. |
| Wy-14,643 | $18.9 \pm 5.6$ | $0.36 \pm 0.12$ | $1297.3 \pm 19.6$ | $2.3 \pm 0.6$ | n.b. | n.b. |
| Rosiglitazone | n.b. | n.b. | $1.2 \pm 0.7$ | $0.145 \pm 0.030$ | n.b. | n.b. |
| L-165,041 | n.b. | n.b. | n.b. | n.b. | $76.4 \pm 6.1$ | $0.43 \pm 0.08$ |

n.b. = no binding observed.


Figure 2. SPR sensorgrams resulting from the injection of different amounts of compounds $(R) \mathbf{- 1}$ and $(S) \mathbf{- 1}$ on immobilized $\operatorname{PPAR} \alpha$ (a and b), and PPAR $\gamma(\mathbf{c}$ and $\mathbf{d}$ ).

A comparison between SPR and the cell-based assay data indicates that the transactivation of PPAR $\alpha$ and PPAR $\gamma$ by $(R) \mathbf{- 1}$ and $(S) \mathbf{- 1}$ depends on a direct interaction of these compounds with the two PPAR subtypes. Moreover, SPR results show a very similar behavior from the two stereoisomers towards the two receptors, both in terms of thermodynamic stability of the resulting
complexes, as inferred by the comparable $\mathrm{K}_{\mathrm{D}}$ measured, and in terms of binding mode, as suggested by the observed $\mathrm{K}_{\mathrm{d}}$.[40]

In order to rationalize the unexpected similar activity displayed by $(R) \mathbf{- 1}$ and $(S) \mathbf{- 1}$ toward all PPAR subtypes and to explain the lack of activity of diphenylpentane derivatives 3-6, we performed docking experiments.

## 4. Docking calculations

In an effort to understand the molecular basis for the observed partial agonism of $(S) \mathbf{- 1}$ and $(R) \mathbf{- 1}$ toward both PPAR $\alpha$ and PPAR $\gamma$, we undertook docking studies using GOLD 5.2.2 program [41] in combination with the ChemPLP [42] scoring function (rescoring with ChemScore).[43] Docking experiments were carried out into the X-ray crystal structures of PPAR $\gamma$ ligand binding domain (LBD) complexed to partial agonist LT127 (PDB code: 2I4P) [44] and the PPAR $\alpha$ LBD complexed to partial agonist aleglitazar (PDB: 3G8I).[45] The docking experiments predicted that both ( $S$ )-1 and ( $R$ )-1 enantiomers favorably bind to PPAR $\gamma$ LBD, adopting a similar U-shaped conformation that wraps around H3. Figure 3a,b clearly illustrates the binding poses of $(S) \mathbf{- 1}$ and $(R) \mathbf{- 1}$, that appear almost identical when the LBD is superposed (Figure 3c). Superimposition of the two enantiomers on the co-crystal structure of full agonist rosiglitazone bound to PPAR $\gamma$ LBD (PDB code: 2PRG) revealed that the ligands do not make direct contact with residue Y 473 in the activation function (AF-2) domain located on H12, like other partial agonists including GW0072 and nTZDpa (Figure 3d). The location of the polar carboxylate group is too far away from the H 12 in comparison to the thiazolidinedione moiety of rosiglitazone. The carboxylate group of both (S)-1 and (R)-1 forms H-bonds with Y327 $\mathrm{OH}, \mathrm{H} 323 \mathrm{~N}^{\varepsilon 2}$, and S289 OH atoms (Figure 3a,b), while the ether oxygen of the ligands forms an additional H-bond with S289 OH. The remainder structure wraps around H3 and buries the 2phenyloxazole tail into a lipophilic pocket formed by H3, H5, the
$\beta$-sheet and the $\Omega$-loop, a flexible loop region that links H2' to H3. ( $S$ )-1 and ( $R$ )-1 make important hydrophobic interactions with residues K265, G284, F287, C285, H266, R288, L330, L333, I341, and S342. The central phenoxy oxygen accepts a H -bond from the backbone NH of S342 side chain located within the $\beta$-sheet region of the receptor. In addition, the distal ring of 2-phenyloxazole tail appears to be optimally oriented for a favorable $\pi-\pi$ stacking interaction with F287 positioned on H3, which contributes to further increase H 3 stabilization. The lack of interactions with the AF-2 and the indirect stabilization of the PPAR $\gamma$ AF-2 surface via the alternative regions could provide the structural basis for the partial agonist properties of compounds ( $S$ )-1 and $(R)$-1. Co-crystallography, mutagenesis and hydrogen/deuterium exchange (HDX) studies have all displayed that PPAR $\gamma$ full agonists activate the receptor through direct interactions with the Y473 residue located in the AF2 domain of H12 of the receptor [46] so to induce conformational changes that facilitate the recruitment of coactivators and hence transcriptional activation. In contrast, both enantiomers bind in the entryend of the ligand-binding pocket at a distance from H12 that precludes direct contact with Y473. Rather, the high-affinity interactions between the two ligands and receptor originate from a direct H bond with the main-chain NH of S 342 of the $\beta$-sheet and from the numerous hydrophobic van der Waals contacts with residues of H 3 and $\Omega$-loop. These interactions could provoke destabilization of H12 and stabilization of H 3 affecting the recruitment of coactivators and, in turn, PPAR $\gamma$ transactivation potential,[46,47] possibly explaining the partial agonism of $(S) \mathbf{- 1}$ and $(R) \mathbf{- 1}$ ligands.


Figure 3. a) Binding mode of enantiomer ( $S$ )-1 (magenta sticks) into the PPAR $\gamma$ binding site represented as a bluemarine ribbon model. b) Binding mode of enantiomer ( $R$ ) $\mathbf{- 1}$ (yellow sticks) into the PPAR $\gamma$ binding site. c) $\mathrm{C}^{\alpha}$ superposition of the complexes of PPAR $\gamma$ with $(S)-\mathbf{1}$ and $(R) \mathbf{- 1}$. d) $\mathrm{C}^{\alpha}$ superposition of the complexes of PPAR $\gamma$ with $(S)-\mathbf{1},(R)-\mathbf{1}$ and rosiglitazone (cyan sticks, PDB code 2PRG). Only amino acids located within $4 \AA$ of the bound ligand are displayed (bluemarine sticks) and labeled. The $\Omega$-loop, a flexible loop region between H 2 ' and H 3 , and the $\beta$-sheet region of the LBD, are displayed. H-bonds discussed in the text are depicted as dashed black lines.

Similar docking experiments were undertaken to gain insight into the possible binding mode of $(R)$ -
1 and $(S)-\mathbf{1}$ to PPAR $\alpha$ LBD. An extensive survey of reported X-ray crystal structures of ligands complexed with PPAR $\alpha \operatorname{LBD}[45,48]$ shows that one or two water molecules bridge the T 279 OH group on H 3 with the A 333 NH or $\mathrm{L} 331 \mathrm{C}=\mathrm{O}$ backbone on the $\beta$-sheet, mediating contacts between H3 and $\beta$-sheet, likely stabilizing the structure of the receptor LBD. As two intervening water molecules are also present in the aleglitazar/PPAR $\alpha$ complex and facilitate H -bonds between the two regions stabilizing the LBD, both water molecules were included in the docking experiments.

Both $(R)-\mathbf{1}$ and (S)-1 enantiomers fit well within the PPAR $\alpha$ binding pocket. As expected, the carboxylate head group, together with the ether oxygen, form the well recognized H -bonding network with residues H440, Y464, Y314, and S280, which is believed to be critical for PPAR $\alpha$ ligands activity (Figure 4). In addition to these standard carboxylate interactions with the binding pocket, two further H-bonds are also observed: one between the central phenoxy oxygen and T279

OH group and the other between the oxazole nitrogen and the NH backbone of A 333 on the $\beta$-sheet. It is noteworthy that the OH group of T 279 makes an indirect H -bond with the $\mathrm{L} 331 \mathrm{C}=\mathrm{O}$ backbone via two water molecules, thereby further stabilizing the $\beta$-sheet region of PPAR $\alpha$. The remaining interactions between ligand and protein are hydrophobic in nature; in particular, the methyl group binds into a hydrophobic pocket defined by F273, C276, Q277, I354, and V444, while the benzylphenoxy moiety is surrounded by hydrophobic residues V324, L321, I354, and three sulfurcontaining amino acids, C276, M355, and M330, which form strong attractive sulfur-arene interactions;[49] the 2-phenyloxazole tail makes a good fit into the lower part of the binding pocket and would be expected to positively contribute to overall binding through van der Waals interactions with the hydrophobic residues L258, V255, A333, Y334, and C275. All these data support that $(R)-\mathbf{1}$ and (S)-1 act as partial PPAR $\alpha$ agonists and suggest that their binding induces conformational changes in the receptor LBD resulting in a distinct pattern of cofactor recruitment and selective gene induction ability.

Finally, docking experiments were also carried out to rationalize the lack of activity of compounds 36 toward either PPAR $\alpha$ and PPAR $\gamma$ subtypes. The introduction of two bulky ethyl groups on the methylene bridge between the two phenyl rings was not tolerated due to severe steric clashes with specific residues (C285 and M364 for PPAR $\gamma$; and L321, I354 and K358 for PPAR $\alpha$ ) present in the corresponding PPAR binding cleft. These results explicitly account for the lack of ability of 3-6 to bind PPAR $\alpha$ and PPAR $\gamma$.


Figure 4. a) Binding mode of enantiomer ( $S$ )-1 (a, magenta sticks) and ( $R$ ) - $\mathbf{1}$ (b, yellow sticks) into the PPAR $\alpha$ binding site represented as a green ribbon model. Only amino acids located within $4 \AA$ of the bound ligand are displayed (green sticks) and labeled. The $\Omega$-loop, a flexible loop region between H 2 ' and H 3 , and the $\beta$-sheet region of the LBD, are displayed. The ordered water molecules bridging the T279 OH group on H 3 and the $\mathrm{L} 331 \mathrm{C}=\mathrm{O}$ backbone on $\beta$-sheet are displayed as red spheres. H-bonds discussed in the text are depicted as dashed black lines.

## 5. Compounds ( $S$ )-1, ( $R$ )-1 and ( $R, S$ )-2 inhibit the growth of the human CRC cell line HT29

Having all compounds of this series displayed cytotoxicity in the transactivation assays, we decided to evaluate their antiproliferative activity on a different cell type such as the colon cancer derived HT-29 cells which express PPAR $\gamma$ at high levels [38d] and assess differences between full and partial PPAR $\gamma$ agonists. This last point may be relevant as the antiproliferative activity, and thus the antineoplastic benefits, of the full agonists TZDs have produced so far conflicting results in clinical trials, justifying the need to further investigate the anticancer potential of novel PPAR $\gamma$ agonists.[50] For this purpose, we chose both enantiomers of $\mathbf{1}$ and racemic $\mathbf{2}$ and compared their activity with that of the full agonist rosiglitazone (Figure 5).


Figure 5. Evaluation of the antiproliferative potential of $(S) \mathbf{- 1 , (}(R) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ in colon cancer cells. To assess the effect of the different compounds, exponentially growing HT-29 cells were exposed to vehicle alone or to 1,5 and $10 \mu \mathrm{M}$ of each drug for 24,48 and 72 hs , collected and counted. The data shown here are the mean $\pm \mathrm{SD}$ of three independent experiments performed in duplicate. Results were similar in three independent experiments. ${ }^{*} \mathrm{p}<0.05, * * \mathrm{p}<0.01, * * *$ $p<0.001$, respectively, when compared to control.

The ( $S$ ) $\mathbf{- 1}$ compound elicited a robust antiproliferative activity that was dose- and time-dependent reaching the maximum ( $40 \%$ of residual cell vitality) between 5 and $10 \mu \mathrm{M}$ at 72 hs . ( $R$ )-1 also exhibited a dose- and time-dependent activity that was, however, less pronounced than $(S)$ - $\mathbf{1}$ : the residual cell vitality upon $5 \mu \mathrm{M}(S)-1$ treatment was about $45 \%$ as compared with $60 \%$ obtained with $(R)-\mathbf{1}$. The even stronger effect obtained with $10 \mu \mathrm{M}(S)-\mathbf{1}$ and $(R) \mathbf{- 1}$ at $\mathbf{7 2}$ hs may be partially referred to cytotoxic effects. The most striking result obtained in this experimental setting was the different growth inhibitory effect of $(S) \mathbf{- 1}$ with respect to the racemic compound 2 and the full agonist rosiglitazone: the residual cell vitality obtained with these two latter compounds at 72 hs was, in fact, higher than the ( $S$ )-1 enantiomer ( $70 \%$ and $65 \%$ vs $40 \%$ respectively).

We further investigated the cytotoxic effects of both enantiomers of $\mathbf{1}$ and racemic $\mathbf{2}$ in HEK293T, a cell line derived from a human embryonic kidney immortalized by transformation with adenovirus 5 DNA expressing a transfected SV40 large T antigen and generally considered as a normal cell line. Exposure to $(S) \mathbf{- 1}(10 \mu \mathrm{M})$ for 48 hs elicited in HEK293T an antiproliferative effect that was
milder (about $75 \%$ of cells were viable) than HT- 29 cells (about $40 \%$ of cells were viable) treated in parallel (Figure 6A). To confirm these data, in both cellular types, we investigated $\mathrm{p} 21^{\text {wafl/cip1 }}$ expression, one of the well-known regulators of the cell cycle. We found that $10 \mu \mathrm{M}(S)$ - $\mathbf{1}$ treatment for 48 hs induced higher expression of the protein in HT-29 cells compared to $(R)-\mathbf{1}$ and $(R, S)-\mathbf{2}$, whereas treatment with all three compounds slightly affected $\mathrm{p} 21^{\text {wafl/cip1 }}$ expression in HEK293T cells (Figure 6B). The above described results indicate that the colon cancer derived cell line HT-29 appears to be more susceptible to (S)-1 than normal HEK293T cells, considered as normal, thus implying the possibility of future experimentations for these compounds in preclinical and clinical trials.


Figure 6. (A) Evaluation of the antiproliferative potential of $(S) \mathbf{- 1},(R) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ in human embryonic kidney HEK293T and colon cancer HT-29 cells. To assess the effect of the different compounds, exponentially growing HEK293T and HT-29 cells were exposed or not to 5 and $10 \mu \mathrm{M}$ of each drug for 24 and 48 hs , collected and counted. The data shown here are the mean $\pm$ SD of three independent experiments performed in duplicate. Results were similar in three independent experiments. $* \mathrm{p}<0.05, * * \mathrm{p}<0.01$, ${ }^{* * *} \mathrm{p}<0.001$, respectively, when compared to vehicle treated cells. (B) Total protein extracts from HT-29 and HEK293T cells untreated or treated with $10 \mu \mathrm{M}(S)-\mathbf{1},(R)-\mathbf{1}$ and $(R, S)$ 2 for 48 hs were analysed by Western blot with an anti-p21 waf1/cip1 antibody. An anti- $\beta$-actin antibody was used as a loading control. The bar graphs represent the mean $\pm$ SD of $\mathrm{p} 21 / \beta$-actin of at least three independent experiments. The Western blotting assays reported here are representative of a single experiment.

## 6. Compounds $(S)-1,(R)-1$ and $(R, S)-2$ interfere with the $\beta$-catenin/TCF pathway and determine cell proliferation arrest

The strong cell growth inhibition exhibited by $(S)-\mathbf{1}$ associated with the weak transactivation potential prompted us to investigate whether this compound might activate antiproliferative pathways that are not or are only partially activated by classical PPAR $\gamma$ full agonists. The Wnt/ $\beta$-catenin signalling is overexpressed in the majority of colorectal cancers and its dysregulation is one of the leading causes of such a malignancy.[51] We assessed the ability of our compounds to activate this pathway in HEK293T cells in which the Wnt/ $\beta$-catenin pathway is not active and can be induced upon exposure to selected stimuli such as lithium chloride (LiCl).[52] Cells were transiently transfected with the Top-FLASH reporter plasmid carrying the luciferase gene under the control of multiple $\beta$-catenin responsive elements, a well-accepted system to assess activation of this pathway. LiCl administration produced high luciferase activity that was set at $100 \%$ in our assays (Figure 7). A combination of LiCl and rosiglitazone resulted in about $50 \%$ reduction of luciferase activity; the combination of LiCl with $(R) \mathbf{- 1}$ or compound $\mathbf{2}$ produced similar effects (about $50 \%$ reduction) while a slightly further decrease was observed with ( $S$ )-1 (about $40 \%$ reduction).


Figure 7. Top-FLASH assay in HEK293T cells. Cells were transfected with the Top-FLASH reporter plasmid and,
 Luciferase reporter gene activity was determined 48 hs after transfection by a specific assay. The results of the transfections are reported as luciferase fold induction after normalization to $\beta$-galactosidase activity used as control of transfection efficiency. Data shown are mean $\pm$ SD of three independent experiments performed in duplicate.

To further prove the specificity of these effects, we treated HT-29 cells with the various compounds and assessed the levels of c-Myc and cyclin D1, two $\beta$-catenin target proteins. We selected this cell line as it bears a mutation in the APC gene and thus the Wnt/ $\beta$-catenin pathway is already activated;[53] accordingly, c-Myc and cyclin D1 are already elevated in basal conditions. Treatment with rosiglitazone diminished the levels of both proteins; exposure to $(S) \mathbf{- 1},(R) \mathbf{- 1}$ and $(R, S) \mathbf{- 2}$ resulted in a similar reduction for cyclin D1, more evident for c-Myc (Figure 8). Interestingly, (S)-1 displayed a stronger effect on both proteins, in line with the results of the transactivation assay reported above.


Figure 8. $(S)-\mathbf{1},(R)-\mathbf{1}$ and $(R, S)-\mathbf{2}$ interfere with the $\mathrm{Wnt} / \beta$-catenin pathway. Total protein extracts from HT-29 cells untreated or treated with $10 \mu \mathrm{M}$ of each of the compounds $(S)-\mathbf{1},(R) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ or Rosiglitazone for 24 hs were analyzed by Western blot for c-Myc and cyclin D1 expression. $\beta$-actin was used as loading control. The bar graphs represent the mean $\pm \mathrm{SD}$ of proteins/ $\beta$-actin of at least three independent experiments. The results reported here are representative of a single Western blot experiment.

## 7. Effects of $(R)-1,(S)-1$ and $(R, S)-2$ on the carnitine shuttle system and mitochondrial functions.

Fatty acids oxidation takes place in mitochondria and is coupled to oxidative phosphorylation for energy production. In order to be degraded, fatty acids are transported from the cytosol by means of the carnitine shuttle system: once activated to acyl-CoAs, they are transferred to carnitine by the carnitine-palmitoyl-transferase 1 (CPT1), located on the surface of the outer mitochondrial membrane; $[54,55]$ the acylcarnitine crosses the outer membrane [56] and translocates through the inner mitochondrial membrane by virtue of the carnitine/acylcarnitine carrier (CAC). Therefore, the carnitine shuttle system is essential for the mitochondrial oxidation of fatty acids.[57] It has been demonstrated that two fibrates (Wy-14,643 and GW7647) stimulate CAC transcript and protein levels in primary and secondary cell lines as they act as PPAR $\alpha$ ligands, providing a molecular explanation of their hypolipidemic action.[36] Based on these premises, we evaluated whether our fibrate-like drugs could upregulate mitochondrial CAC and CPT1 gene expression by treating HepG2 cells with $1 \mu \mathrm{M}$ of $(R) \mathbf{- 1},(S)-\mathbf{1}$ or $(R, S) \mathbf{- 2}$ for $24 \mathrm{hs} ; \mathrm{Wy}-14,643(50 \mu \mathrm{M})$ was used as reference compound. Both CAC and CPT1 mRNA levels displayed a significant increase in treated HepG2 when compared to control cells by real-time PCR (Figure 9A,B). This was mirrored by a parallel increase of the corresponding proteins that was more evident for CPT1 (Figure 9C,D). These effects turned out to be similar or even greater than those observed for $\mathrm{Wy}-14,643$, whose concentration to activate both genes was fifty-fold higher ( 50 vs $1 \mu \mathrm{M}$ ).


Figure 9. Effect of $(R) \mathbf{- 1},(S) \mathbf{- 1}$ and $(R, S) \mathbf{- 2}$ on CAC and CPT1 gene expression. Total RNA extracted from HepG2 cells, untreated or treated with $1 \mu \mathrm{M}(R)-\mathbf{1}, 1 \mu \mathrm{M}(S)-\mathbf{1}$ or $1 \mu \mathrm{M}(R, S)-\mathbf{2}$ or $50 \mu \mathrm{M}$ Wy-14,643 for 24 hs, was used to quantify A) CAC mRNA and B) CPT1 mRNA by real-time PCR; data are the mean $\pm$ SD of three independent duplicate experiments. C) Total protein extracts from HepG2 cells, untreated or treated for 48 hs as described in panels and B), were immunoblotted for CAC, CPT1 and $\beta$-actin using specific antibodies. D) The intensities of immunolabeled protein bands as in panel C) were quantified by densitometric scanning and quantification is expressed as fold-change of CAC and CPT1 protein signals relative to control cells. In A) and B), data are means $\pm$ SD from three independent experiments $(* \mathrm{p}<0.05, * * \mathrm{p}<0.01$; one-way ANOVA, Bonferroni test). Western blot data in C) are representative of three independent experiments with similar results.

We also tested whether our compounds could activate PPAR $\alpha$ transcription in a positive regulatory loop, as a PPRE (peroxisome proliferator-activated receptor response element) is present in the human $\operatorname{PPAR} \alpha$ promoter.[58,59] To this end, HepG2 cells were incubated with each compound and Wy14,643 as control and PPAR $\alpha$ mRNA and protein levels assessed. All of our compounds, as well as $\mathrm{Wy}-14,643$, increased PPAR $\alpha$ mRNA and protein levels (Figure 10A,B). Also in this case, the Wy14,643 concentration was fifty-fold higher than $(R) \mathbf{- 1 ,}(S) \mathbf{- 1}$ and $(R, S) \mathbf{- 2}$.


Figure 10. PPAR $\alpha$ gene expression in $(R)-\mathbf{1},(S)-\mathbf{1}$ and $(R, S)-\mathbf{2}$ treated cells. A) Total RNA extracted from HepG2 cells untreated or treated with $1 \mu \mathrm{M}(R)-\mathbf{1}, 1 \mu \mathrm{M}(S)-\mathbf{1}, 1 \mu \mathrm{M}(R, S)-\mathbf{2}$ or $50 \mu \mathrm{M} \mathrm{Wy}-14,643$ for 24 hs , was used to quantify PPAR $\alpha$ mRNA by real-time PCR; data are the mean $\pm$ SD of three independent duplicate experiments. B) Total protein extracts from HepG2 cells, untreated or treated for 48 hs as in A), were immunoblotted for PPAR $\alpha$ and $\beta$-actin using specific antibodies. C) The intensities of immunolabeled protein bands as in panel B) were quantified by densitometric scanning and quantification is expressed as fold-change of PPAR $\alpha$ protein signals relative to control cells. In A), data are means $\pm$ SD from three independent experiments ( ${ }^{*} \mathrm{p}<0.05$; one-way ANOVA, Bonferroni test). Western blot data in B) are representative of three independent experiments with similar results.

It is known that mitochondrial dysfunction is linked to dyslipidemia and hypercholesterolemia [60] and impaired mitochondrial biogenesis is observed in these pathological conditions.[61] In addition, numerous studies in humans and animal models have shown that insulin resistance is frequently associated with reduced mitochondrial mass or oxidative function in insulin sensitive tissues, leading to the hypothesis that defective overall mitochondrial activity could play a relevant role in the etiology of insulin resistance and, hence, in type 2 diabetes. The possibility of increasing the mitochondrial mass and oxidative activity is, then, viewed as a potential therapeutic approach for the treatment of insulin resistance.[62] For this reason, we evaluated the effect of $(R) \mathbf{- 1},(S) \mathbf{- 1},(R, S) \mathbf{- 2}$, and Wy-14,643 on the mitochondrial mass in HepG2 cells transduced with CellLight®

Mitochondria-GFP BacMam 2.0, treated for twenty-four hours with each fibrate-like drug and analyzed under a fluorescent microscope or by a microplate reader. All compounds increased the mitochondrial mass by $30 \%$ as proven by a more intense fluorescent signal than control cells (Figure $11 \mathrm{~A}, \mathrm{~B})$. Also in this case, $\mathrm{Wy}-14,643$, the positive control, produced the same effect but at a fiftyfold higher concentration. Collectively, these findings show that all tested compounds significantly induce carnitine shuttle genes as direct targets of PPAR $\alpha$, the master regulator of mitochondrial fatty acid $\beta$-oxidation.


Figure 11. Changes in mitochondrial mass after $(R) \mathbf{- 1},(S) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ treatment. A) HepG 2 cells treated with CellLight® Mitochondria-GFP BacMam 2.0 in the presence or absence (control) of $1 \mu \mathrm{M}(R) \mathbf{- 1}, 1 \mu \mathrm{M}(S)-\mathbf{1}, 1 \mu \mathrm{M}(R, S)$ 2 or $50 \mu \mathrm{M} \mathrm{Wy}-14,643$ were observed (magnification 20x). B) HepG2 cells were treated as in A) and the fluorescence of cell extracts was measured by using a microplate reader at 475 nm for excitation and $500-550 \mathrm{~nm}$ for emission. Data are means $\pm$ SD from three independent experiments ( $* \mathrm{p}<0.05$; one-way ANOVA, Bonferroni test). Images showed in (A) are representative of three independent experiments with similar results.

PPAR $\alpha$ itself is upregulated by these compounds that exhibit also the ability to enhance mitochondrial biogenesis. Indeed, PPAR $\alpha$ activation has been shown to be sufficient to drive mitochondrial biogenesis in pathological conditions,[63] as it activates PGC-1 $\alpha$ and other transcription factors gene expression to coordinate mitochondrial biogenesis. These events, most probably, lead to an adaptive response that results in an increased mitochondrial function and fatty acid $\beta$-oxidation. Our results clearly demonstrate that $(R)-\mathbf{1},(S) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ can activate carnitine shuttle system through upregulation of CAC, CPT1 and PPAR $\alpha$ genes and induce a significant mitochondrial biogenesis that in turn sustains fatty acid $\beta$-oxidation.

## 8. Conclusion

In the present study, we investigated the properties of a short series of compounds characterized by the presence of the diphenylmethane skeleton that has already shown its usefulness as a multitemplate for nuclear receptors. We provide evidence that some of these compounds act as potent dual PPAR $\alpha / \gamma$ partial agonists as documented by docking experiments performed to understand the molecular basis of their partial agonism and the lack of stereoselectivity between the enantiomers of the most active compound $\mathbf{1}$. We also show that both enantiomers of $\mathbf{1}$ and racemic $\mathbf{2}$ have the ability to inhibit cellular proliferation in colon cancer cell lines due, at least in part, to a downregulation of the Wnt/ $\beta$-catenin signalling which is mutated and upregulated in the majority of colorectal cancers. Among the three compounds investigated, $(R) \mathbf{- 1},(S) \mathbf{- 1}$, and $(R, S) \mathbf{- 2}$, the stereoisomer $(S) \mathbf{- 1}$ seems to display a stronger antiproliferative effect and increased inhibition of the Wnt/ $\beta$-catenin signalling. We also demonstrate that $(R) \mathbf{- 1},(S) \mathbf{- 1}$ and $(R, S) \mathbf{- 2}$ are able to activate the carnitine shuttle system through upregulation of carnitine/acylcarnitine carrier (CAC), carnitine-palmitoyl-transferase 1 (CPT1) and PPAR $\alpha$ genes essential for the mitochondrial fatty acids $\beta$-oxidation. Moreover, these
compounds induce a significant mitochondrial biogenesis, an effect which is currently viewed as a potential therapeutic approach for the treatment of dyslipidemia, hypercholesterolemia and insulin resistance, pathological conditions frequently associated with reduced mitochondrial mass or oxidative function in insulin sensitive tissues. Collectively, $(R) \mathbf{- 1},(S) \mathbf{- 1}$ and $(R, S)-\mathbf{2}$ are dual PPAR $\alpha / \gamma$ ligands with promising biological effects in cellular systems and thus could stand as leads for the development of new molecules for the treatment of dyslipidemic type 2 diabetes and/or cancer. The lack of any activity toward FXR, though the high similarity with some recent phenoxyacetic acids reported in the literature as FXR agonists, indicates that this receptor is much more selective than PPARs and even small modifications in the chemical structure of an agonist can produce inactive molecules. However, even though only a few drugs are reported as dual PPAR/FXR agonists, the challenge to identify this type of ligands is still open and, with this aim, a fine-tuned search for new analogs of compounds $\mathbf{1}$ and $\mathbf{2}$ is now in progress.

## 9. Experimental section

### 9.1 Chemical Methods

Column chromatography was performed on ICN silica gel $60 \AA(63-200 \mu \mathrm{~m})$ as a stationary phase. Melting points were determined in open capillaries on a Gallenkamp electrothermal apparatus and are uncorrected. Mass spectra were recorded on an HP GC/MS 6890-5973 MSD spectrometer, electron impact 70 eV , equipped with HP chemstation or an Agilent LC/MS 1100 Series LC/MSD Trap System VL spectrometer, electrospray ionization (ESI). HR-MS analyses were carried out using an electrospray interface and a Q-TOF mass spectrometer (Agilent 6530 Accurate Mass Q-TOF LC/MS, Palo Alto, CA). Ionization was achieved in the negative ion mode. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra were recorded in $\mathrm{CDCl}_{3}$ (the use of other solvents is specified) on a Varian-Mercury 300 (300 $\mathrm{MHz})$ or an Agilent $500(500 \mathrm{MHz})$ spectrometer. Chemical shifts are expressed as parts per million (ppm, $\delta$ ). Microanalyses of solid compounds were carried out with an Eurovector Euro EA 3000 model analyzer and were within $\pm 0.4 \%$ of the theoretical values. Optical rotations were measured with a Perkin-Elmer 341 polarimeter at room temperature $\left(20^{\circ} \mathrm{C}\right)$ : concentrations are expressed as $\mathrm{g} / 100 \mathrm{~mL}$. The enantiomeric excesses of acids were determined by HPLC analysis on a Chiralcel OD column ( 4.6 mm i.d. x 250 mm , Daicel Chemical Industries, Ltd., Tokyo, Japan). Analytical liquid chromatography was performed on a PE chromatograph equipped with a Rheodyne 7725 i model injector, a 785A model UV/Vis detector, a series 200 model pump and NCI 900 model interface. Chemicals were from Sigma-Aldrich (Milan, Italy) and were used without any further purification.
9.2 Preparation of 4-(chloromethyl)-2-phenyl-1,3-oxazole (7) and 4-(chloromethyl)-2-(2,6-dichlorophenyl)-1,3-oxazole (8). General procedure

A mixture of the appropriate benzamide $(0.0083 \mathrm{~mol})$ and 1,3 -dichloroacetone $(0.0099 \mathrm{~mol})$ was warmed in a sealed vial at $180^{\circ} \mathrm{C}$ for 2 h . The resulting solution was cooled at room temperature, added with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 25 \mathrm{~mL})$. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and
filtered off. Then the solvent was evaporated in vacuo to give a brown oil which was chromatographed on a silica gel column ( $n$-hexane/ethyl acetate 95:5 as eluent) affording the desired compound as a white solid.

### 9.2.1. 4-(Chloromethyl)-2-phenyl-1,3-oxazole (7)

$51 \%$ Yield; $\mathrm{mp}=54-55^{\circ} \mathrm{C}$; GC/MS, $m / z(\%): 195\left(\mathrm{M}^{+}+2,16\right), 193\left(\mathrm{M}^{+}, 48\right), 158\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{NO}^{+}, 88\right)$, $103\left(\mathrm{C}_{7} \mathrm{H}_{5} \mathrm{~N}^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 4.58\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.44-7.47$ and 8.03-8.06 (m, 5 H , aromatics), 7.71 (s, 1H, CH).
9.2.2. 4-(Chloromethyl)-2-(2,6-dichlorophenyl)-1,3-oxazole (8)
$6 \%$ Yield; GC/MS, $m / z(\%): 265\left(\mathrm{M}^{+}+4,12\right), 263\left(\mathrm{M}^{+}+2,37\right), 261\left(\mathrm{M}^{+}, 38\right), 226\left(\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{Cl}_{2} \mathrm{NO}^{+}\right.$, 100); ${ }^{1} \mathrm{H}-\mathrm{NMR}: 4.63\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.38-7.42(\mathrm{~m}, 3 \mathrm{H}$, aromatics), $7.83(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH})$.
9.3 Preparation of 4-(chloromethyl)-5-methyl-2-phenyl-1,3-oxazole (9) and 4-(chloromethyl)-2-(2,6-dichlorophenyl)-5-methyl-2-phenyl-1,3-oxazole (10). General procedure A mixture of butan-2,3-dione $(5 \mathrm{~mL}), \mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}(3.97 \mathrm{~g} ; 0.058 \mathrm{~mol})$ and $\mathrm{CH}_{3} \mathrm{COONa}(9.52 \mathrm{~g}$; $0.116 \mathrm{~mol})$ was dissolved in $\mathrm{H}_{2} \mathrm{O} / \mathrm{EtOH} 96 \%(1: 5,250 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and stirred at room temperature for 4 h . Then ethanol was distilled off, and the residue was added with water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The solvent was evaporated in vacuo to give butane-2,3-dione oxime as a pale yellow solid ( $3.96 \mathrm{~g} ; 0.039 \mathrm{~mol}$ ): $68 \%$ yield;

GC/MS, $m / z(\%): 101\left(\mathrm{M}^{+}, 29\right), 58\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{NO}^{+}, 7\right), 43\left(\mathrm{C}_{2} \mathrm{H}_{3} \mathrm{O}^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 1.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\right.$ $\mathrm{C}=\mathrm{O}), 2.26\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{N}\right), 11.27(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$. The so obtained oxime $(1.40 \mathrm{~g} ; 0.014 \mathrm{~mol})$ and the appropriate benzaldheyde ( 0.015 mol ) were dissolved in a mixture of $4 \mathrm{M} \mathrm{HCl} /$ dioxane $(1: 1,20 \mathrm{~mL})$ and stirred at room temperature. After 4 h the solution was cooled to $0^{\circ} \mathrm{C}$ and the resulting precipitate was filtered off, suspended into $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{NH}_{4} \mathrm{OH}(\mathrm{pH}=8)$ and extracted with $\mathrm{CHCl}_{3}(3 \times 30 \mathrm{~mL})$. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The solvent was evaporated in
vacuo to give the corresponding N -oxides as pale yellow solids. 4,5-Dimethyl-2-phenyl-1,3-oxazole 3-oxide. $35 \%$ Yield; ${ }^{1} \mathrm{H}$-NMR: 2.23 and $2.37\left(2 \mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{CH}_{3}\right), 7.44-7.53$ and $8.43-8.47(\mathrm{~m}, 5 \mathrm{H}$, aromatics). 2-(2,6-Dichlorophenyl)-4,5-dimethyl-1,3-oxazole 3-oxide. $18 \%$ yield; ${ }^{1} \mathrm{H}-\mathrm{NMR}: 2.35$ and $2.48\left(2 \mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{CH}_{3}\right) ; 7.20-7.32\left(\mathrm{~m}, 3 \mathrm{H}\right.$, aromatics). $\mathrm{POCl}_{3}(0.802 \mathrm{~g} ; 0.052 \mathrm{mmol})$ in anhydrous $\mathrm{CHCl}_{3}(15 \mathrm{~mL})$ was added dropwise to a solution of the N -oxide ( 0.049 mol ) in anhydrous $\mathrm{CHCl}_{3}$ $(15 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The resulting mixture was refluxed for 8 h under nitrogen atmosphere, then cooled and treated with $\mathrm{H}_{2} \mathrm{O}$ and $\mathrm{NH}_{4} \mathrm{OH}(\mathrm{pH}=8)$. The two layers were separated and the aqueous phase was extracted with $\mathrm{CHCl}_{3}(2 \times 30 \mathrm{~mL})$. The organic layers were collected, washed with brine ( $2 \times 50 \mathrm{~mL}$ ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered off. The solvent was evaporated in vacuo to give a brown solid residue which was chromatographed on a silica gel column ( $n$-hexane/ethyl acetate $80: 20$ as eluent) affording the desired compounds as pale yellow solids.

### 9.3.1. 4-(Chloromethyl)-5-methyl-2-phenyl-1,3-oxazole (9)

$53 \%$ Yield; $\mathrm{mp}=83-84^{\circ} \mathrm{C}$; GC/MS, $m / z(\%): 209\left(\mathrm{M}^{+}+2,12\right), 207\left(\mathrm{M}^{+}, 38\right), 172\left(\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{NO}^{+}, 100\right)$, $77\left(\mathrm{C}_{6} \mathrm{H}_{5}{ }^{+}, 10\right)$; ${ }^{1} \mathrm{H}-\mathrm{NMR}: 2.43\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.56\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.42-7.46$ and 7.98-8.02 (m, 5 H , aromatics).
9.3.2. 4-(Chloromethyl)-2-(2,6-dichlorophenyl)-5-methyl-2-phenyl-1,3-oxazole (10)
$18 \%$ Yield; GC/MS, $m / z(\%): 279\left(\mathrm{M}^{+}+4,9\right), 277\left(\mathrm{M}^{+}+2,27\right), 275\left(\mathrm{M}^{+}, 27\right), 240\left(\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{NO}^{+}\right.$, 100); ${ }^{1} \mathrm{H}-\mathrm{NMR}: 2.45\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.59\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.35-7.42$ (m, 3 H , aromatics).

### 9.4. Preparation of 4-[1-ethyl-1-(4-hydroxyphenyl)propyl]phenol (11)

Phenol ( 0.106 mol ), 3-pentanone ( 3.5 mL ) and methansulfonic acid ( 2.6 mL ) were stirred at room temperature for 5 days. Then, the mixture was dissolved in EtOAc and washed ( $3 \times 30 \mathrm{~mL}$ ) with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to dryness affording a white solid which was crystallized from $\mathrm{CHCl}_{3} / n$-hexane. $34 \%$ Yield; mp: $85-86^{\circ} \mathrm{C}$; GC/MS, $m / z(\%): 256\left(\mathrm{M}^{+}, 5\right), 227\left(\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{O}_{2}{ }^{+}\right.$,
100), $212\left(\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{2}{ }^{+}, 3\right), 133\left(\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{O}^{+}, 14\right) ;{ }^{1} \mathrm{H}$ NMR: $0.60\left(\mathrm{t}, \mathrm{J}=7.42,6 \mathrm{H}, 2 \mathrm{CH}_{3}\right), 2.01(\mathrm{q}, \mathrm{J}=7.42$, $4 \mathrm{H}, 2 \mathrm{CH}_{2}$ ), $4.60\left(\mathrm{bs}, 2 \mathrm{H}, 2 \mathrm{OH}, \mathrm{D}_{2} \mathrm{O}\right.$ exchanged), 6.70-6.74 and 7.00-7.04 ( $\mathrm{m}, 8 \mathrm{H}$, aromatics).
9.5. Preparation of 2-[4-(4-hydroxybenzyl)phenoxy]propionic acid methyl ester (12) and 2-\{4-[1-ethyl-1-(4-hydroxyphenyl)propyl]phenoxy]propionic acid ethyl ester (13). General procedure A suspension of $\mathrm{NaH}(0.009 \mathrm{~mol})$ in anhydrous THF $(20 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ and added with a solution of commercially available bis(4-hydroxyphenyl)methane or $\mathbf{1 1}(0.009 \mathrm{~mol})$ in anhydrous THF ( 15 mL ). After 30 minutes, a solution of methyl or ethyl 2-bromopropionate ( 0.009 mol ) in anhydrous THF ( 15 mL ) was added dropwise and the mixture warmed at $60^{\circ} \mathrm{C}$ for 96 h . The solvent was evaporated and the residue was added with ice and extracted with ethyl acetate. The organic layer was washed with 2 N HCl and twice with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. The solid residue was chromatographed on a silica gel column ( $n$-hexane/ethyl acetate 85:15 as eluent) affording the desired compounds as colorless oils.

### 9.5.1. 2-[4-(4-Hydroxybenzyl)phenoxy]propionic acid methyl ester (12)

$36 \%$ Yield; GC/MS, $m / z(\%): 286\left(\mathrm{M}^{+}, 100\right), 227\left(\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{O}_{2}{ }^{+}, 64\right), 199\left(\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{O}_{2}{ }^{+}, 62\right), 107\left(\mathrm{C}_{7} \mathrm{H}_{7} \mathrm{O}^{+}\right.$, 54); ${ }^{1} \mathrm{H}-\mathrm{NMR}: 1.60\left(\mathrm{~d}, \mathrm{~J}=6.88,3 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 3.75\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COOCH}_{3}\right), 3.83\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.72(\mathrm{q}+$ bs, $\mathrm{J}=6.88,2 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}+\mathrm{OH}, \mathrm{D}_{2} \mathrm{O}$ exchanged), 6.73-6.80 (m, 4H, aromatics), 7.00-7.07 (m, 4H, aromatics).
9.5.2. 2-\{4-[1-Ethyl-1-(4-hydroxyphenyl)propyl]phenoxy\}propionic acid ethyl ester (13)
$18 \%$ Yield; GC/MS, $m / z(\%): 356\left(\mathrm{M}^{+}, 4\right), 327\left(\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{O}_{4}{ }^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 0.58(\mathrm{t}, \mathrm{J}=7.42,6 \mathrm{H}, 2$ $\left.\mathrm{CH}_{3}-\mathrm{CH}_{2}\right), 1.23\left(\mathrm{t}, \mathrm{J}=7.15,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{O}\right), 1.58\left(\mathrm{~d}, \mathrm{~J}=6.88,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}\right), 2.01(\mathrm{q}, \mathrm{J}=7.42,4 \mathrm{H}, 2$ $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.17-4.26\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{OCH}_{2}-\mathrm{CH}_{3}\right), 4.72\left(\mathrm{q}+\mathrm{bs}, \mathrm{J}=6.88,2 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}+\mathrm{OH}, \mathrm{D}_{2} \mathrm{O}\right.$ exchanged $)$ 6.68-6.77 and 6.98-7.02 (m, 8H, aromatics).

### 9.6. Preparation of the methyl esters ( $R$ )-12 and (S)-12. General procedure

To an ice-bath cooled mixture of $(R)$-or $(S)$-methyl lactate ( 4.8 mmol ), bis(4-hydroxyphenyl)methane ( 4.8 mmol ), and triphenylphosphine ( 4.8 mmol ) in anhydrous THF ( 30 mL ), a solution of diisopropyl azodicarboxylate (DIAD, 4.8 mmol ) in anhydrous THF ( 20 mL ) was added dropwise under nitrogen atmosphere. The mixture was allowed to warm to room temperature and stirred for overnight. The solvent was evaporated affording an oil which was chromatographed on a silica gel column ( $n$ hexane/ethyl acetate $9: 1$ as eluent) to give the target methyl esters as white solids.
9.6.1. (R)-2-[4-(4-Hydroxybenzyl)phenoxy]propionic acid methyl ester, ( $R$ )-12. $37 \%$ Yield.
9.6.2. (S)-2-[4-(4-Hydroxybenzyl)phenoxy]propionic acid methyl ester, (S)-12. 30\% Yield.

### 9.7. Preparation of the esters 14-19. General Procedure

$\mathrm{Cs}_{2} \mathrm{CO}_{3}(3 \mathrm{mmol})$ was added to a stirred solution of $\mathbf{1 2}$ or $\mathbf{1 3}(1 \mathrm{mmol})$ in acetonitrile $(10 \mathrm{~mL})$. After 0.5 h , a solution of the suitable 4-chloromethyl-1,3-oxazole 7-10 (1.5 mmol) in acetonitrile ( 10 mL ) was added dropwise and the resulting mixture was refluxed for 4 h . Then, acetonitrile was evaporated and the residue was taken up with ethyl acetate. The organic phase was washed with 0.5 N NaOH and brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to dryness affording yellows oils, which were chromatographed on a silica gel column ( $n$-hexane/ethyl acetate 95:5 as eluent) affording the desired compounds as colourless oils.
9.7.1. 2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy\}propionic acid methyl ester (14) $90 \%$ Yield; GC/MS, $m / z(\%): 443\left(\mathrm{M}^{+}, 18\right), 158\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{NO}^{+}, 100\right), 103\left(\mathrm{C}_{7} \mathrm{H}_{5} \mathrm{~N}^{+}, 49\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO- $d_{6}$ ): 1.45 (d, J=6.88, $3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}$ ), 3.64 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{O}$ ), 3.78 (s, $2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{Ph}$ ), 4.88 ( q , $\left.\mathrm{J}=6.88,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 4.99\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.77-7.12(\mathrm{~m}, 8 \mathrm{H}$, aromatics), 7.51-7.97 (m, 5 H , aromatics), 8.27 (s, 1H, CH-C=C-N).
9.7.2. 2-\{4-[4-(5-Methyl-2-phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy]propionic acid methyl ester (15)
$67 \%$ Yield; GC/MS, $m / z(\%): 457\left(\mathrm{M}^{+}, 4\right), 172\left(\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{NO}^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 1.45(\mathrm{~d}$, $\left.\mathrm{J}=6.87,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}\right), 2.41\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\mathrm{N}\right), 3.63\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{O}\right), 3.78\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{Ph}\right) ; 4.86$ (q, $\left.\mathrm{J}=6.87,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right) ; 4.93\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right) ; 6.74-7.93$ (m, 13H aromatics).
9.7.3. 2-\{4-\{1-Ethyl-1-[4-(2-phenyl-1,3-oxazol-4-ylmethoxy)phenyl]propyl\}phenoxy\}propionic acid ethyl ester (16)
$51 \%$ Yield; GC/MS, $m / z(\%): 513\left(\mathrm{M}^{+}, 2\right), 484\left(\mathrm{C}_{32} \mathrm{H}_{35} \mathrm{NO}_{5}{ }^{+}, 93\right), 158\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{NO}^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO- $d_{6}$ ): $0.59\left(\mathrm{t}, \mathrm{J}=7.16,6 \mathrm{H}, 2 \mathrm{CH}_{3}-\mathrm{CH}_{2}\right), 0.85\left(\mathrm{t}, \mathrm{J}=7.15,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{O}\right), 1.23(\mathrm{~d}, \mathrm{~J}=6.88,3 \mathrm{H}$, $\left.\mathrm{CH}_{3}-\mathrm{CH}\right) ; 2.02\left(\mathrm{q}, \mathrm{J}=7.16,4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 4.20\left(\mathrm{q}, \mathrm{J}=7.15,2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 4.68(\mathrm{q}, \mathrm{J}=6.88,1 \mathrm{H}$, $\mathrm{CH}-\mathrm{CH}_{3}$ ), $5.09\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right) ; 6.69-7.09,7.45-7.49$ and 7.98-8.13 (m, 13 H , aromatics), $7.74(\mathrm{~s}, 1 \mathrm{H}$, $\mathrm{O}-\mathrm{CH}=\mathrm{C})$.
9.7.4. 2-\{4-\{1-Ethyl-1-[4-(5-methyl-2-phenyl-1,3-oxazol-4ylmethoxy)phenyl]propyl\}phenoxy\}propionic acid ethyl ester (17) $88 \%$ Yield; GC/MS, $m / z(\%): 527\left(\mathrm{C}_{33} \mathrm{H}_{37} \mathrm{NO}_{5}{ }^{+}, 2\right), 498\left(\mathrm{C}_{31} \mathrm{H}_{32} \mathrm{NO}_{5}{ }^{+}, 13\right), 327\left(\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{O}_{4}{ }^{+}, 35\right)$, $172\left(\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{NO}^{+}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 0.59\left(\mathrm{t}, \mathrm{J}=7.42,6 \mathrm{H}, 2 \mathrm{CH}_{3}-\mathrm{CH}_{2}\right), 1.23\left(\mathrm{t}, \mathrm{J}=7.15,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}_{2}{ }^{-}\right.$ O), $1.59\left(\mathrm{~d}, \mathrm{~J}=6.60,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}\right), 2.04\left(\mathrm{q}, \mathrm{J}=7.42,4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 2.46\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\mathrm{N}\right)$, $4.19\left(\mathrm{q}, \mathrm{J}=7.15,2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 4.70\left(\mathrm{q}, \mathrm{J}=6.60,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 5.04\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.74-7.49$ (m, 13H, aromatics).
9.7.5. 2-\{4-\{1-\{4-[2-(2,6-Dichloro-phenyl)-1,3-oxazol-4-ylmethoxy]phenyl\}-1-ethyl-propyl\}phenoxy]propionic acid ethyl ester (18)
$52 \%$ Yield; ESI-MS, $m / z(\%): 608\left(\mathrm{M}^{+}+27,13\right), 606\left(\mathrm{M}^{+}+25,66\right), 604\left(\mathrm{M}^{+}+23,100\right), 551$ $\left(\mathrm{C}_{30} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{NO}_{3}{ }^{+}, 13\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 0.59\left(\mathrm{t}, \mathrm{J}=7.42,6 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.23\left(\mathrm{t}, \mathrm{J}=7.15,3 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$,
$1.57\left(\mathrm{~d}, \mathrm{~J}=6.88,3 \mathrm{H}, \mathrm{CHCH}_{3}\right), 2.02\left(\mathrm{q}, \mathrm{J}=7.42,4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.20\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.70(\mathrm{q}$, $\left.\mathrm{J}=6.88,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 5.11\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{O}\right), 6.73-7.42(\mathrm{~m}, 11 \mathrm{H}$, aromatics $), 7.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{O}-\mathrm{CH}=\mathrm{C})$.
9.7.6. 2-\{4-\{1-\{4-[2-(2,6-Dichloro-phenyl)-5-methyl-1,3-oxazol-4-ylmethoxy]phenyl\}-1-ethyl-propylf-phenoxylpropionic acid ethyl ester (19) $48 \%$ Yield; ESI-MS, $m / z(\%): 622\left(\mathrm{M}^{+}+27,14\right), 620\left(\mathrm{M}^{+}+25,70\right), 618\left(\mathrm{M}^{+}+23,100\right), 551$ $\left(\mathrm{C}_{30} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{NO}_{3}{ }^{+}, 17\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 0.58\left(\mathrm{t}, \mathrm{J}=7.43,6 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.24\left(\mathrm{t}, \mathrm{J}=7.15,3 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, 1.58 (d, J=6.88, 3H, CHCH3), 2.04 (q, J=7.43, 4H, $2 \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.43 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\mathrm{N}$ ), 4.21 (m, $\left.2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.69\left(\mathrm{q}, \mathrm{J}=6.88,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 5.02\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{O}\right), 6.73-7.25(\mathrm{~m}, 11 \mathrm{H}$, aromatics).

### 9.8. Preparation of the methyl esters ( $R$ )-14 and ( $S$ )-14

For the preparation of these isomers the same procedure to obtain the racemate $\mathbf{1 4}$ was followed starting from $(R) \mathbf{- 1 2}$ or $(S) \mathbf{- 1 2}$, respectively, and 1,3-oxazole 7.
9.8.1. (R)-2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy]propionic acid methyl ester, (R)-14. 83\% Yield.
9.8.2. (S)-2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy\}propionic acid methyl ester, (S)-14. 74\% Yield.

### 9.9. Preparation of the final acids 1-6. General Procedure

A solution of the corresponding ethyl or methyl ester ( 5 mmol ) in THF ( 30 mL ) and $1 \mathrm{~N} \mathrm{NaOH} \mathrm{(30}$ mL ) was stirred overnight at room temperature. The organic solvent was removed under reduced pressure and the aqueous phase was acidified with 2 N HCl and extracted with ethyl acetate. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to dryness affording the final acids as white solids.
9.9.1. 2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy]propionic acid (1)
$54 \%$ Yield; $\mathrm{mp}=185-186{ }^{\circ} \mathrm{C}$ (methanol/ $\mathrm{CHCl}_{3}$ ); IR ( KBr ) cm ${ }^{-1}: 3200-2800(\mathrm{OH}), 1731(\mathrm{C}=\mathrm{O})$; ESI-MS, $m / z(\%): 428(\mathrm{M}-1,92), 356\left(\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{NO}_{3}{ }^{-}, 100\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 1.44$ (d, J=6.87, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 3.77 ( s, 2H, CH2-Ph), 4.72 (q, J=6.87, 1H, CH-CH $), 4.99\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.73-6.76$, 6.93-6.96, 7.09-7.12, 7.50-7.53, 7.94-7.99 (m, 14H, 13 aromatics $+\mathrm{COOH}, \mathrm{D}_{2} \mathrm{O}$ exchanged), 8.26 $(\mathrm{s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{C}=\mathrm{C}-\mathrm{N}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta=173.7,161.3,156.7,156.1,138.6,138.2,134.5$, 131.2, 130.0, 129.9, 129.6, 127.1, 126.4, 115.1, 72.0, 62.0, 18.8; HR-MS: $\left(\left[\mathrm{C}_{26} \mathrm{H}_{23} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}, \mathrm{m} / \mathrm{z}$ 428.1493 (calc. 428.1503 ). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{23} \mathrm{NO}_{5}: \mathrm{C}, 72.71 \%$; H, $5.40 \%$; N, 3.26\%. Found: C, $73.56 \%$; H, 5.84\%; N, 3.06\%.
9.9.2. (R)-2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy\}propionic acid, ((R)-1) $40 \%$ Yield; $\mathrm{mp}=202-204{ }^{\circ} \mathrm{C}$ (methanol). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{23} \mathrm{NO}_{5}: \mathrm{C}, 72.71 \% ; \mathrm{H}, 5.40 \% ; \mathrm{N}$, $3.26 \%$. Found: C, $72.56 \%$; H, $5.34 \%$; N, $3.44 \% .[\alpha]^{20}{ }_{D}=+34.7(c=1.0$, DMF). HPLC (Chiralcel OD, flux $1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}, n$-hexane/isopropanol/trifluoroacetic acid $80 / 20 / 0.2$ ): e.e. $\geq 98 \%$. 9.9.3. (S)-2-\{4-[4-(2-Phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy]propionic acid, ((S)-1) $56 \%$ Yield; $\mathrm{mp}=203-205{ }^{\circ} \mathrm{C}$ (methanol). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{23} \mathrm{NO}_{5}: \mathrm{C}, 72.71 \% ; \mathrm{H}, 5.40 \% ; \mathrm{N}, 3.26 \%$. Found: C, $70.94 \%$; H, $5.30 \%$; N, $3.39 \% .[\alpha]^{20}{ }_{D}=-34.4(\mathrm{c}=1.0$, DMF). HPLC (Chiralcel OD, flux 1 $\mathrm{mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}, n$-hexane/isopropanol/trifluoroacetic acid 80/20/0.2): e.e. $\geq 99 \%$.
9.9.4. 2-\{4-[4-(5-Methyl-2-phenyl-1,3-oxazol-4-ylmethoxy)benzyl]phenoxy]propionic acid (2) $86 \%$ Yield; $\mathrm{mp}=167-168{ }^{\circ} \mathrm{C}\left(\mathrm{CHCl}_{3} / n\right.$-hexane $)$; IR $(\mathrm{KBr}) \mathrm{cm}^{-1}: 3200-2800(\mathrm{OH}), 1731(\mathrm{C}=\mathrm{O})$; ESI-MS, $m / z(\%): 466(\mathrm{M}+23,100), 444(\mathrm{M}+1,19) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 1.43(\mathrm{~d}, \mathrm{~J}=6.60,3 \mathrm{H}$, $\mathrm{CH}_{3}-\mathrm{CH}$ ), 2.41 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\mathrm{N}$ ), 3.77 ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{Ph}$ ), $4.70\left(\mathrm{q}, \mathrm{J}=6.60,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 4.92(\mathrm{~s}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.72-6.75,6.91-6.94,7.06-7.12,7.48-7.54$ and $7.90-7.93(\mathrm{~m}, 14 \mathrm{H}, 13$ aromatics + $\mathrm{COOH}, \mathrm{D}_{2} \mathrm{O}$ exchanged). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta=173.6,159.2,156.9,156.1,147.9,134.6$, 134.4, 132.5, 130.8 130.0, 129.9, 129.6, 127.3, 126.1, 115.2, 115.1, 71.9, 61.8, 18.7, 10.4; HR-MS: ( $\left.\left[\mathrm{C}_{27} \mathrm{H}_{25} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}, \mathrm{m} / \mathrm{z} 442.1652$ (calc. 442.1660 ). Anal. Calcd for $\mathrm{C}_{27} \mathrm{H}_{25} \mathrm{NO}_{5}: \mathrm{C}, 73.12 \%$; H , $5.68 \%$; N, $3.16 \%$. Found: C, $73.57 \%$; H, $5.59 \%$; N, $2.91 \%$.
9.9.5. 2-\{4-\{1-Ethyl-1-[4-(2-phenyl-1,3-oxazol-4-ylmethoxy)phenyl]propyl\}phenoxy]propionic acid (3)
$49 \%$ Yield; $\mathrm{mp}=140-141^{\circ} \mathrm{C}\left(\mathrm{CHCl}_{3} / n\right.$-hexane $)$; IR $(\mathrm{KBr}) \mathrm{cm}^{-1}: 3200-2800(\mathrm{OH}), 1730(\mathrm{C}=\mathrm{O})$; ESIMS, $m / z(\%): 486(\mathrm{M}+1,68), 508(\mathrm{M}+23,100), 484(\mathrm{M}-1,35) ;{ }^{1} \mathrm{H}-\mathrm{NMR}: 0.52\left(\mathrm{t}, \mathrm{J}=6.96,6 \mathrm{H}, 2 \mathrm{CH}_{3^{-}}\right.$ $\left.\mathrm{CH}_{2}\right), 1.44\left(\mathrm{~d}, \mathrm{~J}=6.59,3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{CH}\right), 1.96\left(\mathrm{q}, \mathrm{J}=6.96,4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 4.70(\mathrm{q}, \mathrm{J}=6.59,1 \mathrm{H}, \mathrm{CH}-$ $\left.\mathrm{CH}_{3}\right), 4.99\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.70-6.73,6.90-7.05,7.51-7.54,7.95-7.99(\mathrm{~m}, 14 \mathrm{H}, 13$ aromatics + $\mathrm{COOH}, \mathrm{D}_{2} \mathrm{O}$ exchanged), 8.27 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{O}-\mathrm{CH}=\mathrm{C}$ ). ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO- $d_{6}$ ): $\delta=173.7,161.3,156.1$, $155.5,141.2,138.6,138.3,131.2,129.6,129.1,129.0,127.1,126.4,114.3,71.9,62.0,48.4,28.9$, 18.8, 8.7; HR-MS: ([ $\left.\left.\mathrm{C}_{30} \mathrm{H}_{31} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}$, m/z 484.2119 (calc. 484.2129). Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{31} \mathrm{NO}_{5}$ : C, $74.21 \%$; H, $6.43 \%$; N, 2.88\%. Found: C, $73.77 \%$; H, 6.89\%; N, 3.12\%.
9.9.6. 2-\{4-\{1-Ethyl-1-[4-(5-methyl-2-phenyl-1,3-oxazol-4-ylmethoxy)phenyl]propyl\}phenoxy\}propionic acid (4)
$38 \%$ Yield; $\mathrm{mp}=65-66{ }^{\circ} \mathrm{C}\left(\mathrm{CHCl}_{3} / n\right.$-hexane $) ;$ IR $(\mathrm{KBr}) \mathrm{cm}^{-1}: 3200-2800(\mathrm{OH}), 1730(\mathrm{C}=\mathrm{O}) ;$ ESIMS, $m / z(\%): 498(\mathrm{M}-1,100), 522(\mathrm{M}+23,100), 425\left(\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{NO}_{3}{ }^{-}, 19\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 0.53$ (t, J=7.15, 6H, $2 \mathrm{CH}_{3}-\mathrm{CH}_{2}$ ), 1.45 (d, J=6.87, 3H, $\mathrm{CH}_{3}$ - CH ), $1.99\left(\mathrm{q}, \mathrm{J}=7.15,4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 2.42$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\mathrm{N}$ ), $4.72\left(\mathrm{q}, \mathrm{J}=6.87,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 4.93\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{O}\right), 6.71-7.05,7.43-7.54$ and 7.83-7.94 (m, $14 \mathrm{H}, 13$ aromatics $+\mathrm{COOH}, \mathrm{D}_{2} \mathrm{O}$ exchanged). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta=173.7,159.3$, $156.3,155.5,147.8,141.2,141.1,134.7,132.5,131.6,130.8,129.7,129.5,129.0,128.6,127.9$, 126.0, 114.3, 71.9, 61.7, 48.4, 28.9, 18.8, 10.4, 8.7; HR-MS: $\left(\left[\mathrm{C}_{31} \mathrm{H}_{33} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}, \mathrm{m} / \mathrm{z} 498.2277$ (calc. 498.2286). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{33} \mathrm{NO}_{5} \times 3 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 67.25 \%$; H, $7.10 \%$; N, $2.53 \%$. Found: C, 68.27\%; H, 6.89\%; N, 2.60\%.
9.9.7. 2-\{4-\{1-\{4-[2-(2,6-Dichloro-phenyl)-1,3-oxazol-4-ylmethoxy]phenyl\}-1-ethyl-propyl\}-phenoxy\}-propionic acid (5)
$38 \%$ Yield; $\mathrm{mp}=79-80^{\circ} \mathrm{C}$ ( $n$-hexane); IR (KBr) $\mathrm{cm}^{-1}: 3200-2800(\mathrm{OH}), 1730(\mathrm{C}=\mathrm{O}) ;$ ESI-MS, $m / z$ (\%): $556(\mathrm{M}+4-1,16), 554(\mathrm{M}+2-1,78), 552(\mathrm{M}-1,100) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 0.51(\mathrm{t}, \mathrm{J}=6.876 \mathrm{H}$, $\left.2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.44\left(\mathrm{~d}, \mathrm{~J}=6.60,3 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 1.97\left(\mathrm{q}, \mathrm{J}=6.87,4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.71(\mathrm{q}, \mathrm{J}=6.60$,
$\left.1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 5.03\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{O}\right), 6.71-7.05$ and $7.60-7.69\left(\mathrm{~m}, 12 \mathrm{H}, 11\right.$ aromatics $+\mathrm{COOH}, \mathrm{D}_{2} \mathrm{O}$ exchanged), 8.41 (s, 1H, O-CH-C=C). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta=173.7,156.1,155.5,141.3,141.2$, $139.8,138.9,137.8,135.6,133.9,129.1,129.0,128.9,127.3,114.9,114.5,114.3,71.9,61.9,48.5$, 29.0, 18.8, 8.7; HR-MS: $\left(\left[\mathrm{C}_{30} \mathrm{H}_{29} \mathrm{Cl}_{2} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}$, m/z 552.1338 (calc. 552.1350). Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{29} \mathrm{Cl}_{2} \mathrm{NO}_{5}$ : C, $64.99 \%$; H, $5.27 \%$; N, $2.53 \%$. Found: C, $66.56 \% ; \mathrm{H}, 5.67 \% ; \mathrm{N}, 2.16 \%$.
9.9.8. 2-\{4-\{1-\{4-[2-(2,6-Dichloro-phenyl)-5-methyl-1,3-oxazol-4-ylmethoxy]phenyl\}-1-ethyl-propylf-phenoxy\}-propionic acid (6)
$57 \%$ Yield; $\mathrm{mp}=83-85^{\circ} \mathrm{C}$ ( $n$-hexane). IR (KBr) $\mathrm{cm}^{-1}: 3200-2800(\mathrm{OH}), 1730(\mathrm{C}=\mathrm{O}) ;$ ESI-MS, $m / z$ (\%): $570(\mathrm{M}+4-1,15), 568(\mathrm{M}+2-1,72), 566(\mathrm{M}-1,100) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): 0.52(\mathrm{t}, \mathrm{J}=6.88,6 \mathrm{H}$, $\left.2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.44\left(\mathrm{~d}, \mathrm{~J}=6.60,3 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 1.98\left(\mathrm{q}, \mathrm{J}=6.88,4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.40\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{C}=\mathrm{C}-\right.$ $\mathrm{N}), 4.71\left(\mathrm{q}, \mathrm{J}=6.60,1 \mathrm{H}, \mathrm{CH}-\mathrm{CH}_{3}\right), 4.97\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{O}\right), 6.70-7.67(\mathrm{~m}, 12 \mathrm{H}, 11$ aromatics +COOH , $\mathrm{D}_{2} \mathrm{O}$ exchanged). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): ~ \delta=173.7,156.2,155.5,153.9,149.0,141.3,141.2,135.6$, 133.8, 132.1, 129.1, 129.0, 127.5, 114.7, 114.3, 71.9, 61.8, 48.5, 29.0, 18.8, 10.3, 8.7; HR-MS: $\left(\left[\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{Cl}_{2} \mathrm{NO}_{5}\right]-\mathrm{H}\right)^{-}, \mathrm{m} / \mathrm{z} 566.1495$ (calc. 566.1507). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{Cl}_{2} \mathrm{NO}_{5}$ : C, $65.50 \% ; \mathrm{H}$, $5.50 \%$; N, $2.46 \%$. Found: C, $66.34 \%$; H, $5.72 \%$; N, $2.43 \%$.

### 9.10. Surface Plasmon Resonance

SPR analyses were carried out on a BIACORE 3000 instrument (GE-Healthcare) according to our previously published procedures.[39] Briefly, PPAR $\alpha, \operatorname{PPAR} \gamma$, and PPAR $\delta$ surfaces were immobilized on a research-grade CM5 sensor chip (GE Healthcare) using a standard amine-coupling protocol. A densities of $3-5 \mathrm{kRU}$ ( 1000 RU corresponds to the binding of $\sim 1 \mathrm{ng}$ per square mm of protein on the dextran surface) was achieved. Each compound was injected on the protein chips at five different concentrations ranging from 0.1 to 50 nM . Measurements were performed at $25^{\circ} \mathrm{C}$, using a $50 \mu \mathrm{~L} \mathrm{~min}{ }^{-1}$ flow rate. Association and dissociation times were set at 60 s and 300 s , respectively. Interactions curves were fit to a single-site bimolecular interaction model to yield $K_{D}$. BIAevaluation software (GE Healthcare) was used for sensorgrams elaboration.

### 9.11. Biological Methods

Reference compounds, the medium, and other cell culture reagents were purchased from SigmaAldrich (Milan, Italy) and Invitrogen (Basel, Switzerland).

### 9.12. Plasmids

The expression vectors expressing the chimeric receptors containing the yeast GAL4-DNA binding domain fused to either the human PPAR $\alpha, \operatorname{PPAR} \gamma, \operatorname{PPAR} \delta$ or FXR ligand binding domain (LBD) and the reporter plasmid for these GAL4 chimeric receptors (pGAL5TKpGL3) containing five repeats of the GAL4 response elements upstream of a minimal thymidine kinase promoter upstream to the luciferase gene were described previously.[64]

### 9.13. Cell Culture and Transfections

Human hepatoblastoma cell line HepG2 (Interlab Cell Line Collection, Genoa, Italy) was cultured in Minimum Essential Medium (MEM) containing 10\% heat-inactivated foetal bovine serum, 100 U of penicillin $\mathrm{G} \cdot \mathrm{mL}^{-1}$ and $100 \mu \mathrm{~g}$ of streptomycin sulfate $\cdot \mathrm{mL}^{-1}$ at $37^{\circ} \mathrm{C}$ in a humidified atmosphere of 5\% $\mathrm{CO}_{2}$. For transactivation assays, $10^{5}$ cells/well were seeded in a 24 -well plate and transfections were performed after 24 hs with the $\mathrm{CAPHOS}^{\circledR}$, a calcium-phosphate method, according to the manufacturer's guidelines. Cells were transfected with expression plasmids encoding the fusion protein GAL4-PPAR $\alpha-L B D, ~ G A L 4-P P A R ~ \gamma-L B D, ~ G A L 4-P P A R \delta-L B D ~ o r ~ G A L 4-F X R-L B D ~(30 ~$ ng), pGAL5TKpGL3 (100 ng), pCMVßgal ( 250 ng ). Four hours after transfection, cells were treated for 20 hs with the indicated ligands in triplicate. Luciferase activity in cell extracts was then determined by a luminometer (VICTOR ${ }^{3}$ V Multilabel Plate Reader PerkinElmer). $\beta$ -

Galactosidase activity was determined using ortho-nitro-phenyl- $\beta$-D-galactopyranoside as described previously.[65] All transfection experiments were repeated at least twice.

### 9.14. Real-time PCR

HepG2 cells in serum-free medium were incubated for 24 hs with $1 \mu \mathrm{M}(R)-\mathbf{1},(S) \mathbf{- 1},(R, S) \mathbf{- 2}$ or 50 $\mu \mathrm{M}$ Wy-14,643. Total RNA was extracted from $1 \times 10^{6}$ cells; reverse transcription and real-time PCR were performed as previously reported.[66] Specific TaqMan Gene Expression Assays for human CPT1, CAC, PPAR $\alpha$ and $\beta$-actin (Hs00912671m1, Hs01088810g1, Hs00947536_m1 and Hs00357333g1, respectively) were purchased from Life technologies. All transcript levels were normalized against the $\beta$-actin expression levels.
9.15. Cell culture, treatment with compounds $(R)-1,(S)-1$ or $(R, S)-2$ and viability evaluation by counting

The human colon adenocarcinoma derived cells HT-29 and human kidney HEK293T cells were grown as a monolayer in D-MEM containing 10\% FBS, 1\% penicillin-streptomycin and 1\% Lglutamine and cultured in 100 mm plates at $37^{\circ} \mathrm{C}$ in a $5 \% \mathrm{CO}_{2}$ humidified atmosphere up to $70-80 \%$ confluence. $(R) \mathbf{- 1},(S)-\mathbf{1}$ or $(R, S)-\mathbf{2}$ or rosiglitazone were dissolved in DMSO as vehicle and treatments were carried out in the presence of $10 \%$ FBS containing $1 \%$ penicillin-streptomycin and $1 \%$ Lglutamine. In all experiments, the DMSO final concentration in the medium was less than $0.1 \%$. To evaluate cell growth after exposure to test compounds, cells were plated in 12 -well plates at $10^{6}$ cells $/ \mathrm{cm}^{2}$ density (24-well plates at $10^{5}$ cells $/ \mathrm{cm}^{2}$ density). After treatment, cells were washed with PBS, trypsinized and collected in culture medium. Cell counting was performed by means of a Bürker's hemocytometer. Three counts for each well were made and the mean value and the standard deviation calculated.

### 9.16. Top-FLASH reporter assay

The Top-FLASH reporter plasmid was transiently transfected into HEK293 cells with Lipofectamine 2000 (Invitrogen, Carlsbad, CA, USA) in 24 -well plates. To activate the Wnt/ $\beta$ catenin pathway, cells were treated with 20 mM LiCl alone or in combination with $1 \mu \mathrm{M}$ rosiglitazone or compounds $(R)-\mathbf{1},(S) \mathbf{- 1}$ or $(R, S)-\mathbf{2}$ for 24 hs. Luciferase activity was determined in the cell extracts 48 hs after transfection. All experiments were performed three times with triplicate replicates.

### 9.17. Protein extract preparation and western blotting analysis

HepG2 cells were incubated for 48 hs with $1 \mu \mathrm{M}(R) \mathbf{- 1},(S)-\mathbf{1},(R, S)-\mathbf{2}$ or $50 \mu \mathrm{M} \mathrm{Wy}-14,643$, starting 24 hs after having been depleted of serum. Proteins were electroblotted onto nitrocellulose membranes (Bio-Rad) and subsequently treated with anti-CPT1 (ARP44796_P050, Aviva Systems Biology), anti-CAC, [28] anti-PPAR $\alpha$ (sc-9000, Santa Cruz) and anti- $\beta$-actin (sc-58619, Santa Cruz) antibodies. The immunoreaction was detected by the Immobilon ${ }^{\text {TM }}$ Western Chemiluminescent HRP Substrate (Millipore).

Treated and untreated HT-29 cells were lysed in Ripa buffer ( $150 \mathrm{mM} \mathrm{NaCl}, 50 \mathrm{mM}$ Tris-HCl, pH 7.6, 10 mM EDTA, $1 \%$ NP-40) containing also a protease inhibitors cocktail (AEBSF, Aprotinin, Bestatin hydrochloride, E-64, Leupeptin hemisulfate salt, Pepstatin A) and then centrifugated at $13,000 \mathrm{rpm}$ for 10 min , at $4^{\circ} \mathrm{C}$. Supernatant containing total proteins was quantified and $80 \mu \mathrm{~g}$ of each sample were used for western blot experiments. Antibodies against p21 waf1/cip1 , cyclin D1, cmyc and $\beta$-actin were purchased from Santa Cruz Biotechnology. The relative intensity of protein bands was measured using the Molecular Imager Chemi-Doc imaging system (Bio-Rad, Hercules CA, USA) and evaluated by the Quantity One software (Bio-Rad, Hercules CA, USA).

### 9.18. Assessment of mitochondrial mass

Changes in mitochondrial mass were evaluated by performing fluorescence microscopy (Floid Cell Imaging Station, Life Technologies). We used CellLight® Mitochondria-GFP BacMam 2.0 (Life Technologies) - a fusion construct of the Leader sequence of E1 alpha pyruvate dehydrogenase and green fluorescent protein (GFP) providing accurate and specific targeting to live cell mitochondria independently of mitochondrial membrane potential - following the manufacturer's instructions. Briefly, $60 \%$ confluent HepG2 cells in serum-free medium were transduced with CellLight ${ }^{\circledR}$ Mitochondria-GFP BacMam 2.0 (50 particles per cell) and treated with $1 \mu \mathrm{M}(R) \mathbf{- 1},(S)$ $\mathbf{1},(R, S)-\mathbf{2}$ or $50 \mu \mathrm{M} \mathrm{Wy}-14,643$ for 24 h . Then, cells were used for microscopy analysis by GFP filter set (excitation at 485 nm and emission at 535 nm ). Images were captured with X20 lens and were processed identically. In another set of experiments, HepG2 cells in serum-free medium were transduced with CellLight ${ }^{\circledR}$ Mitochondria-GFP BacMam 2.0 in the presence or absence of $1 \mu \mathrm{M}(R)$ $\mathbf{1},(S) \mathbf{- 1},(R, S)-\mathbf{2}$ or $50 \mu \mathrm{M} \mathrm{Wy}-14,643$ for 24 hs , rinsed twice with PBS and lysed in buffer containing $10 \%$ SDS and 0.1 M Tris (pH 7.5). Cell extract fluorescence was then measured using GloMax plate reader (Promega) with excitation at 475 nm and emission at $500-550 \mathrm{~nm}$.

### 9.19. Computational chemistry

Molecular modeling and graphics manipulations were performed using Maestro 10.1 (Schrödinger)[67] and UCSF-Chimera 1.8 .1 software packages [68] running on a E4 Computer Engineering E1080 workstation provided of a Intel Core i7-930 Quad-Core processor. GOLD 5.2.2 [41] was used for all docking calculations. Figures were generated using Pymol 1.0.[69]

### 9.20. Protein and Ligand Preparation

The crystal structures of PPAR $\alpha$ and PPAR $\gamma$ complexed with the partial agonists aleglitazar (PDB ID: 3G8I) [45] and LT127 (PDB ID: 2I4P),[44] respectively, were download from the PDB Bank,[70] and employed for the automated docking studies. The proteins were processed through
the Protein Preparation Wizard in Maestro.[71] The right bond orders as well as charges and atom types were assigned and the hydrogen atoms were added to both proteins. Arginine and lysine side chains were considered as cationic at the guanidine and ammonium groups, and the aspartic and glutamic residues were considered as anionicat the carboxylate groups. For PPAR $\gamma$, all crystallographic water molecules were deleted. For PPAR $\alpha$, the two well-defined water molecules bridging between the T 279 OH group on helix 3 and the main chain $\mathrm{C}=\mathrm{O}$ oxygen of L331 on $\beta$-sheet, were included in the docking experiments. Imidazole rings of H 440 into PPAR $\alpha$ and H449 and H323 into PPAR $\gamma$ were set in their $\mathrm{N}^{\varepsilon} 2-\mathrm{H}(\mathrm{N}$ tau-H) tautomeric state. Moreover, an exhaustive sampling of the orientations of groups, whose H -bonding network needs to be optimized, was performed. Finally, the protein structures were refined with a restrained minimization with the OPLS2005 force field [72] by imposing a $0.3 \AA$ root-mean-square deviation (rmsd) limit as the constraint.

The structures of $(R) \mathbf{- 1}$ and $(S)$ - $\mathbf{1}$ enantiomers were built using the fragment dictionary of Maestro and preprocessed with LigPrep 3.3 [73] which prepares the ligands in multiple protonation and tautomerization states at a neutral pH. Ligands were then optimized by Macromodel 10.7 [74] using the MMFFs force field with the steepest descent (1000 steps) followed by truncated Newton conjugate gradient (500 steps) methods. Partial atomic charges were computed using the OPLS-AA force field.

### 9.21. Docking Simulations

Docking of $(R) \mathbf{- 1}$ and $(S)-\mathbf{1}$ to both PPAR $\gamma$ and PPAR $\alpha$ was performed with the GOLD software, which uses a genetic algorithm (GA) for determining the docking modes of ligands and proteins. The coordinates of the co-crystallized ligands aleglitazar and LT127 were chosen as active-site origin. The active-site radius was set equal to $8 \AA$. Each GA run used the default parameters of 100000 genetic operations on an initial population of 100 members divided into five
subpopulations, with weights for crossover, mutation, and migration being set to 95,95 , and 10 , respectively. For PPAR $\alpha$, the two crystal waters connecting T279 and L331 via H-bonds were allowed to toggle on during the individual docking runs (i.e., these waters were present in the binding site).[75] GOLD allows a user-definable number of GA runs per ligand, each of which starts from a different orientation. For these experiments, the number of GA runs was set to 200 without the option of early termination, and scoring of the docked poses was performed with the original ChemPLP scoring function rescoring with ChemScore.[38,43,76] The final receptor-ligand complex for each ligand was chosen interactively by selecting the highest scoring pose that was consistent with experimentally-derived information about the binding mode of the ligand.

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## Supporting information

Proton $\left({ }^{1} \mathrm{H}\right)$ and carbon $\left({ }^{13} \mathrm{C}\right)$ NMR spectra of final compounds.

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