1	Authentication of pineapple (Ananas comosus [L.] Merr.) juice from pulp and peel by
2	HPLC-DAD-(HR)-ESI-MS <sup>n</sup> analysis
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# 14 Abstract

15 Large quantities of by-products, such as peels and trimmings, emerge during industrial pineapple (Ananas comosus [L.] Merr.) processing. The latter may be further exploited by juice extraction. 16 However, the low-quality juices obtained may be marketed as genuine pineapple juice from pulp or be 17 admixed to the latter, thus adulterating the final product. To identify chemical markers, juice was 18 extracted from edible pulp, from flesh adhered to the peel, and from milled peel. The metabolite pattern 19 20 in the juices was elucidated by HPLC-DAD-(HR)-ESI-MS<sup>n</sup>. Unsupervised principal component analysis (PCA) and hierarchical cluster analysis (HCA) calculated on the basis of physico-chemical parameters 21 and metabolite profiles enabled the distinction between juices from pulp and those from peel, 22 irrespective of the fruit maturity degree. In addition, specific ratios of selected marker compounds were 23 calculated, permitting the unambiguous distinction between pulp and peel juices as well as the three 24 25 maturity stages assessed.

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28 Keywords: adulteration; juice quality; phenolic compounds; furanones

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*Abbreviations: CID*, collision-induced dissociation; *GSH*, glutathione; *HDMF*, 4-hydroxy-2,5dimethyl-3(2*H*)-furanone; *HPLC-DAD*, high performance liquid chromatography-diode array detection; *(HR)-ESI-MS*, (high-resolution) electrospray ionisation mass spectrometry; λ<sub>max</sub>, UV maxima; *MDMF*,
4-methoxy-2,5-dimethyl-3(2*H*)-furanone; *n. d.*, not detected; *MW*, molecular weight; PCA, principal
component analysis; *sh*, shoulder; *t*<sub>R</sub>, retention time

## 36 1 Introduction

37 Pineapple (Ananas comosus [L.] Merr.) represents one of the leading fruits of the tropics worldwide. The major producing areas are Southeast Asia and Latin America. Considerable amounts of fresh fruit, 38 39 juice, jam, jelly, and dried pineapple products are exported to North America and Europe (Li et al., 2014; Steingass, Glock, Schweiggert, & Carle, 2015; Mhatre, Tilak-Jain, De, & Devasagayam, 2009). 40 Pineapples and derived products are popular owing to their pleasant aroma and flavour. They contain 41 considerable concentrations of polyphenols, vitamins, and other compounds possibly exerting health 42 43 benefits. Merely 60% of the pineapple infructescence is edible, thus processing residuals range between 45 and 65% (da Silva, Nogueira, Duzzioni, & Barrozo, 2013). The canning industry is producing large 44 45 quantities of liquid and solid wastes *inter alia* peels and trimmings (Li et al., 2014). In order to further exploit these residues; processors also extract juice from the aforementioned by-products. The low-46 valued juice extracted from the peel or even the "mill juice", i.e., the liquid obtained from finely milled 47 pineapple shell, is commonly used as syrup during canning. However, it may also be admixed to 48 pineapple juice, thus adulterating the final product, or even be marketed as the genuine produce. 49

Adulteration of food, and especially beverages, is a serious issue in the global market. Fruit juices 50 51 represent one of the most common targets for fraud (Jandrić et al., 2014). According to the Directive 2001/112/EC, fruit juice shall be exclusively obtained from the edible fraction of the fruits. Pineapple 52 peel is considered as 'not edible', but is a good source for recovery of valuable compounds such as 53 polyphenols. Peel-specific metabolites may provide a useful tool to reveal the fraudulent admixture of 54 55 juice extracted from pineapple shell (Wen & Wrolstad, 2002; Fügel, Carle, & Schieber, 2005; Steingass 56 et al., 2015a). Several analytical methods have been proposed to identify different types of adulterations in fruit juice. Techniques applied comprise profile analysis of sugars, organic acids or flavonoids, as 57 well as analysis of minerals, trace metals, and stable isotopes using high performance liquid 58 59 chromatography or gas chromatography (Ehling & Cole, 2011; Gómez-Ariza, Villegas-Portero, & Bernal-Daza, 2005; Muntean, 2010), capillary electrophoresis (Saavedra, Rupérez, & Barbas, 2001), <sup>1</sup>H 60 61 NMR spectroscopy (Cuny et al., 2008), inductively coupled plasma mass spectrometry (Schwartz & 62 Hecking, 1991), and neutron activation analysis (Anderson, Cunningham, & Alvarez, 1992). In most instances, the analytical tools are comparatively expensive. A metabolomics approach (Jandrić et al., 63

64 2014) also in combination with chemometrics (Jandrić, & Cannavan, 2017) has been proposed as an 65 economically reasonable alternative. However, also a LC-MS based profiling requires advanced 66 analytical tools being unaffordable for, e.g., small juice processing companies.

In most instances, adulterations are difficult to uncover. However, in order to protect the consumers, it is essential to guarantee authenticity and compliance with the product specification. Moreover, from the economic point of view, product authentication is essential to avoid unfair competition that may even destabilise the market (Hong et al., 2017).

For these reasons, the aim of the present work was to establish an appropriate analytical tool to reveal the fraudulent usage of juice from pineapple peel. Specific patterns of metabolites that can be easily detected by HPLC-UV, comprising phenolic and amino compounds as well as furanones and their glycosides, were assessed in pineapple juices obtained from pulp and peel. To unravel a possible influence of fruit maturity, juices were produced from fresh and stored pineapples as well as fruits at the end of their commercial shelf-life. Compound identifications shall be substantiated by detailed HPLC-DAD-(HR)-ESI-MS<sup>*n*</sup> analyses.

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## 79 2 Materials and methods

## 80 2.1 Reagents

4-Hydroxy-2,5-dimethyl-3(2H)-furanone (HDMF), 4-methoxy-2,5-dimethyl-3(2H)-furanone (MDMF), 81 sinapyl alcohol, coniferyl alcohol, gallic acid, and tris(2-carboxyethyl)phosphine hydrochloride were 82 83 purchased from Sigma-Aldrich (Steinheim, Germany). Caffeic, sinapic, p-coumaric, and ferulic acid 84 were from Roth (Karlsruhe, Germany). Syringic acid was from Extrasynthese (Genay, France), Ltyrosine from Fluka Chemie (Buchs, Switzerland), and serotonin hydrochloride from Sigma-Aldrich. 85 Methanol, L-ascorbic, hydrochloric, and meta-phosphoric acid were purchased from VWR International 86 87 (Darmstadt, Germany). Formic acid, Folin-Ciocalteu reagent, formaldehyde solution (~37%), dipotassium hydrogen phosphate (K<sub>2</sub>HPO<sub>4</sub>), sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>), and sodium hydroxide 88 solution (0.25 N, Titripur<sup>®</sup>) were obtained from Merck (Darmstadt, Germany). Potassium hydrogen 89 phthalate was purchased from Th. Geyer (Renningen, Germany). Double-distilled water (ddH2O) 90

prepared with an arium<sup>®</sup> 611 UV (Sartorius, Göttingen, Germany) ultrapure water system was used
throughout.

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# 94 2.2 Pineapple juice samples

95 Air-freighted MD2 (syn. "Extra Sweet") pineapples (*Ananas comosus* (L.) Merr.) from Ghana were 96 purchased from a local fruit distributor (Schumacher, Filderstadt-Bernhausen, Germany). The 97 pineapples were stored for up to two weeks at a temperature of 10–13 °C and a relative humidity of 50– 98 60%.

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# 100 2.2.1 Laboratory scale production

Juice production was performed from fresh pineapples (T0) as well as after seven (T7) and fourteen days (T14) of storage, respectively. Five individual batches were produced each maturity stage. The juice was extracted from pulp with a food mill (Gastroback, Hollenstedt, Germany) and from peel and peel milled with a crusher (Bucher, Niederweningen, Switzerland), respectively, using a Hafico tincture press (Fischer Maschinenfabrik, Neuss, Germany). The pasteurisation was achieved at 90 °C for 5 min by incubating the samples filled in glass bottles in a water bath (Lauda; Lauda-Königshofen, Germany).

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## 109 2.2.2 Pilot plant production

Juice processing was additionally performed at a pilot plant scale using fruits stored for seven days (T7). 110 111 After removing their crowns, pineapples were washed and manually peeled. For the pilot plant production, juice was extracted from pulp using a food mill (Bertuzzi, Brugherio, Italy) equipped with 112 a sieve of 1.6 mm mesh width followed by a second separation step with a 0.8 mm sieve. The juice from 113 peel and adhered pulp and from milled trimmings was extracted using a Wahler (Stuttgart, Germany) 114 115 and a Bucher (Niederweningen, Switzerland) pack press, respectively. The juice was pasteurised at 90 °C using a pilot plant scale pasteuriser (Ruland Engineering & Consulting, Neustadt, Germany). 116 Samples were filled in 0.1-L clear glass bottles using a Schmalbach-Lubeca (Braunschweig, Germany) 117

filling and sealing machine, and immediately cooled to ~8 °C in a water bath. Two individual batches
were produced for each juice type.

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# 121 2.3 Physico-chemical parameters

For the determination of total soluble solids (TSS) and pH, a digital refractometer (RX-5000; Atago, 122 Tokyo, Japan) and pH-meter (inoLab pH 720; WTW, Weilheim, Germany), respectively, were used. 123 Titratable acidity (TA), expressed as citric acid in g/100 mL of juice, was determined by titration with 124 125 normalised 0.25 M NaOH to a pH of 8.1 using an automatic titration system (Titrino 718 STAT; Metrohm, Herisau, Switzerland). Subsequently, a formaldehyde solution adjusted to pH 8.1 was added. 126 The formol number (FN) expressed as mL 0.1 M NaOH/100 mL juice was determined by re-titration to 127 pH 8.1. Potassium hydrogen phthalate was used for normalisation of the titration solution. All physico-128 chemical parameters were determined in duplicate. The determination of total phenol content (TPC) was 129 performed by Folin-Ciocalteu method according to Difonzo et al. (2017) with some modifications. 20 130 µL of juice was added to 980 µL of ddH2O and 100 µL of Folin-Ciocalteu reagent. After 3 min, 5% 131 Na<sub>2</sub>CO<sub>3</sub> solution was added, following incubation at room temperature for 60 min. The absorbance was 132 133 read at 750 nm using a spectrophotometer (Perkin Elmer, Dreieich, Germany). The TPC was expressed as gallic acid equivalents (GAE) in mg/100 mL juice. 134

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## 136 2.4 Quantitation of vitamin C

Vitamin C, i.e., the sum of ascorbic and dehydroascorbic acid, was quantitated by HPLC-UV following a modified procedure according to Aschoff et al. (2015). Briefly, an aliquot of 0.5 mL of juice was made up to 5 mL with water containing 1.5% (w/v) meta-phosphoric acid and 20 mM tris(2carboxyethyl)phosphine hydrochloride. The pH of the buffer was adjusted to 3.5 with aqueous 2 M K<sub>2</sub>HPO<sub>4</sub> beforehand. After thorough mixing and incubation for 30 min, the samples were filtered through 0.45 µm regenerated cellulose filters (Chromafil<sup>®</sup>, Macherey-Nagel, Düren, Germany) into amber glass vials and immediately analysed by HPLC.

HPLC analyses were carried out using a Merck Hitachi LaChrom Elite HPLC system equipped with an
L-2130 pump module, an L-2200 autosampler, a JetStream 2 plus column thermostat, and an L-2450

diode array detector (all from Hitachi High-Technologies, Tokyo, Japan). The HPLC system was 146 operated with a RP-C18 column (Kinetex<sup>TM</sup>, 250 × 4.6 mm, 5 µm particle size, 100 Å pore size), 147 protected by a guard column of the same material (both from Phenomenex, Aschaffenburg, Germany). 148 Water containing 1% (w/v) meta-phosphoric acid, adjusted to pH 2.6 with 2 M aqueous K<sub>2</sub>HPO<sub>4</sub>, was 149 used as eluent, applying an isocratic elution at 25 °C. Total run time was 20 min at a flow rate of 150 1.0 mL/min. The injection volume was 10 µL. Ascorbic acid was detected at a wavelength of 254 nm, 151 and quantitated by external linear calibration. The reported vitamin C concentrations equal the sum of 152 153 ascorbic and dehydroascorbic acid.

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## 155 **2.5 HPLC-DAD-ESI-MS<sup>n</sup> analysis**

Juice samples were centrifuged for 10 min at  $10.000 \times g$  (MiniSpin plus, Eppendorf, Wesseling-Berzdorf, Germany) and subsequently filtered with 0.45 µm regenerated cellulose filters (Chromafil<sup>®</sup>, Macherey-Nagel, Düren, Germany). HPLC-DAD-ESI-MS<sup>*n*</sup> analyses were conducted applying an Agilent 1100 series HPLC system (Agilent Technologies, Waldbronn, Germany) fitted with the abovementioned RP-C18 column. Elution solvents, the gradient, and system settings were used as reported previously (Steingass et al., 2015a). Electrospray ionisation (ESI) mass spectra were acquired at a scan range of *m/z* 50–800 using an Esquire 3000+ ion-trap mass spectrometer (Bruker Daltonics, Bremen, Germany).

## 164 2.6 HPLC-DAD-HR-ESI-MS analysis

HPLC-DAD-HR-ESI-MS analyses were performed using an Agilent 1290 UHPLC system interfaced
with a Q Exactive Plus high-resolution mass spectrometer (Thermo Fisher Scientific, Bremen,
Germany). HPLC conditions and mass spectrometer settings were as given above and detailed in
Steingass et al. (2015a), respectively. Data evaluation was performed with Esquire Control software
(Bruker).

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# 171 2.7 Quantitation of amino acids, amines, furanones, and phenolic compounds by HPLC-DAD

HPLC-DAD analysis was conducted applying a Waters Acquity H-class UPLC, equipped with a
quaternary solvent manager, a sample manager, and a photodiode array detector (all from Waters,

Milford, USA) applying the conditions detailed in Steingass et al. (2015a). External standards were used 174 for quantitation using linear calibration curves. However, most of the pineapple-specific metabolites 175 were not commercially available. Thus, structurally related compounds with a similar chromophore 176 177 were used and molecular weight (MW) correction factors (MW<sub>compound</sub>/MW<sub>standard</sub>) were applied. HDMF and MDMF standards were used for the quantitation of free and glycosylated furanones. Coniferyl and 178 p-coumaryl derivatives, i.e., S-coniferyl-L-cysteine, S-p-coumarylglutathione, S-coniferylglutathione, 179  $N-L-\gamma$ -glutamyl-S-coniferyl-L-cysteine were quantitated with coniferyl alcohol, the sinapyl conjugates 180 181 S-sinapyl-L-cysteine, S-sinapylglutathione, N-L-γ-glutamyl-S-sinapyl-L-cysteine with sinapyl alcohol. Ferulic acid was used for (di-E,E)-N,N'-diferuloylspermidine quantitation, syringic acid for syringoyl 182 hexoside, sinapic acid for sinapoyl hexoside, and caffeic acid for caffeoylisocitrate. L-Tyrosine and 183 serotonin were quantitated using authentic reference standards. The concentrations were expressed as 184 mg/100 mL of juice. 185

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#### 187 2.8 Statistics

The analysis of variance (ANOVA) and Tukey's test for multiple comparison were carried out using Minitab Statistical Software (Minitab Inc., State College, PA, USA) considering the pineapple tissue (T) and storage (S) as independent variables and their interaction ( $T^*S$ ); the uppercase letters indicate the statistical differences resulting from two-way ANOVA and Tukey's test above mentioned. To visualise the clustering of the samples, unsupervised principal component analysis (PCA) and hierarchical cluster analysis (HCA) were calculated with Solo software version 8.0.1 (Eigenvector Research, Wenatchee, WA, USA).

Box-plots were constructed using GraphPad software (GraphPad Software Inc., La Jolla, CA, USA).
Boxes represent the lower and upper quartiles (25 and 75%), the bands inside the boxes the median
(50%). Arithmetic means are indicated by cross symbols, minimum and maximum values by the
whiskers, and outliers by circles.

#### 200 3 Results

#### 201 **3.1 Pineapple juice characterisation**

Table 1 summarises pH, TSS, TA, TSS/TA ratio, formol number, TPC, and vitamin C of the pineapple 202 203 juice from pulp, pressed and milled peel during storage. The pH was not significantly influenced from the tissue variable (T). No clear trend was observed during storage. TSS were significantly higher (p < p204 0.0001) in juice from pulp  $(12.57 \pm 0.51 \text{ to } 13.07 \pm 0.60 \text{ g}/100 \text{ g})$  than from pressed and milled peel 205  $(10.74 \pm 0.46 \text{ to } 11.89 \pm 0.35 \text{ g}/100 \text{ g})$ . No significant differences were found during storage. Both the 206 207 values of TA and formol number significantly increased during storage (T14) of the peel juices, whereas they did not vary significantly in those from pulp. Differences in calculated TSS/TA ratios were found 208 to be insignificant with the exception of juices obtained from peel after two weeks of storage (T14). 209

Elevated TPC values of  $92.06 \pm 4.90$  and  $95.52 \pm 5.11$  mg GAE/100 mL were detected in milled peel juice at T7 and T14, respectively. In the pulp juice, both TPC and vitamin C significantly decreased after fourteen days (T14).

In Table S1, the physico-chemical characteristics of pineapple juice produced at a pilot plant scale were compiled. The tissue variable *T* significantly influenced all considered parameters with the exception of TA. Moreover, the juice obtained at laboratory scale at T7 showed the same trend for all the parameters. Consequently, the results from small-scale processing also apply for the juice produced at the pilot plant. However, significantly elevated concentrations of vitamin C between  $57.55 \pm 9.14$  and  $61.46 \pm 9.35$ mg/100 mL were found in the juice produced at pilot plant scale compared to the laboratory scale (32.57  $\pm 2.28$  and  $38.35 \pm 3.59$  mg/100 mL). This may be attributed to the differing thermal treatments applied.

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# 221 **3.2** Identification of individual compounds

Table 2 summarises the UV absorption spectra, accurate mass, and  $MS^n$  mass fragmentations of the pineapple metabolites that were assigned according to Steingass et al. (2015a) and Steingass, Glock, Lieb, & Carle (2017).

L-Tyrosine (no. 1) had a protonated molecule  $[M+H]^+$  at m/z 182.0813 as well as an abundant precursor 227 ion at m/z 165.0547 resulting from in-source deamination ([M+H-NH<sub>3</sub>]<sup>+</sup>). CID of the protonated 228 molecule resulted in prevailing fragment ions at m/z 165, 147, and 136 from the elimination of NH<sub>3</sub> (17 229 230 amu), H<sub>2</sub>O (18 amu) and NH<sub>3</sub> well as CO (28 amu) and H<sub>2</sub>O. CID of the [M+H-NH<sub>3</sub>]<sup>+</sup> precursor ion at m/z 165 resulted in a unique product ion at m/z 123. Similarly, both  $[M+H]^+$  and  $[M+H-NH_3]^+$  at m/z231 177.1023 and 160.0758 were detected in the MS<sup>1</sup> spectrum of serotonin (no. 2). CID of the [M+H]<sup>+</sup> at 232 m/z 177 resulted in a single fragment ion at m/z 160 ([M+H-HN<sub>3</sub>]<sup>+</sup>), thus confirming the identity as an 233 234 amine. L-Tryptophan (no. 6) had a deprotonated molecule [M-H]<sup>-</sup> at m/z 203.0820 and fragment ions at m/z 159 [M-H-CO<sub>2</sub>]<sup>-</sup> and 142 ([M-H-CO<sub>2</sub>-NH<sub>3</sub>]<sup>-</sup>) in the ESI(-)-MS<sup>n</sup> experiment. The elimination of NH<sub>3</sub> 235 was also observed in the positive ion mode. CID of the  $[M+H]^+$  at m/z 205.0972 resulted in an abundant 236 product ion at m/z 188 ([M+H-NH<sub>3</sub>]<sup>+</sup>). 237

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#### 239 Furanones

The mass spectrometric assignment of a HDMF hexoside (no. 3) and two corresponding malonyl 240 hexosides (nos. 8 and 9) in the negative ion mode has been previously reported in detail (Steingass et 241 242 al., 2017). Noteworthy, the CID of the abundant sodium adducts [M+Na]<sup>+</sup> detected in the positive ion mode at m/z 313.0895 (no. 3) and 399.0900 (nos. 8 and 9), respectively, resulted in a product ion at m/z243 151 resembling [HDMF+Na]<sup>+</sup>. The analogous elimination of a dehydrated hexose (162 amu) from no. 244 3 and dehydrated malonyl hexoses (248 amu) from nos. 8 and 9, respectively, was observed from the 245 246 protonated molecules, resulting in [HDMF+H]<sup>+</sup> at m/z 129. The latter was detected at m/z 129.0548 for 247 compound no. 4 assigned to free HDMF. CID resulted in product ions at m/z 111 ([M+H-H<sub>2</sub>O]<sup>+</sup>), 101  $([M+H-CO]^+)$ , and 83  $([M+H-H_2O-CO]^+)$ , possibly generated by water (18 amu) and carbon monoxide 248 (28 amu) eliminations from the keto- and the hydroxyl groups of the protonated HDMF molecule. A 249 250 further free furanone was assigned to the methoxy derivative MDMF (no. 7) displaying a protonated molecule  $[M+H]^+$  at m/z 143.0704. The most abundant ion in the MS<sup>2</sup> spectrum was detected at m/z 111 251 ([M+H-CH<sub>3</sub>OH]<sup>+</sup>), resulting from the elimination of methanol (32 amu) from the methoxy group. 252 Assignment of the latter two compounds as well as nos. 1, 2, and 6 was substantiated by comparing their 253 254  $t_{\rm R}$ , UV, and mass spectra with those of authentic reference standards.

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#### 256 *Phenolic compounds*

The mass spectrometric assignment of *S*-sinapyl-L-cysteine (no. 12), *S*-sinapylglutathione (no. 16), and *N*-L- $\gamma$ -glutamyl-*S*-sinapyl-L-cysteine (no. 19) as well as the structurally related *S*-*p*-coumaryl (no. 14) and *S*-coniferyl conjugates (nos. 11, 15, and 18) by HR-ESI-MS and MS<sup>*n*</sup> experiments have been discussed in detail in our previous contributions (Steingass et al., 2015a, Steingass et al., 2017).

Compounds no. 13, and 17 were assigned to caffeoyl (no. 13) and p-coumaroylisocitrates (no. 17). Their 261 assignment by HR-ESI(-)-MS and ESI(-)-MS<sup>n</sup> experiments has been previously reported (Steingass et 262 al., 2017). The assignment of these acyl-isocitrates was substantiated by the MS experiment in the 263 positive ion mode. Both acyl-isocitrates displayed abundant sodium adducts  $[M+Na]^+$  at m/z 377.0840 264 (no. 13) and 361.0529 (no. 17). In the MS<sup>2</sup> experiment, product ions at m/z 215 and 197 resembling the 265 sodium adduct of isocitric ([isocitric acid+Na]<sup>+</sup>) and dehydrated isocitric acid ([isocitric acid-H<sub>2</sub>O+Na]<sup>+</sup>) 266 were detected. These product ions were generated by the elimination of caffeic (180 amu) and 267 dehydrated caffeic acid (162 amu) from no. 13, and the corresponding *p*-coumaric acid eliminations 268 (164 and 146 amu) from no. 17. In addition, cleavage of the ester bonds resulted in characteristic product 269 270 ions resembling the sodium adducts and protonated molecules, respectively, of the dehydrated hydroxycinnamic acids. This reaction, i.e., the elimination of isocitric acid (192 amu) and dehydrated 271 isocitric acid (174 amu) resulted in product ions at m/z 185 ([caffeic acid-H<sub>2</sub>O+Na]<sup>+</sup>) and 163 ([caffeic 272 acid-H<sub>2</sub>O+H]<sup>+</sup>) in the MS<sup>2</sup> experiment of the caffeoyl, and m/z 169 ([p-coumaric acid-H<sub>2</sub>O+Na]<sup>+</sup>) and 273 147 ([*p*-coumaric acid-H<sub>2</sub>O+H]<sup>+</sup>) of the *p*-coumaroylisocitrate. 274

The assignment of syringoyl (no. 5) and sinapoyl hexosides (no. 10) as well as (di-E,E)-N,N'diferuloylspermidine (no. 20) on the basis of their UV and mass spectra has been previously reported (Steingass et al., 2015a, Steingass et al., 2017). Interestingly, the phenolic glycosides no. 5 and 10 also displayed sodium adducts [M+Na]<sup>+</sup> at *m/z* 383.0949 and 409.1108, respectively, in the positive ion mode as also detected for the abovementioned phenolic esters.

#### 281 **3.3 Quantitation of selected compounds**

282 Table 3 summarises the concentrations of L-tyrosine, serotonin, furanones, and phenolic compounds in pulp and peel (pressed and milled) juices. L-Tyrosine and serotonin are common pineapple juice 283 284 constituents (Wen et al., 2002). Their concentrations in the juice from pulp were in accordance with those reported by others (Fedman & Lee, 1985; Wen et al., 2002). The concentration of serotonin was 285 not significantly influenced from T and S. L-Tyrosine showed a significantly higher concentration in the 286 287 milled peel juices at T7 and T14. HDMF exerting an intense aroma described as "burnt pineapple" or 288 "fruity" and "caramel-like" has been reported for the first time in pineapples by Rodin et al. (1965). The 289 HDMF concentrations in all types of juices significantly increased with progressing storage duration. 290 Juice from fresh pineapple pulp contained 1.26 mg/100 mL (T7) and the concentration amounted to 2.52mg/100 mL during storage (T14). Similarly, Brat et al. (2004) and Steingass et al. (2015b) have reported 291 292 increasing HDMF concentrations with progressing maturation in the pulp of fresh pineapple fruits. Concomitantly, the concentrations of the methoxy derivative MDMF significantly increased during 293 storage. Comparable concentrations of HDMF like those determined herein have been previously 294 reported in the literature (Elss et al., 2005; Tokitomo, Steinhaus, Büttner, & Schieberle, 2005). 295 296 Noteworthy, also the concentrations of HDMF hexoside increased during storage, whereas those of the two HDMF malonyl hexoside isomers remained constant. The highest values of the latter were observed 297 in juice from milled peel. 298

299 In agreement with previous studies, the most abundant phenolic compounds were S-sinapyl-L-cysteine, 300 S-sinapylglutathione, and N-L-y-glutamyl-S-sinapyl-L-cysteine (Steingass et al., 2017; Wen, Wrolstad, 301 & Hsu, 1999) amounting to 8.12, 4.94, and 6.97 mg/100 mL, respectively, in juice from fresh pulp. Even though no significant difference between the total concentrations of phenolic compounds determined by 302 303 HPLC in pulp and peel (pressed and milled) juices was found, the individual compounds showed 304 different trends. The variable T significantly (p < 0.0001) influenced all phenolic compounds with the 305 exception of p-coumaroylisocitrate. Elevated concentrations of sinapoyl hexoside and S-sinapyl-L-306 cysteine were found in juice from pulp compared to those from pressed and milled peel. Whereas the 307 storage duration (S) did not show a significant influence on sinapoyl hexoside, the concentration of S-308 sinapyl-L-cysteine increased in the pressed and milled peel juices. The opposite trend was found for S- 309 coniferyl-L-cysteine, caffeoylisocitrate, *N*-L- $\gamma$ -glutamyl-*S*-coniferyl-L-cysteine, and (di-*E*,*E*)-*N*,*N'*-310 diferuloylspermidine. The aforementioned compounds showed significantly higher concentrations in 311 juice from pressed and milled peel. In most instances, their concentration was higher in peel juices from 312 stored pineapples. The same trends were observed for pineapple juice produced at a pilot plant scale 313 (Table S2).

314

# 315 **3.4** Multivariate statistics and marker identification

#### 316 Unsupervised pattern recognition by PCA and HCA

317 In order to further explore differences among the juices, the physico-chemical parameters (Table 1) and the concentrations of the pineapple juice constituents (Table 3) were subjected to HCA and PCA (Fig. 318 2). Fig. 2 a,a' displays the PCA that was calculated on the basis of the physico-chemical parameters. 319 The first three principal components (PCs) of the model explained 78% of the variance among the 320 considered parameters with a contribution of PC1 of 44%. Two separate clusters were formed. All pulp 321 juices and those from fresh peel (pressed and milled) formed one cluster, the second comprised the peel 322 juices from stored pineapples (T7 and T14). As deduced from the location of the loadings, the parameters 323 324 related with juice from fresh pineapple were TSS and TSS/TA. By contrast, the remaining parameters contributed to the differentiation of the peel juice from stored pineapples. 325

Moreover, HCA and PCA were calculated on the basis of the concentrations of L-tyrosine, serotonin, furanones, and phenolic compounds (Table 3). All pulp samples formed one cluster, whereas a second comprising pressed and milled peel juice samples was differentiated when plotting PC1 against PC2. Moreover, a distribution of the samples according to the storage duration was observed (Fig. 2 b,b').

Fig. 2c,c' illustrates the PCA obtained from both physico-chemical parameters and the chemical composition. The combination of both data sets resulted in a clear-cut differentiation of all pulp from all peel samples as well as an arrangement of the samples according to the storage duration of the fruits. Consequently, marker compounds that contribute to the differentiation of pulp and peel samples and those describing the influence of storage may be deduced from this plot. The compounds with positive loadings on PC2 are related with juices from pulp, those with negative loadings with pressed and milled peel. In line with the results compiled in Table 3, sinapoyl hexoside (no. 10) and *S*-sinapyl-L-cysteine (no. 12) were related with pulp juice showing positive loadings on PC2. *N*-L- $\gamma$ -Glutamyl-*S*-coniferyl-Lcysteine (no. 19) with a negative loading on PC2 contributed to the separation of the peel juice. Similarly, HDMF hexoside (no. 3) and the HDMF malonyl hexosides (nos. 8 and 9) contributed to the clustering described above.

341

342 PLS-DA and PLS regression

343

# 344 Marker identification

Calculating the ratios between peel- and pulp-specific compounds may permit an unambiguous distinction of the juices. Selected ratios are illustrated by Fig 3. HDMF hexoside concentrations increased during storage, whereas those of the two HDMF malonyl hexoside isoforms remained constant (Table 3). Consequently, the calculated ratio for both pulp and peel juices showed significantly higher values at T14 (Fig. 3a). Still, these ratios permitted to discriminate juices from pulp, pressed, and milled peel, independently of the storage time of the pineapples.

The glutathione (GSH) conjugates (nos. 14, 15, and 16) had negative, the *N*-L- $\gamma$ -glutamyl-L-cysteine and L-cysteine derivatives (nos. 11, 12, 19, and 20) positive loadings on PC1. The ratio between the aforementioned compounds was calculated that may possibly describe the effect of storage. However, merely for pulp and milled peel juice a significantly higher ratio was found at T14 (Fig. 3b) due to the comparatively large range of this parameter (see boxplots in Fig. 3b).

Caffeoylisocitrate (no. 13) is a compound related with peel juice as deduced from the loadings plot (Fig. 2b and c) and the concentrations compiled in Table 3. The highest concentrations were detected in milled peel juice. A further acyl-isocitric acid, namely *p*-coumaroylisocitrate (no. 18) was not correlated with the juice type (as shown in Table 3). Calculating the ratio between caffeoyl and *p*-coumaroylisocitrates permitted the clear-cut distinction of milled peel juice (Fig. 3c). In addition, the (*S*-sinapyl-L-cysteine)/*N*-L- $\gamma$ -glutamyl-*S*-coniferyl-L-cysteine ratio permitted the

In addition, the (S-sinapyl-L-cysteine)/N-L-γ-glutamyl-S-coniferyl-L-cysteine ratio permitted the
 unambiguous discrimination of pulp juice from all peel-derived samples (Fig. 3d).

365 Fruit juices represent an important and rapidly growing sector of the beverage industry. Similar to other high-priced food commodities, fruit juices and purees are targets for adulterations (Hong et al., 2017). 366 367 Our approach for detecting the usage of juice obtained from the non-edible parts of the pineapple infructescence was based on the determination of specific markers for pulp and peel (pressed and peel) 368 juices. Jandrić et al. (2014) used a metabolomics approach for detecting fruit juice adulterations. 369 Moreover, Steingass et al. (2015a) have previously studied the phenolic pattern of different tissues of 370 371 the pineapple infructescence and have proposed phenolic compounds as possible tools to discriminate pulp and peel. In the present study, among the physico-chemical parameters, merely TSS and TPC 372 permitted the discrimination between pulp and peel juice (see Table 1). The metabolites detected in the 373 juices from pulp and peel permitted to discriminate both juice categories by PCA. Moreover, this 374 multivariate statistical tool allowed deducing specific chemical markers and ratios calculated therefrom 375 to discriminate between the individual samples. The higher ratio of caffeoylisocitrate/p-376 coumaroylisocitrate was found to be characteristic of juice extracted from milled peel. In addition, the 377 S-sinapyl-L-cysteine)/N-L-γ-glutamyl-S-coniferyl-L-cysteine 378 ratio (sinapoyl hexoside +was 379 significantly higher in pulp juice, thus permitted to differentiate genuine pineapple juice from minor quality juices, i.e., those from pressed peel or mill juice. Regardless of the storage time, the mean value 380 of this ratio in juice from pulp was 11 and 6 in those from pressed and milled peel. The maximum value 381 determined among all peel juices was 7.8. Concluding, a (sinapoyl hexoside + S-sinapyl-L-cysteine)/N-382 383 L-y-glutamyl-S-coniferyl-L-cysteine ratio higher than 8 may indicate that no peel extract was added to 384 pulp juice. Moreover, among all juice categories, the ratio HDMF hexoside/HDMF malonyl hexosides was significantly higher in the juice from pineapples processed after two weeks (T14). Consequently, 385 this ratio may represent a suitable marker to evaluate the pineapple maturity and freshness. The 386 387 applicability of the aforementioned parameters for authentication of juices produced at a large scale was 388 successfully demonstrated (see data compiled in Table S2).

# 390 5 Conclusions

HPLC-DAD-(HR)-ESI-MS<sup>n</sup> profiling, HPLC-DAD quantitation and subsequent PCA analysis 391 permitted to unravel distinctive chemical markers to authenticate pineapple juices from edible pulp, pulp 392 adhered to the peel (pressed peel), and inedible peel fractions (mill juice). Overall, the present 393 contribution revealed distinct chemical markers to authenticate pineapple juices from peel and pulp. 394 They may be suitable tools in the quality assurance of fruit processing companies or the official food 395 control. Future studies may define the levels of admixed peel juice that can be evidenced using the 396 397 proposed methodology, e.g., by analysing genuine pulp juice spiked with different levels of peel juices. In addition, continuative studies may further explore the applicability of the proposed indicators 398 determined using MD2 "Extra Sweet" pineapples from Ghana, to authenticate pulp and peel juices from 399 different genotypes, provenances or harvesting seasons. 400

401

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407

#### 408 6 References

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484 7 Captions

485 **7.1 Figures** 

Fig. 1 HLPC-DAD chromatogram (280 nm) of pineapple (*Ananas comosus* [L.] Merr.). juice from pulp
(a), pressed (b), and milled peel (c). The chemical structure of selected compounds are displayed. For
peak assignment, see Table 2.

489

Fig. 2 Score plots of the principal component analysis (PCA) calculated on the basis of physico-chemical parameters (a), pineapple juice constituents (b) and both data sets (c). The corresponding loading plots are displayed in a', b', and c'. The circles in the score plots illustrate clusters from hierarchical cluster analysis (HCA). Circle, rhombus, and triangle represent juice from pulp, pressed, and milled peel, respectively. White, grey, and black colour indicate T0 (fresh pineapples), T7 (one week of storage), and T14 (two weeks of storage), respectively.

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497 **Fig. 3** Box plots illustrating the ratio of HDMF hexoside/HDMF malonyl hexosides (a), (*N*-L-γ-498 glutamyl-L-cysteines + L-cysteines)/GSHs (b), caffeoylisocitrate/*p*-coumaroylisocitrate (c), (sinapoyl 499 hexoside + *S*-sinapyl-L-cysteine)/*N*-L-γ-glutamyl-*S*-coniferyl-L-cysteine (d), in juices from pulp and 500 peel (pressed and milled). Different letters indicate significant (p < 0.05) differences determined by 501 ANOVA and Tukey's test.

502

503 7.2 Tables

Table 1. Physico-chemical parameters of pineapple (*Ananas comosus* [L.] Merr.) juice from pulp and
 peel.

506

Table 2. HPLC-DAD-(HR)-ESI-MS<sup>n</sup> data of phenolic compounds and other metabolites detected in
 pineapple (*Ananas comosus* [L.] Merr.) juice.

509

Table 3. Quantitation of phenolic compounds and other metabolites in pulp and peel juice obtained from
three pineapple (*Ananas comosus* [L.] Merr.) maturity stages.

# **7.3** Appendix

- Table S1. Physico-chemical parameters of pineapple (*Ananas comosus* [L.] Merr.) juice from pulp and
  peel produced at pilot plant and laboratory scale (both T7).
- **Table S2.** Quantitation of phenolic compounds and other metabolites in pineapple (*Ananas comosus* [L.]
- 518 Merr.) juice produced at pilot plant and laboratory scale (both T7).

# Table 1.

				Pulp			Peel (pressed)			Peel (milled)		
Parameter	Т	S	$T^*S$	Т0	T7	T14	Т0	T7	T14	Т0	T7	T14
pН	0.278	0.000	0.722	$4.03\pm0.02^{\rm b}$	$4.12\pm0.04^{\rm a}$	$4.02\pm0.02^{\text{b}}$	$4.11\pm0.02^{\rm a}$	$4.02\pm0.01^{\text{b}}$	$4.10\pm0.01^{\rm a}$	$4.03\pm0.02^{\text{b}}$	$4.11\pm0.01^{\mathtt{a}}$	$4.03\pm0.01^{\rm b}$
TSS (g/100 g)	0.000	0.000	0.030	$13.1\pm0.6^{\rm a}$	$12.6\pm0.4^{\text{a}}$	$12.6\pm0.5^{ab}$	$11.6\pm0.7^{\text{cd}}$	$10.7\pm0.5^{\text{e}}$	$11.0{\pm}~0.2^{\text{de}}$	$11.7\pm0.4^{\text{cd}}$	$11.1\pm0.6^{\text{de}}$	$11.89\pm0.35^{\rm bc}$
TA (g/100 mL)	0.013	0.000	0.000	$0.49\pm0.02^{\texttt{bc}}$	$0.46\pm0.07^{\text{bc}}$	$0.52\pm0.03^{\text{b}}$	$0.43\pm0.09^{\rm c}$	$0.45\pm0.04^{\text{bc}}$	$0.61\pm0.04^{\rm a}$	$0.47\pm0.05^{\texttt{bc}}$	$0.47\pm0.04^{\rm bc}$	$0.66\pm0.02^{\rm a}$
TSS/TA	0.000	0.000	0.012	$26.5\pm2.0^{\rm a}$	$27.7\pm3.5^{\rm a}$	$24.0\pm1.9^{\rm a}$	$28.0\pm6.6^{\rm a}$	$23.9\pm2.7^{\rm a}$	$18.0\pm1.3^{\text{b}}$	$25.3\pm2.9^{\rm a}$	$23.9\pm3.0^{\rm a}$	$18.03\pm0.56^{\text{b}}$
Formol number (mL 0.1 M NaOH/100 mL)	0.000	0.000	0.000	$8.08\pm0.57^{\rm c}$	$8.84 \pm 1.32^{\circ}$	$8.50\pm0.55^{\rm c}$	$8.26\pm2.11^{\circ}$	$12.43\pm0.98^{\text{b}}$	$12.09\pm0.86^{\text{b}}$	$9.02\pm0.94^{\circ}$	$11.46\pm0.87^{b}$	$14.83 \pm 1.44^{\rm a}$
TPC (mg GAE/100 mL)	0.000	0.004	0.000	$78.9\pm3.3^{\text{d}}$	$80.2\pm4.7^{\rm cd}$	$72.3\pm3.3^{\text{e}}$	$82.8\pm4.8^{\text{cd}}$	$86.2\pm4.7^{bc}$	$82.3\pm2.4^{\rm cd}$	$86.2\pm4.1^{bc}$	$92.1\pm4.9^{ab}$	$95.5\pm5.1^{\mathtt{a}}$
Vitamin C (mg/100 mL)	0.002	0.000	0.017	$50.6\pm6.7^{\text{abcd}}$	$57.6\pm9.1^{\text{ab}}$	$42.0\pm4.6^{\rm d}$	$39.5\pm 6.4^{\rm d}$	$59.4\pm7.7^{\rm ab}$	$42.6\pm2.6^{\texttt{cd}}$	$48.4\pm8.0^{\text{bcd}}$	$61.5\pm9.4^{\rm a}$	$53.4\pm4.8^{\text{abc}}$

*T*, pineapple tissue variable; *S*, storage duration of the fruits; *T*\**S*, interaction of the variables; T0, fresh pineapples; T7, one week of storage at room temperature; T14, two weeks of storage. Different letters in one row indicate a significant (p < 0.05) difference of means determined by two-way ANOVA and Tukey's test.

Table 2.

No.	t <sub>R</sub> (min)	$\lambda_{max}$ (nm)	HR-ESI(-)-MS ( <i>m</i> / <i>z</i> )	ESI(-)-MS <sup>n</sup> experiment ( <i>m</i> / <i>z</i> )	HR-ESI(+)-MS ( <i>m</i> / <i>z</i> )	$ESI(+)-MS^n$ experiment $(m/z)$	Proposed structure
1	6.7	274, sh281	n.d.	n.d.	182.0813 ° (182.0812) 165.0547 <sup>d</sup>	[182]: 165, 147, 136, 123, 119 [165]: 147, 123	L-Tyrosine
					(165.0546)	[105]. 147, 125	
2	8.1	277, sh298	n.d.	n.d.	177.1023 °	[177]: 160	Serotonin
					(177.1022)		
					160.0758 <sup>a</sup>	[160]: 95	
3	12.9	276	289.0930ª	[289]: 161 127 113 101	(160.0756) 313.0895 °	[313] 184 169 151	4-Hydrovy-2 5-dimethyl-3(2H)-
5	12.9	270	(289.0929)	[207]. 101, 127, 115, 101	(313.0894)	[515]. 164, 169, 151	furanone hexoside
			(		291.1075°	[291]: 129	
					(291.1074)		
4	13.2	287	n.d.	n.d.	129.0548 °	[129]: 111, 101, 87, 83	4-Hydroxy-2,5-dimethyl-3(2 <i>H</i> )-
5	179	270	250 0099 a	[250], 107	(129.0546) 282.0040 °	[202], 270 221 105	Turanone Suringovi hovosido
5	17.0	219	(359.0988)	[339]. 197	(383.0949)	[303]. 270, 221, 103	Syningoyi nexoside
6	17.8	sh271, 279, 288	203.0820 ª	[203]: 159, 142, 116	205.0972 °	[205]: 188, 146	L-Tryptophan
			(203.0826)		(205.0972)		
					188.0706 <sup>d</sup>	[188]: 146, 119	
7	21.7	270			(188.0706)	[142], 120, 111, 92	4 Mathema 2.5 dimethad
/	21.7	279	n. d.	n. d.	$(143.0704^{\circ})$	[143]: 129, 111, 83	4-Methoxy-2,5-dimethyl- 3(2H)-furanone
8	21.9	277	331.1037 <sup>ь</sup>	[331]: 289, 271, 203, 193,	399.0900 °	[399]: 355, 151	4-Hvdroxy-2.5-dimethyl-3(2 <i>H</i> )-
			(331.1035)	169, 161, 127, 113, 101	(399.0898)	[,],	furanone malonyl hexoside (1)
					377.1080 °	[377]: 129	
0	22.4	277	221 1027h	[221] 200 271 202 102	(377.1078)	[200]. 255 151	4 Hadresson 2.5 dimethod 2(21)
9	22.4	211	$(331,103)^{-1}$	[551]: 289, 271, 205, 195, 169, 161, 127, 113, 101	(399.0900*	[399]: 355, 151	4-Hydroxy-2,5-dimethyl- $3(2H)$ - furanone malonyl hexoside (2)
			(551.1055)	10), 101, 12/, 115, 101	377.1080 °	[377]: 129	
					(377.1078)		
10	24.4	331	385.1145 ª	[385]: 223, 205	409.1108 °	[409]: 247, 185	Sinapoyl hexoside
11	25.0	268 -1202	(385.1140)		(409.1105)	[204], 162 121 102	S Coniforni L ovistaina
11	23.8	208, \$11505	n. a.	n. u.	284.0947	[284]: 105, 151, 105	S-Conneryi-L-cysteme
12	26.3	282	n. d.	n. d.	314.1052 °	[314]: 193, 161, 133	S-Sinapyl-L-cysteine
					(314.1057)		
13	28.1	299, 328	353.0516ª	[353]: 191, 173, 155, 111	377.0840 °	[377]: 215, 197, 185, 163	Caffeoylisocitrate
1.4	20.2	266 1200	(353.0514)	[420] 20( 200 254	(377.0479)	[440] 200 170 170	
14	30.3	266, sh300	438.134/"	[438]: 306, 288, 254	440.1482° (440.1486)	[440]: 308, 179, 162	S-p-CoumaryIglutathione
15	31.8	269, sh303	468.1451 ª	[468]: 306, 288, 254	470.1588°	[470]: 308, 179, 162	S-Conifervlglutathione
		,	(468.1446)	[],,	(470.1592)	[],	~
16	32.2	279	498.1558 ª	[498]: 306, 288, 254	500.1694 °	[500]: 308, 179, 162	S-Sinapylglutathione
			(498.1552)		(500.1697)		
17	33.8	315	337.0568ª	[337]: 173, 155, 111	361.0529 °	[361]: 215, 197, 169, 147	<i>p</i> -Coumaroylisocitrate
18	33.9	269 sh303	(337.0303) 411 1238ª	[411] 281 249 128	(301.0330) 413.1371°	[413]: 251 163 131	N-I-y-Glutamyl-S-coniferyl-I-
10		207, 511303	(411.1231)	[111]. 201, 277, 120	(413.1377)	[10]. 201, 100, 101	cysteine
19	34.2	281	441.1339ª	[441]: 249, 153, 128	443.1478°	[443]: 251, 193, 161	N-L-γ-Glutamyl-S-sinapyl-L-
			(441.1337)	- '	(443.1483)	-	cysteine
20	37.8	sh294, 319	496.2458 ª	[496]: 346	498.2594 °	[498]: 481, 322, 234, 177	(di- <i>E</i> , <i>E</i> )- <i>N</i> , <i>N</i> '-
			(496.2453)		(498.2599)		DiteruloyIspermidine

 $\frac{(496.2433)}{t_{R}: \text{ retention time. } \lambda_{max}: UV \text{ maxima (sh: shoulder). n. d.: not detected. Calculated exact masses are given in parenthesis.}}^{(496.239)}$   $a_{m/z} \text{ of } [M-H]^{-}$   $b_{m-source decarboxylation} ([M-H-CO_{2}]^{-})$   $c_{m/z} \text{ of } [M+H]^{+}$   $d_{m-source deamination} [M+H-NH_{3}]^{+}$   $e_{s} \text{ Sodium adduct } [M+Na]^{+}$ 

#### Table 3.

				Pulp			Peel (pressed)			Peel (milled)		
Compounds	Т	S	$T^*S$	Т0	T7	T14	Т0	T7	T14	Т0	T7	T14
Amino acids and amines												
L-Tyrosine	0.000	0.000	0.119	$3.97\pm0.42^{\circ}$	$4.54\pm1.21^{\rm bc}$	$4.00\pm0.55^{\circ}$	$4.49\pm0.60^{\rm bc}$	$5.32\pm1.39^{ab}$	$5.46\pm0.64^{ab}$	$4.98\pm0.56^{\rm bc}$	$6.52\pm0.84^{\rm a}$	$6.54\pm1.15^{\rm a}$
Serotonin	0.083	0.115	0.451	$1.56\pm0.22^{\rm a}$	$1.90\pm0.12^{\rm a}$	$1.89\pm0.12^{\rm a}$	$1.75\pm0.35^{\rm a}$	$1.96\pm0.48^{\rm a}$	$2.09\pm0.16^{\rm a}$	$2.55\pm0.35^{\rm a}$	$2.94\pm0.31^{\rm a}$	$2.91\pm0.36^{\rm a}$
Total concentration	0.000	0.001	0.868	$5.53\pm0.50^{\text{b}}$	$6.44 \pm 1.16^{\text{b}}$	$7.29\pm4.16^{ab}$	$6.24\pm0.76^{\text{b}}$	$7.28\pm1.73^{ab}$	$7.54\pm0.71^{ab}$	$7.53\pm0.87^{ab}$	$9.46\pm0.83^{\rm a}$	$9.45\pm1.29^{\text{a}}$
Furanones												
4-Hydroxy-2,5-dimethyl- 3(2 <i>H</i> )-furanone hexoside	0.000	0.000	0.044	$5.86\pm0.45^{\rm c}$	$6.87\pm0.59^{\text{b}}$	$7.89\pm0.40^{\rm a}$	$4.56\pm0.54^{\rm d}$	$4.95\pm0.62^{\rm d}$	$6.33\pm0.28^{\text{bc}}$	$4.39\pm0.55^{\rm d}$	$4.91\pm0.54^{\rm d}$	$6.84\pm0.48^{\text{b}}$
4-Hydroxy-2,5-dimethyl- 3(2 <i>H</i> )-furanone	0.000	0.000	0.514	$1.26\pm0.14^{\rm c}$	$1.98\pm0.09^{\text{b}}$	$2.52\pm0.11^{\rm a}$	$0.86\pm0.13^{\rm d}$	$1.43\pm0.16^{\rm c}$	$2.09\pm0.18^{\text{b}}$	$0.81\pm0.35^{\text{d}}$	$1.49\pm0.17^{\rm c}$	$2.20\pm0.29^{\text{b}}$
4-Methoxy-2,5-dimethyl- 3(2 <i>H</i> )-furanone	0.000	0.000	0.168	$0.28\pm0.04^{\text{d}}$	$0.38\pm0.05^{\rm cd}$	$0.51\pm0.08^{\rm b}$	$0.33\pm0.05^{\text{d}}$	$0.46\pm0.12^{\rm bc}$	$0.64\pm0.03^{\rm a}$	$0.32\pm0.09^{\rm d}$	$0.46\pm0.09^{\rm bc}$	$0.68\pm0.12^{\rm a}$
4-Hydroxy-2,5-dimethyl- 3(2 <i>H</i> )-furanone malonyl hexoside (1)	0.000	0.885	0.000	$1.45\pm0.11^{\text{b}}$	$1.48\pm0.10^{\text{b}}$	$1.26\pm0.11^{\text{b}}$	$1.44\pm0.23^{\text{b}}$	$1.29\pm0.22^{\text{b}}$	$1.35\pm0.11^{\text{b}}$	$1.76\pm0.16^{\rm a}$	$1.85\pm0.20^{\rm a}$	$1.99\pm0.08^{\text{a}}$
4-Hydroxy-2,5-dimethyl- 3(2 <i>H</i> )-furanone malonyl hexoside (2)	0.000	0.038	0.114	$1.59\pm0.30^{\text{b}}$	$1.93\pm0.27^{\text{b}}$	$1.86\pm0.11^{\text{b}}$	$1.86\pm0.45^{\text{b}}$	$1.82\pm0.16^{\text{b}}$	$1.85\pm0.32^{\text{b}}$	$2.36\pm0.25^{\mathtt{a}}$	$2.44\pm0.30^{\rm a}$	$2.69\pm0.17^{\rm a}$
Total concentration	0.000	0.000	0.019	$10.44\pm0.82^{\text{ef}}$	$12.65\pm0.91^{bc}$	$14.03\pm0.58^{\text{ab}}$	$9.05\pm1.21^{\rm f}$	$9.95\pm1.09^{\text{ef}}$	$12.25\pm0.49^{\text{cd}}$	$9.63\pm1.31^{\rm f}$	$11.15\pm1.15^{\text{de}}$	$14.38\pm0.77^{\rm a}$
Phenolic compounds												
Syringoyl hexoside	0.000	0.349	0.001	$2.76\pm0.50^{\rm a}$	$2.53\pm0.44^{\rm ab}$	$2.38\pm0.23^{ab}$	$2.13\pm0.29^{\rm b}$	$2.14\pm0.37^{\rm b}$	$2.32\pm0.23^{ab}$	$2.15\pm0.11^{\text{b}}$	$2.55\pm0.29^{ab}$	$2.71\pm0.20^{\rm a}$
Sinapoyl hexoside	0.000	0.500	0.460	$5.68\pm0.51^{ab}$	$5.90\pm0.65^{\rm a}$	$5.97\pm0.52^{\rm a}$	$4.71\pm0.86^{\circ}$	$4.46\pm0.77^{\circ}$	$4.69\pm0.42^{\rm c}$	$4.53\pm0.71^{\circ}$	$5.07\pm0.63^{abc}$	$4.81\pm0.59^{bc}$
S-Coniferyl-L-cysteine	0.000	0.000	0.039	$0.70\pm0.11^{\text{d}}$	$0.85\pm0.14^{\rm cd}$	$0.89\pm0.07^{\rm cd}$	$0.82\pm0.11^{\text{cd}}$	$0.99\pm0.15^{\rm bc}$	$1.15\pm0.21^{ab}$	$0.83\pm0.48^{\rm cd}$	$1.22\pm0.23^{\rm a}$	$1.30\pm0.17^{\rm a}$
S-Sinapyl-L-cysteine	0.000	0.000	0.228	$8.12\pm1.56^{ab}$	$8.46\pm1.12^{\rm a}$	$8.73\pm0.46^{\rm a}$	$5.77\pm0.87^{\rm cd}$	$6.62\pm1.19^{bcd}$	$7.45\pm0.82^{\rm ab}$	$5.13\pm0.57^{\rm d}$	$6.74\pm1.42^{\text{bc}}$	$7.23 \pm 1.01^{abc}$
Caffeovlisocitrate	0.000	0.000	0.000	$1.22\pm0.28^{\rm f}$	$1.43\pm0.23^{\rm f}$	$2.32\pm0.17^{\text{de}}$	$1.63\pm0.47^{\rm ef}$	$1.94\pm0.37^{\text{def}}$	$2.55\pm0.45^{\rm d}$	$3.59\pm0.82^{\circ}$	$4.98\pm0.57^{\text{b}}$	$6.38\pm0.82^{\rm a}$
S-p-Coumarylglutathione	0.000	0.007	0.001	$0.24\pm0.05^{\rm a}$	$0.24\pm0.07^{\rm a}$	$0.18\pm0.02^{\text{bcd}}$	$0.15\pm0.03^{\text{cde}}$	$0.21\pm0.06^{\rm ab}$	$0.21\pm0.03^{abc}$	$0.13\pm0.01^{\text{de}}$	$0.14\pm0.03^{\text{de}}$	$0.10\pm0.01^{\circ}$
S-Conifervlglutathione	0.000	0.000	0.782	$0.79\pm0.13^{ab}$	$0.66\pm0.20^{\rm bc}$	$0.54\pm0.06^{\rm c}$	$1.02\pm0.16^{\rm a}$	$0.84\pm0.30^{ab}$	$0.78\pm0.04^{\text{abc}}$	$0.98\pm0.09^{\rm a}$	$0.88\pm0.18^{\text{ab}}$	$0.67\pm0.11^{\text{bc}}$
S-Sinapylglutathione	0.000	0.000	0.328	$4.94\pm0.60^{ab}$	$4.15 \pm 1.21^{\rm bc}$	$3.28\pm0.35^{\text{cd}}$	$5.44\pm0.76^{\rm a}$	$4.68 \pm 1.26^{\mathrm{ab}}$	$4.44\pm0.31^{abc}$	$4.90\pm0.67^{ab}$	$4.18 \pm 0.67^{bc}$	$2.95\pm0.69^{\rm d}$
<i>p</i> -Coumarovlisocitrate	0.095	0.000	0.162	$2.90\pm0.35^{\rm d}$	$3.28 \pm 0.22^{bcd}$	$4.69\pm0.92^{\mathrm{a}}$	$2.32\pm0.15^{\rm d}$	$3.14\pm0.32^{\text{cd}}$	$4.26\pm0.91^{\text{ab}}$	$3.00\pm0.57^{\rm cd}$	$3.05 \pm 1.11^{cd}$	$3.97\pm0.65^{abc}$
N-L-y-Glutamyl-S-coniferyl-	0.000	0.001	0.000	$1.33 \pm 0.16^{de}$	$1.27 \pm 0.14^{de}$	$1.20 \pm 0.12^{\circ}$	$1.74 \pm 0.19^{bc}$	$1.85 \pm 0.29^{abc}$	$2.01 \pm 0.25^{ab}$	$1.59 \pm 0.22^{cd}$	$2.05 \pm 0.30^{ab}$	$2.11 \pm 0.29^{a}$
L-cvsteine	01000	01001	0.000	1100 - 0110	1127 - 0111	1120 - 0112		1100 - 0127	2101 - 0120	1107 - 0122	2100 - 0100	2011 - 0129
$N-L-\gamma$ -Glutamyl-S-sinapyl-L-	0.000	0.017	0.013	$6.97\pm0.78^{abc}$	$6.79\pm0.61^{\rm abc}$	$6.09\pm0.24^{\rm bc}$	$7.07\pm0.82^{abc}$	$8.07\pm1.55^{\rm a}$	$8.26\pm0.73^{\text{a}}$	$5.90\pm0.66^{\circ}$	$7.59 \pm 1.32^{ab}$	$7.04 \pm 1.79^{abc}$
cysteine												
(di- <i>E</i> , <i>E</i> )- <i>N</i> , <i>N</i> '- Diferuloylspermidine	0.000	0.019	0.000	$0.23\pm0.03^{\text{e}}$	$0.18\pm0.09^{\rm e}$	$0.18\pm0.03^{\text{e}}$	$0.32\pm0.04^{\text{bcd}}$	$0.30\pm0.05^{\rm cd}$	$0.28\pm0.01^{\rm d}$	$0.34\pm0.03^{\rm bc}$	$0.38\pm0.02^{\rm a}$	$0.35\pm0.04^{ab}$
Total concentration	0.141	0.000	0.013	$35.87\pm2.90^{\rm ab}$	$35.75\pm2.47^{ab}$	$36.44 \pm 1.90^{ab}$	$33.10\pm2.90^{\text{b}}$	$35.23\pm3.24^{ab}$	$38.39\pm2.18^{\rm a}$	$33.07\pm3.40^{\rm b}$	$38.83\pm3.75^{\mathtt{a}}$	$39.63\pm3.90^{\mathtt{a}}$

The results are expressed as mg/100 mL. *T*, pineapple tissue variable, *S*, storage duration of the fruits; *T*\**S*, interaction of the variables; T0, fresh pineapples; T7, one week of storage at room temperature; T14, two weeks of storage. Different letters in one row indicate a significant (p < 0.05) difference of means determined by two-way ANOVA and Tukey's test.



Figure 1



Figure 2







# Authentication of pineapple (*Ananas comosus* [L.] Merr.) juice from pulp and peel by HPLC-DAD-(HR)-ESI-MS<sup>n</sup> analysis

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# Supplementary data

		Pilot plant (T7)	Pilot plant (T7)		Laboratory scale	e (T7)	
Parameter	Т	Pulp	Peel (pressed)	Peel (milled)	Pulp	Peel (pressed)	Peel (milled)
pН	0.009	$4.11\pm0.03^{ab}$	$4.08\pm0.04^{b}$	$4.08\pm0.03^{b}$	$4.12\pm0.04^{ab}$	$4.02\pm0.01^{b}$	$4.11\pm0.01^{ab}$
TSS (g/100 g)	0.000	$12.8\pm0.4^{\text{a}}$	$9.4\pm0.3^{\rm d}$	$10.3\pm0.3^{\texttt{c}}$	$12.6\pm0.4^{a}$	$10.7\pm0.5^{bc}$	$11.1\pm0.6^{b}$
TA (g/100 mL)	0.960	$0.48\pm0.04^{\rm a}$	$0.46\pm0.04^{\text{a}}$	$0.47\pm0.03^{\text{a}}$	$0.46\pm0.07^{a}$	$0.45\pm0.04^{\text{a}}$	$0.47\pm0.04^{a}$
TSS/TA	0.000	$26.9\pm2.6^{ab}$	$20.5\pm2.0^{\rm c}$	$22.3\pm1.7^{\text{bc}}$	$27.7\pm3.5^{a}$	$23.9\pm2.7^{abc}$	$23.9\pm3.0^{\text{abc}}$
Formol number (mL 0.1 M NaOH/100 mL)	0.000	$9.26\pm0.76^{\text{b}}$	$12.03\pm0.80^{\text{a}}$	$12.30\pm0.68^{a}$	$8.84 \pm 1.32^{\text{b}}$	$12.43\pm0.98^{\mathtt{a}}$	$11.46\pm0.87^{a}$
TPC (mg GAE/100 mL)	0.000	$84.3\pm4.6^{\text{cd}}$	$89.9\pm4.8^{bc}$	$100.6\pm3.3^{\text{a}}$	$80.2\pm4.7^{\text{d}}$	$86.2\pm4.7^{bc}$	$92.1\pm4.9^{b}$
Vitamin C (mg/100 mL)	0.000	$38.4\pm 3.6^{b}$	$32.6\pm2.3^{b}$	$35.4\pm1.6^{\text{b}}$	$57.6\pm9.1^{\text{a}}$	$59.4\pm7.7^{\rm a}$	$61.5\pm9.4^{a}$

Table S1.

*T*, pineapple tissue variable; T7, one week of storage Different letters in row indicate a significant (p < 0.05) difference of means determined by ANOVA and Tukey's test.

# Table S2.

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		Pilot plant (T7)			Laboratory scal			
Compounds	Т	Pulp	Peel (pressed)	Peel (milled)	Pulp	Peel (pressed)	Peel (milled)	
Amino acids and amines								
L-Tyrosine	0.000	$3.84\pm0.14^{\text{b}}$	$5.21\pm0.33^{ab}$	$4.96 \pm 0.45^{b}$	$4.54 \pm 1.21^{b}$	$5.32\pm1.39^{ab}$	$6.52\pm0.84^{\rm a}$	
Serotonin	0.000	$1.60\pm0.09^{\rm c}$	$2.09\pm0.20^{\mathrm{bc}}$	$2.60\pm0.02^{ab}$	$1.90\pm0.12^{\rm c}$	$1.96\pm0.48^{\rm c}$	$2.94\pm0.31^{\rm a}$	
Total concentration	0.000	$5.44\pm0.17^{\rm c}$	$7.30\pm0.38^{\rm b}$	$7.55\pm0.46^{\text{b}}$	$6.44 \pm 1.16^{\circ}$	$7.28\pm1.73^{\text{b}}$	$9.46\pm0.83^{\rm a}$	
Furanones								
4-Hydroxy-2,5-dimethyl-3(2H)-furanone hexoside	0.000	$7.30\pm0.10^{\rm a}$	$5.81\pm0.26^{\rm b}$	$5.95\pm0.17^{\rm b}$	$6.87\pm0.59^{\rm a}$	$4.95\pm0.62^{\circ}$	$4.91\pm0.54^{\rm c}$	
4-Hydroxy-2,5-dimethyl-3(2H)-furanone	0.000	$2.12\pm0.03^{\rm a}$	$1.73\pm0.10^{\rm b}$	$1.74\pm0.04^{\rm b}$	$1.98\pm0.09^{\rm a}$	$1.43\pm0.16^{\circ}$	$1.49\pm0.17^{\rm c}$	
4-Methoxy-2,5-dimethyl-3(2H)-furanone	0.000	$0.42\pm0.04^{\text{ab}}$	$0.55\pm0.03^{\text{a}}$	$0.56\pm0.03^{\rm a}$	$0.38\pm0.05^{\rm b}$	$0.46\pm0.12^{\rm ab}$	$0.46\pm0.09^{\text{ab}}$	
4-Hydroxy-2,5-dimethyl-3(2 <i>H</i> )-furanone malonyl hexoside (1)	0.000	$1.65\pm0.04^{\rm bc}$	$1.78\pm0.10^{\rm b}$	$2.31\pm0.13^{\rm a}$	$1.48\pm0.10^{\rm cd}$	$1.29\pm0.22^{\rm d}$	$1.85\pm0.20^{\text{b}}$	
4-Hydroxy-2,5-dimethyl-3(2 <i>H</i> )-furanone malonyl hexoside (2)	0.000	$2.09\pm0.36^{\rm cd}$	$2.57\pm0.15^{\rm b}$	$3.12\pm0.19^{\rm a}$	$1.93\pm0.27^{\rm d}$	$1.82\pm0.16^{\text{d}}$	$2.44\pm0.30^{\rm bc}$	
Total concentration	0.000	$13.58\pm0.38^{\rm a}$	$12.44\pm0.62^{ab}$	$13.69\pm0.55^{\rm a}$	$12.65\pm0.91^{\mathtt{a}}$	$9.95\pm1.09^{\rm c}$	$11.15\pm1.15^{\rm bc}$	
Phenolic compounds								
Svringovl hexoside	0.066	$2.54\pm0.07^{\rm a}$	$2.47\pm0.17^{\rm a}$	$2.39\pm0.06^{\rm a}$	$2.53\pm0.44^{\rm a}$	$2.14\pm0.37^{\rm a}$	$2.55\pm0.29^{\rm a}$	
Sinapovl hexoside	0.000	$6.16\pm0.12^{\rm a}$	$4.44\pm0.25^{\rm bc}$	$4.04\pm0.04^{\rm c}$	$5.90\pm0.65^{\rm a}$	$4.46\pm0.77^{\text{bc}}$	$5.07\pm0.63^{\rm b}$	
S-Conifervl-L-cvsteine	0.000	$0.89\pm0.07^{\rm b}$	$1.05\pm0.06^{\rm ab}$	$1.01\pm0.03^{ab}$	$0.85\pm0.14^{\text{b}}$	$0.99\pm0.15^{\text{b}}$	$1.22\pm0.23^{\rm a}$	
S-Sinapyl-L-cysteine	0.000	$8.53\pm0.42^{\rm a}$	$6.49\pm0.21^{\text{b}}$	$5.60\pm0.11^{\text{b}}$	$8.46 \pm 1.12^{\rm a}$	$6.62 \pm 1.19^{\text{b}}$	$6.74\pm1.42^{\text{b}}$	
Caffeoylisocitrate	0.000	$1.92\pm0.08^{\rm d}$	$3.96\pm0.36^{\circ}$	$7.67\pm0.57^{\rm a}$	$1.43\pm0.23^{\rm d}$	$1.94\pm0.37^{\text{d}}$	$4.98\pm0.57^{\text{b}}$	
S-p-Coumarylglutathione	0.000	$0.25\pm0.01^{\rm a}$	$0.12\pm0.02^{\text{b}}$	$0.13\pm0.00^{\text{b}}$	$0.24\pm0.07^{\rm a}$	$0.21\pm0.06^{\rm a}$	$0.14\pm0.03^{\text{b}}$	
S-Coniferylglutathione	0.025	$0.77\pm0.01^{ab}$	$1.01\pm0.06^{\rm a}$	$0.92\pm0.04^{ab}$	$0.66\pm0.20^{\text{b}}$	$0.84\pm0.30^{ab}$	$0.88\pm0.18^{ab}$	
S-Sinapylglutathione	0.587	$4.52\pm0.11^{\rm a}$	$4.84\pm0.35^{\rm a}$	$4.32\pm0.05^{\rm a}$	$4.15\pm1.21^{\rm a}$	$4.68 \pm 1.26^{\rm a}$	$4.18\pm0.67^{\rm a}$	
<i>p</i> -Coumaroylisocitrate	0.000	$4.32\pm0.24^{\rm a}$	$3.55\pm0.18^{ab}$	$2.31\pm0.52^{\circ}$	$3.28\pm0.22^{\rm b}$	$3.14\pm0.32^{\rm bc}$	$3.05\pm1.11^{\text{bc}}$	
N-L-γ-Glutamyl-S-coniferyl-L-cysteine	0.000	$1.40\pm0.02^{\rm b}$	$2.16\pm0.13^{\text{a}}$	$1.89\pm0.09^{\rm a}$	$1.27\pm0.14^{\rm b}$	$1.85\pm0.29^{\rm a}$	$2.05\pm0.30^{\rm a}$	
N-L-y-Glutamyl-S-sinapyl-L-cysteine	0.025	$6.61\pm0.15^{\rm a}$	$7.56\pm0.64^{\rm a}$	$6.49\pm0.08^{\rm a}$	$6.79\pm0.61^{\rm a}$	$8.07\pm1.55^{\rm a}$	$7.59 \pm 1.32^{\rm a}$	
(di- <i>E</i> , <i>E</i> )- <i>N</i> , <i>N</i> '-Diferuloylspermidine	0.000	$0.18\pm0.01^{\rm d}$	$0.23\pm0.04^{\circ}$	$0.22\pm0.00^{\rm cd}$	$0.18\pm0.09^{\rm d}$	$0.30\pm0.05^{\rm b}$	$0.38\pm0.02^{\rm a}$	
Total concentration	0.057	$38.10\pm1.10^{\rm a}$	$38.02\pm1.82^{\rm a}$	$36.99\pm0.34^{\rm a}$	$35.75\pm2.47^{\rm a}$	$35.23\pm3.24^{\rm a}$	$38.83 \pm 3.75^a$	
Ratio								
HDMF hexoside/HDMF malonvl hexosides	0.000	$1.97\pm0.23^{\rm a}$	$1.34\pm0.02^{\rm b}$	$1.10\pm0.04^{\rm c}$	$2.01\pm0.08^{\rm a}$	$1.59\pm0.11^{\rm b}$	$1.14\pm0.05^{\rm c}$	
$(N-L-\gamma-glutamvl-L-cvsteines + L-cvsteines)/GSHs$	0.327	$3.15\pm0.04^{\rm a}$	$2.89\pm0.10^{\text{a}}$	$2.79\pm0.02^{\rm a}$	$3.79 \pm 1.24^{\mathrm{a}}$	$3.38 \pm 1.24^{\mathrm{a}}$	$3.50\pm0.90^{\rm a}$	
Caffeovlisocitrate/ <i>n</i> -coumarovlisocitrate	0.000	$0.45 \pm 0.04^{\circ}$	$1.12 \pm 0.11^{\rm bc}$	$3.56 \pm 1.05^{a}$	$0.44 \pm 0.07^{\circ}$	$0.62 \pm 0.10^{\circ}$	$1.82 \pm 0.58^{b}$	
Sinapovl hexoside + $S$ -sinapvl- $L$ -cysteine/ $N$ - $L$ - $\gamma$ -glutamvl- $S$ -	0.000	$10.51 \pm 0.33^{a}$	$5.06 \pm 0.11^{b}$	$5.11 \pm 0.32^{b}$	$11.42 \pm 1.62^{a}$	$6.07 \pm 1.26^{\text{b}}$	$5.78 \pm 0.56^{\text{b}}$	
coniferyl-L-cysteine	0.000							

The results are expressed as mg/100 mL. *T*, pineapple tissue variable; T7, one week of storage at room temperature. Different letters in one row indicate a significant (p < 0.05) difference of means determined by ANOVA and Tukey's test.